

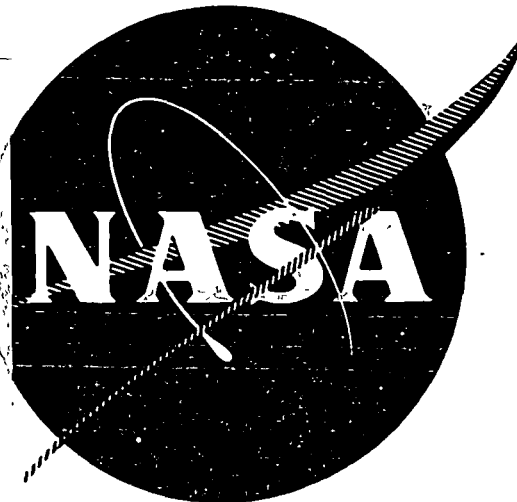
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DEVELOPMENT AND EVALUATION OF MAGNETIC AND ELECTRICAL MATERIALS CAPABLE OF OPERATING IN THE 800° TO 1600°F TEMPERATURE RANGE

First Quarterly Report

by

P. E. Kueser et al

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Report No. 65.17E

March 1965

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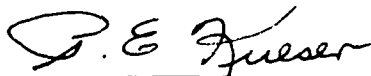
FIRST QUARTERLY REPORT
(NOVEMBER 12, 1964 - FEBRUARY 28, 1965)
NAS3-6465

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PREFACE

The work reported here was sponsored by the Space Power Systems Division of the NASA Lewis Research Center under contract NAS3-6465. Mr. R. A. Lindberg of NASA has provided the Project Management for the program. His review and suggestions as well as those of Mr. T. A. Moss, also of NASA, are gratefully acknowledged. The Westinghouse Aerospace Electrical Division (WAED) is responsible for the Technical Direction of the program. The Westinghouse Research and Development Center (WR&D) is conducting Tasks 2, 3 and 4 of Program I on Optimized Magnetic Materials for Application in the 1000 to 1200°F Range, the Investigation for Raising the Alpha to Gamma Transformation, and Creep Testing of Rotor Materials. Eitel-McCullough (EIMAC) is responsible for the Bore Seal Development, Task 1 of Program III. All other tasks are being conducted at the Westinghouse Aerospace Electrical Division (WAED).

In a program of this magnitude a large group of engineers and scientists are involved in its progress. An attempt to recognize those who are contributing directly, together with their area of endeavor, follows:

Program I - Magnetic Materials for High Temperature Operation

Task 1 - Optimized Precipitation Hardened Magnetic Materials for Application in the 1000 to 1200°F Range.

Dr. K. Detert (WR&D); J. W. Toth (WAED)

Task 2 - Investigation for Raising the Alpha to Gamma Transformation Temperature in Cobalt-Iron Alloys.

Dr. K. Detert (WR&D); J. W. Toth (WAED)

Task 3 - Dispersion-strengthened Magnetic Materials for Application in the 1200-1600°F Range.

Dr. R. J. Towner (WAED)

Task 4 - Creep Testing.

M. Spewock (WR&D); D. H. Lane (WAED)

Program II - High Temperature Capacitor Feasibility

R. E. Stapleton (WAED)

Program III - Bore Seal Development and Combined Material Investigation Under a Space Simulated Environment

Task 1 - Bore Seal Development

R. C. McRae, Dr. L. Reed (EIMAC); J. W. Toth (WAED)

Tasks 2, 3, 4 - Stator and Bore Seal, Transformer and Solenoid

**W. L. Grant, H. E. Keneipp, D. H. Lane, R. P. Shumate,
J. W. Toth (WAED)**

Dr. A. C. Beiler (WAED) and Dr. G. W. Weiner (WR&D) are acting as consultants on Program I.

ABSTRACT

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This is the first quarterly report on Contract NAS3-6465 for the Development and Evaluation of Magnetic and Electrical Materials Capable of Operating in the Temperature Range from 800°F to 1600°F. Advanced space electric power systems are the area of eventual application.

Program I is directed at developing high-temperature magnetic material with satisfactory strength for rotor use. The ternary alloy systems of cobalt-nickel-iron are under study with the cobalt and iron corners of the phase diagram. Thirty to forty per cent cobalt gives the greatest saturation at 600°C. The additions of beryllium to a composition of 15% nickel-25% cobalt-balance iron improves the stability of the alpha phase. Eighteen pre-alloyed atomized powders and 12 composite powders have been examined for use as dispersion-strengthened magnetic materials.

Program II will determine the feasibility of a high-temperature capacitor using high quality dielectric materials. Pyrolytic boron-nitride has been prepared in thicknesses of 0.0015 ± 0.0005 inches; Lucalox alumina rods to 0.006 inches thick.

Program III incorporates combinations of materials into a stator with bore seal, a transformer, and a solenoid for investigations of compatibility under electrical stress and space-simulated conditions of elevated temperature. A system for loading and sealing alkali-metals in vacuum is in the progress of assembly. The stator, transformer and solenoid designs for use in space-simulated testing have been completed.

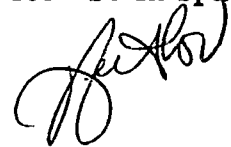


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SECTION I

INTRODUCTION

This is the first quarterly report on Contract NAS3-6465 for the Development and Evaluation of Magnetic and Electrical Materials Capable of Operating in the Temperature Range from 800°F to 1600°F. The period of performance is from November 12, 1964 through February 28, 1965. The program consists of three Programs with their related tasks as follows:

Program I - Magnetic Materials for High-Temperature Operation

Task 1 - Optimized Precipitation Hardened Magnetic Materials for Application in the 1000 to 1200°F Range

Task 2 - Investigation for Raising the Alpha to Gamma Transformation Temperature in Cobalt-Iron Alloys

Task 3 - Dispersion-Strengthened Magnetic Materials for Application in the 1200 to 1600°F Range

Task 4 - Creep Testing

Program II - High-Temperature Capacitor Feasibility

Program III - Bore Seal Development and Combined Material Investigations Under a Space Simulated Environment

Task 1 - Bore Seal Development

Task 2 - Stator and Bore Seal

Task 3 - Transformer

Task 4 - Solenoid

In Program I, limitations in magnetic material performance at elevated temperature have been recognized from Contract NAS3-4162 and the development of materials incorporating improved magnetic and mechanical properties is being pursued. In most cases, high-strength compromises the magnetic properties; therefore, a balance between these two variables is sought.

Program II is directed at determining the feasibility of applying high-quality dielectric materials and their processes to a high-temperature (1100°F) capacitor which is lightweight and suitable for static power conditioning apparatus used in space applications.

Program III incorporates combinations of materials previously evaluated under Contract NAS3-4162 into a stator with a bore seal, a transformer, and a solenoid; all of which will be evaluated under space-simulated conditions at elevated temperature.

The three Programs will be reported consecutively in Sections II, III and IV. Each of these Sections is further subdivided into a summary of the technical progress, a discussion, and the program for the next quarter.

SECTION II

PROGRAM I - MAGNETIC MATERIALS FOR HIGH-TEMPERATURE OPERATION

Program I is directed at the improvement and further understanding of magnetic materials suitable for application in the rotor of an alternator or motor in advanced space electric power systems.

Task 1 is concerned with precipitation-hardened magnetic materials in the 1000 to 1200°F range. These materials are of the iron-cobalt-nickel ternary system. The two specific areas of interest are the iron and cobalt corners of the ternary system.

Task 2 is a small research investigation for determining the feasibility of raising the alpha to gamma transformation temperature in the iron-cobalt system; thereby increasing the useful magnetic temperature of this system.

Task 3 is directed at applying dispersion-strengthening mechanisms to magnetic materials to achieve useful and invariant mechanical and magnetic properties in the 1200 to 1600°F range. Because both variables are influenced differently by particle size and spacing, a compromise is sought thereby tailoring the materials to the need of dynamic electric machines.

Task 4 is a creep program on Nivco alloy (approximately 72% cobalt, 23% nickel and certain other elements) which will generate 5000 hour design data in a vacuum environment (1×10^{-6} torr). This material represents a presently available magnetic material with the highest useful application temperature.

A. TASK 1 - OPTIMIZED PRECIPITATION-HARDENED MATERIALS FOR HIGH-TEMPERATURE APPLICATION.

1. Summary of Technical Progress

- a.) Twenty ferritic and twenty cobalt base alloys were selected and approved for evaluation as high-strength magnetic alloys.
- b.) All of the ferritic alloys and most of the cobalt-base alloys have been melted.
- c.) Transformation temperatures of ten ferritic alloys were determined during fast and slow cooling. Several of these alloys were subjected to isochronal aging between 842°F (450°C) and 1202°F (650°C). Hardness and coercivity were tested at room temperature.
- d.) Magnetic saturation measurements at room temperature were made on all of the ferritic alloys.
- e.) Three of the cobalt base alloys have been subjected to isochronal aging between 932°F (500°C) and 1382°F (750°C). Hardness and coercivity were determined at room temperature.

2. Discussion

A suitable alloy composition must be found which gives good strength at elevated temperatures combined with good magnetic properties. The target ultimate tensile strength for the alloy at 1100°F is 125,000 psi or better. The target stress to produce 0.40% creep strain in 1000 hours at 1100°F is 76,000 psi or greater. The 10,000 hour stress target at 1100°F is 80 to 90% of that at 1000 hours. The target magnetic saturation for the developmental alloy is 13,000 gauss or better at 1100°F and a coercive force less than 25 oersteds. In order to screen a variety of compositions, the Vickers hardness, coercivity, magnetic saturation, and temperature stability of the magnetic phase were selected as suitable parameters to evaluate the different alloy compositions.

Two general areas of basic alloy composition are regarded as most hopeful. First, body-centered ferritic alloys of iron and cobalt using additional components for obtaining high strengths by a precipitation hardening mechanism (Group A). Second, cobalt base alloys with alloying additions to obtain precipitation strengthening (Group B).

In the ferritic alloys, the maraging treatment gives high strength compared to highly alloyed carbon steels. The maraging treatment requires a certain amount of nickel present (15% - 25%) in the alloy. One goal of the screening program will be to find a suitable composition which can make use of a maraging treatment applied to a Fe-Co alloy with sufficient temperature stability of the alpha phase.

In the cobalt base alloys, another strengthening mechanism will be tried. This mechanism gives rise to considerable strengths in nickel-cobalt alloys by the formation of the face centered, ordered cubic phase (Ni_3Al) designated gamma prime. The formation of gamma prime, however, requires the presence of a certain amount of nickel in the alloy.

The Group A alloys 1-A-1 to 1-A-10 (Table II-1) were selected based on a review of literature and on results available from previous studies by Westinghouse. The composition, 15 nickel and 25 cobalt, will allow a martensitic transformation which is a prerequisite for maraging treatment. Furthermore, it has been established that stability of the alpha phase can be expected up to 1202°F (650°C) for 500 to 1000 hours. The influence of titanium, molybdenum and tantalum has been recognized as suitable for maraging treatment. W, Nb, V, Cr, Al, Be, Sb, Sn, Si and Mn will be evaluated for their potential to exert a strengthening influence on a maraging treatment. The effect of alloying on the temperature of transformation will be checked to determine whether the martensitic transformation can be applied and whether the martensitic phase is sufficiently stable at temperature.

Tests on the Group A alloys 1-A-11 to 1-A-20 (Table II-1) will define a suitable composition which is lower in nickel (12%). It is expected that saturation will be increased compared to the 15 nickel containing alloys. However, there is some doubt whether a martensitic transformation will occur during cooling.

In the cobalt-base alloys, the series 1-B-1 to 1-B-12 (Table II-2) will produce some information as to the region in which high saturation is attained and what portion will be sacrificed when nickel is added to obtain a suitable strength by gamma prime formation. The alloys 1-B-13 to 1-B-15 were selected to define the role of nickel in achieving strength by the formation of gamma prime.

TABLE II-1. Group A - Precipitation-Hardening Alloys

Number	Nominal Alloy Composition (Weight Percent)
1-A-1	55 Fe - 15 Ni - 25 Co - 5 W
1-A-2	58 Fe - 15 Ni - 25 Co - 2 Nb
1-A-3	58 Fe - 15 Ni - 25 Co - 2 V
1-A-4	58 Fe - 15 Ni - 25 Co - 2 Cr
1-A-5	59.5 Fe - 15 Ni - 25 Co - 0.5 Al
1-A-6	58 Fe - 15 Ni - 25 Co - 2 Be
1-A-7	58 Fe - 15 Ni - 25 Co - 2 Sb
1-A-8	58 Fe - 15 Ni - 25 Co - 2 Sn
1-A-9	58 Fe - 15 Ni - 25 Co - 2 Si
1-A-10	58 Fe - 15 Ni - 25 Co - 2 Mn
1-A-11	88 Fe - 12 Ni
1-A-12	83 Fe - 12 Ni - 5 Co
1-A-13	78 Fe - 12 Ni - 10 Co
1-A-14	73 Fe - 12 Ni - 15 Co
1-A-15	68 Fe - 12 Ni - 20 Co
1-A-16	63 Fe - 12 Ni - 25 Co
1-A-17	58 Fe - 12 Ni - 30 Co
1-A-18	53 Fe - 12 Ni - 35 Co
1-A-19	48 Fe - 12 Ni - 40 Co
1-A-20	43 Fe - 12 Ni - 45 Co

TABLE II-2. Group B - Precipitation-Hardening Alloys

Number	Nominal Alloy Composition (Weight Percent)
1-B-1	95 Co - 5 Fe
1-B-2	90 Co - 5 Fe - 5 Ni
1-B-3	85 Co - 5 Fe - 10 Ni
1-B-4	80 Co - 5 Fe - 15 Ni
1-B-5	75 Co - 5 Fe - 20 Ni
1-B-6	70 Co - 5 Fe - 25 Ni
1-B-7	65 Co - 5 Fe - 30 Ni
1-B-8	85 Co - 10 Fe - 5 Ni
1-B-9	80 Co - 15 Fe - 5 Ni
1-B-10	75 Co - 20 Fe - 5 Ni
1-B-11	70 Co - 25 Fe - 5 Ni
1-B-12	65 Co - 30 Fe - 5 Ni
1-B-13	86 Co - 5 Fe - 5 Ni - 3 Ti - 1 Al
1-B-14	81 Co - 5 Fe - 10 Ni - 3 Ti - 1 Al
1-B-15	76 Co - 5 Fe - 15 Ni - 3 Ti - 1 Al
1-B-16	81 Co - 10 Fe - 5 Ni - 3 Ti - 1 Al
1-B-17	76 Co - 15 Fe - 5 Ni - 3 Ti - 1 Al
1-B-18	71 Co - 20 Fe - 5 Ni - 3 Ti - 1 Al
1-B-19	66 Co - 25 Fe - 5 Ni - 3 Ti - 1 Al
1-B-20	61 Co - 30 Fe - 5 Ni - 3 Ti - 1 Al

a. EXPERIMENTAL PROCEDURE

All of the alloys were prepared by the levitation melting technique. In general, about 20 grams of the pure constituents, present as tiny chips and in powder form, were compacted at a pressure of 220,000 psi applied for one minute. High purity materials were used. Iron and cobalt of electrolytic grade, better than 99.9 percent pure, and nickel in the powder form, with a purity of 99.98 percent, were used. The levitation melting was performed in a funnel-shaped coil with five windings. The frequency used was in the range of 550 kc; the input of energy was in the range of 3 to 5 KW. The melting was done in a chamber which was evacuated to 1×10^{-2} torr, then back-filled using argon (10 ppm O₂ maximum, 5 ppm H₂, 40 ppm N₂) with a dewpoint of -85°F under reduced pressure on the order of 75 mm Hg. The material was held for about eight seconds in the molten state and then dropped into a copper mold by cutting the power. The mold was slightly tapered to provide a bar shaped ingot with a 9/32 inch diameter on one end and 5/16 inch diameter on the upper end. The length was 1-7/8 inches. Dilatometer test specimens were machined from the ingot. The specimens were one inch long, 1/4 inch in diameter and contained a hole for inserting a thermocouple. The samples for saturation measurements were small cylinders with a 1/10 inch diameter and 1/10 inch in length. The dilatometer tests were made under argon by heating up to 1832°F (1000°C), then cooling while recording the change in length at temperature. The heating rate applied was of the order of 90°F/min (50°C/min) and 1.8 to 3.6°F/min (1 to 2°C/min.). The cooling rate was 90°F/min (50°C/min) and 9°F/min (5°C/min). After the samples had been tested in the dilatometer, the samples were cold rolled to flat strips of 95 mil thickness. The samples were homogenized and austenitized in an argon flushed tube furnace at 1832°F (1000°C) for 30 minutes. Then the samples underwent isochronal anneal as follows. The samples were held for one hour at temperature with intermittent increases of 90°F (50°C) between 842°F (450°C) and 1382°F (750°C). Aging at these temperatures was done in a salt bath. * Hardness and coercivity were measured on these

*Thermosalt No. 311 was used for aging treatments in the range of 842°F (450°C) to 1022°F (550°C).

Thermosalt No. 914 was used for aging treatments in the range of 1112°F (600°C) to 1292°F (700°C).

Thermosalt No. 1018 was used for aging treatments at 1382°F (750°C). All designations are those of the Carmac Chemical Co.

samples. The Vickers hardness was measured at room temperature with a load of 50 kilograms. Coercivity was measured in a Forster probe. The magnetizing field was 1300 oersteds.

Saturation was measured in a magnetic field with a gradient of 980 oersteds per centimeter. The mean value of the applied field was on the order of 10,000 oersteds. For measuring the saturation at high temperature, a small electric resistance furnace was placed into the gap of the magnet which was double wound to avoid any additional field by the electric current passing through the windings. This technique reduced the residual field below 100 millioersteds.

b. RESULTS

Testing of the alloys 1-A-1 to 1-A-20 was undertaken. The alloys 1-A-1, 1-A-5, 1-A-6, 1-A-7, 1-A-8, 1-A-9 broke during cold rolling. Further investigations will be made to determine modifications needed to permit rolling. Dilatometer measurements were made on alloys 1-A-1 to 1-A-20. The results of the dilatometer tests are listed in Tables II-3 and II-4. The room temperature saturation measurements of the samples 1-A-1 to 1-A-10 are listed in Table II-5. Saturation curves at room temperature and at 1112°F (600°C) for samples 1-A-11 to 1-A-20 are shown in Figure II-1. The approximate saturation in gauss may be obtained by multiplying the cgs magnetic moment by one hundred. In Table II-6, the maximum hardness which could be obtained by the isochronal anneal is listed for several alloys at the particular aging temperature. Coercivity is also listed. Isochronal annealing has shown that only the 1-A-2 alloy showed considerable strengths after maraging treatment. Figure II-2 shows coercivity and hardness as a function of aging temperature for this alloy.

Testing of all the additions to the alloy composition of 15 nickel, 25 cobalt balance iron (Alloys 1-A-1 to 1-A-10) had shown that beryllium was the only element which would increase the stability of the alpha phase considerably. The addition of 5% tungsten (1-A-1), 2% chromium (1-A-4), 2% manganese (1-A-10), 2% tin (1-A-8) and 2% vanadium (1-A-3) decreased the temperature stability sufficiently that one cannot expect adequate stability at 1112°F (600°C). All the additions except beryllium decreased the transformation temperature into alpha phase during the cooling (See Table II-4). The influence of silicon was very pronounced.

**TABLE II-3. Transformation Temperature in Degrees Fahrenheit and Centigrade of
Fe-15Ni-25Co Alloys With Modifiers**

Alloy Designation	Nominal Alloy Composition (Weight Percent)	Heating $\alpha \xrightarrow{\gamma}$			Cooling $\gamma \xrightarrow{\alpha}$			
		90 °F/min	(50 °C/min)	1.8 °F/min	(1 °C/min)	90 °F/min	(50 °C/min)	9 °F/min (5 °C/min)
1-A-1	55 Fe - 15 Ni - 25 Co - 5 W	1425 - 1551	774 - 844	1087 - 1450	586 - 788	752 - 572	400 - 300	721 - 554 383 - 290
1-A-2	58 Fe - 15 Ni - 25 Co - 2 Nb	1472 - 1634	800 - 890	1193 - 1542	645 - 839	916 - 662	491 - 350	891 - 680 477 - 360
1-A-3	58 Fe - 15 Ni - 25 Co - 2 V	1472 - 1603	800 - 873	1112 - 1530	600 - 832	878 - 678	470 - 359	948 - 783 509 - 417
1-A-4	58 Fe - 15 Ni - 25 Co - 2 Cr	1450 - 1587	788 - 864	1065 - 1472	574 - 800	828 - 628	442 - 331	863 - 680 462 - 360
1-A-5	59.5 Fe - 15 Ni - 25 Co - 0.5 Al	1521 - 1652	827 - 900	1164 - 1515	629 - 824	986 - 781	530 - 416	1033 - 839 556 - 448
1-A-6	58 Fe - 15 Ni - 25 Co - 2 Be	1548 - 1666	842 - 908	1447 - 1630	786 - 888	1326 - 1207	719 - 653	1517 - 1265 825 - 685
1-A-7	58 Fe - 15 Ni - 25 Co - 2 Sb	1515 - 1634	824 - 890	1175 - 1490	635 - 810	846 - 631	452 - 333	903 - 729 484 - 387
1-A-8	58 Fe - 15 Ni - 25 Co - 2 Sn	1456 - 1594	791 - 868	1112 - 1515	600 - 824	932 - 709	500 - 376	966 - 752 519 - 400
1-A-9	58 Fe - 15 Ni - 25 Co - 2 Si	1434 - 1612	779 - 878	1195 - 1503	646 - 817	815 - 572	435 - 300	855 - 669 457 - 354
1-A-10	58 Fe - 15 Ni - 25 Co - 2 Mn	1472 - 1618	800 - 881	1058 - 1450	570 - 788	786 - 545	419 - 285	842 - 610 450 - 321

TABLE II-4. Transformation Temperature in Degrees Fahrenheit and Centigrade of Iron - 12 Nickel Alloys With Varying Cobalt Content

Alloy Designation	Nominal Alloy Composition (Weight Percent)	Heating $\alpha \longrightarrow \gamma$			Cooling $\gamma \longrightarrow \alpha$			
		90°F/min	(50°C/min)	1.8°F/min (1°C/min)	90°F/min	(50°C/min)	9°F/min (5°C/min)	
1-A-11	88 Fe - 12 Ni	1240 - 1328	671 - 720	1143 - 1261	617 - 683	347 - 443	984 - 721	529 - 383
1-A-12	83 Fe - 12 Ni - 5 Co	1323 - 1386	717 - 752	1207 - 1384	653 - 751	511 - 415	943 - 779	506 - 415
1-A-13	78 Fe - 12 Ni - 10 Co	1386 - 1441	752 - 783	1207 - 1463	653 - 795	529 - 435	997 - 826	536 - 441
1-A-14	73 Fe - 12 Ni - 15 Co	1459 - 1530	793 - 832	1249 - 1521	676 - 827	587 - 490	1072 - 896	578 - 480
1-A-15	68 Fe - 12 Ni - 20 Co	1494 - 1573	812 - 856	1175 - 1560	635 - 849	625 - 522	1143 - 1006	617 - 541
1-A-16	63 Fe - 12 Ni - 25 Co	1551 - 1639	844 - 893	1193 - 1585	645 - 863	634 - 529	1170 - 932	632 - 500
1-A-17	58 Fe - 12 Ni - 30 Co	1515 - 1634	824 - 890	1249 - 1582	676 - 861	636 - 524	1182 - 945	639 - 508
1-A-18	53 Fe - 12 Ni - 35 Co	1481 - 1625	805 - 885	1227 - 1594	664 - 868	610 - 511	1155 - 1009	624 - 543
1-A-19	48 Fe - 12 Ni - 40 Co	1472 - 1625	800 - 885	1157 - 1618	625 - 881	597 - 494	1173 - 993	634 - 534
1-A-20	43 Fe - 12 Ni - 45 Co	1463 - 1643	795 - 895	1130 - 1594	610 - 868	585 - 485	1107 - 968	597 - 520

TABLE II-5. Saturation Magnetic Moment (emu/g) at Room Temperature
Annealed at 1832°F (1000°C) for 30 Minutes, No Aging

Alloy Designation	Nominal Alloy Composition (Weight Percent)	Saturation (emu/g)	Change in Saturation vs. 60 Fe-15 Ni-25 Co (emu/g)
1-A-1	55 Fe - 15 Ni - 25 Co - 5 W	197.6	-17.6
1-A-2	58 Fe - 15 Ni - 25 Co - 2 Nb	208.4	-6.8
1-A-3	58 Fe - 15 Ni - 25 Co - 2 V	206.4	-8.8
1-A-4	58 Fe - 15 Ni - 25 Co - 2 Cr	204.8	-10.4
1-A-5	59.5 Fe - 15 Ni - 25 Co - 0.5 Al	212.0	-3.2
1-A-6	58 Fe - 15 Ni - 25 Co - 2 Be	194.8	-20.4
1-A-7	58 Fe - 15 Ni - 25 Co - 2 Sb	210.8	-4.4
1-A-8	58 Fe - 15 Ni - 25 Co - 2 Sn	210.8	-4.4
1-A-9	58 Fe - 15 Ni - 25 Co - 2 Si	202.0	-13.2
1-A-10	58 Fe - 15 Ni - 25 Co - 2 Mn	214.0	-1.2
	60 Fe - 15 Ni - 25 Co	215.2	

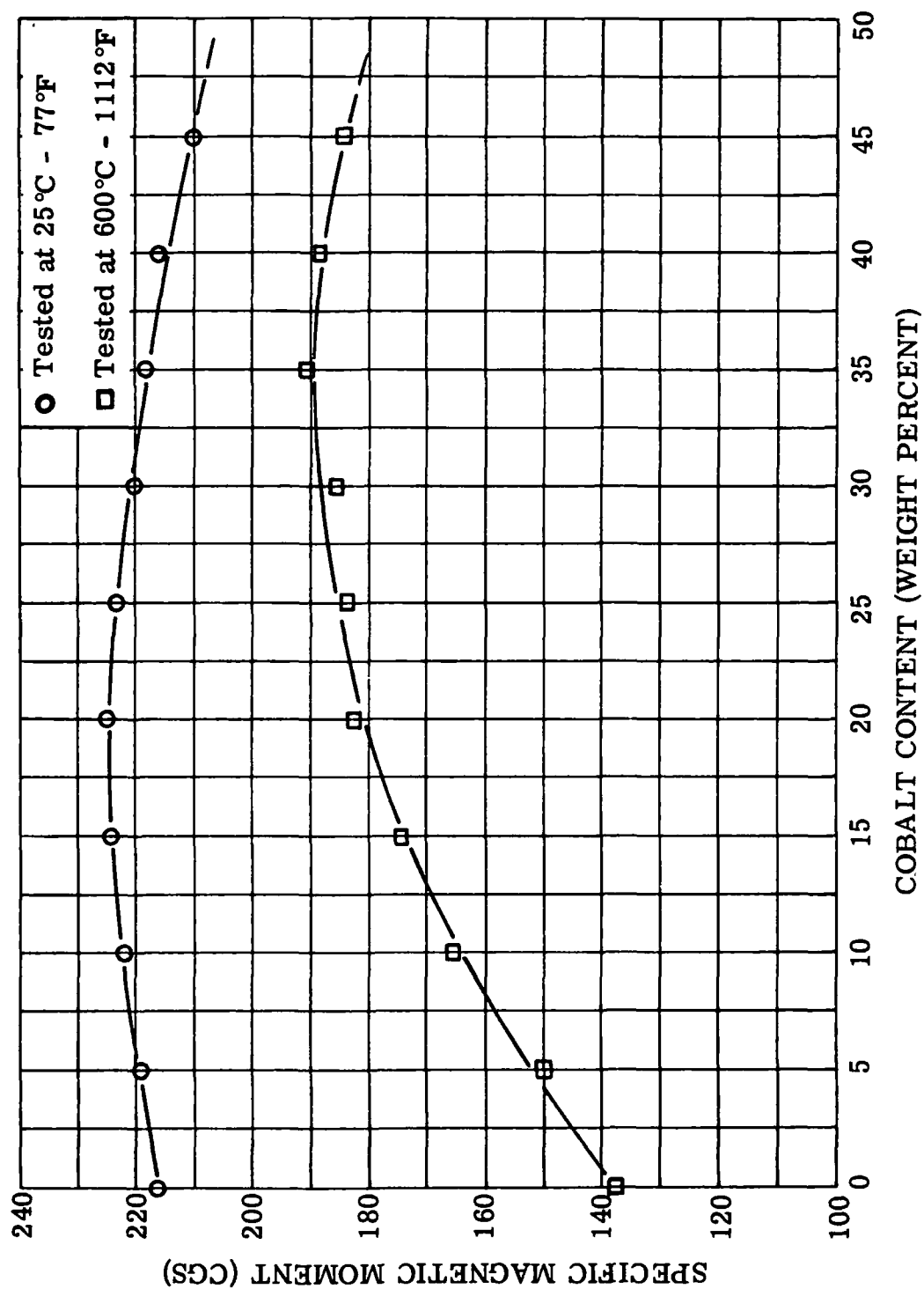


FIGURE II-1. Influence of Cobalt Content on Saturation in a Ternary Alloy With 12 Percent Nickel in Iron

Figure II-1. Magnetic Moment vs. Cobalt Content for 12% Ni in Iron

TABLE II-6. Maximum Hardness Obtained by Isochronal Aging

Alloy Designation	Nominal Alloy Composition (Weight Percent)	Aging Temperature at Which Maximum Room Temperature Hardness was Obtained (°F)		Aging Temperature (°C)	Total Aging Time (Hours)	Maximum Room Temperature Hardness (VHN)	Room Temperature Coercivity at Maximum Hardness (Oersteds)
1-A-2	58Fe-15Ni-25Co-2Nb	932		500	2 ⁽¹⁾	558	27.4
1-A-3	58Fe-15Ni-25Co-2V	932		500	2 ⁽¹⁾	390	18.3
1-A-4	58Fe-15Ni-25Co-2Cr	1022		550	3 ⁽¹⁾	391	20.1
1-B-13	86Co-5Fe-5Ni-3Ti-1Al	1112		600	4 ⁽¹⁾	218	1.62
1-B-14	81Co-5Fe-10Ni-3Ti-1Al	1292		700	6 ⁽¹⁾	239	2.45
1-B-15	76Co-5Fe-15Ni-3Ti-1Al	1292		700	6 ⁽¹⁾	273	3.36
(1) - Total aging time consists of one hour at 450°C plus one hour at each 50°C increment. (e.g. total time for alloy 1-A-4 equals 1 hour at 450°C plus 1 hour at 500°C plus 1 hour at 550°C). See text page 8.							

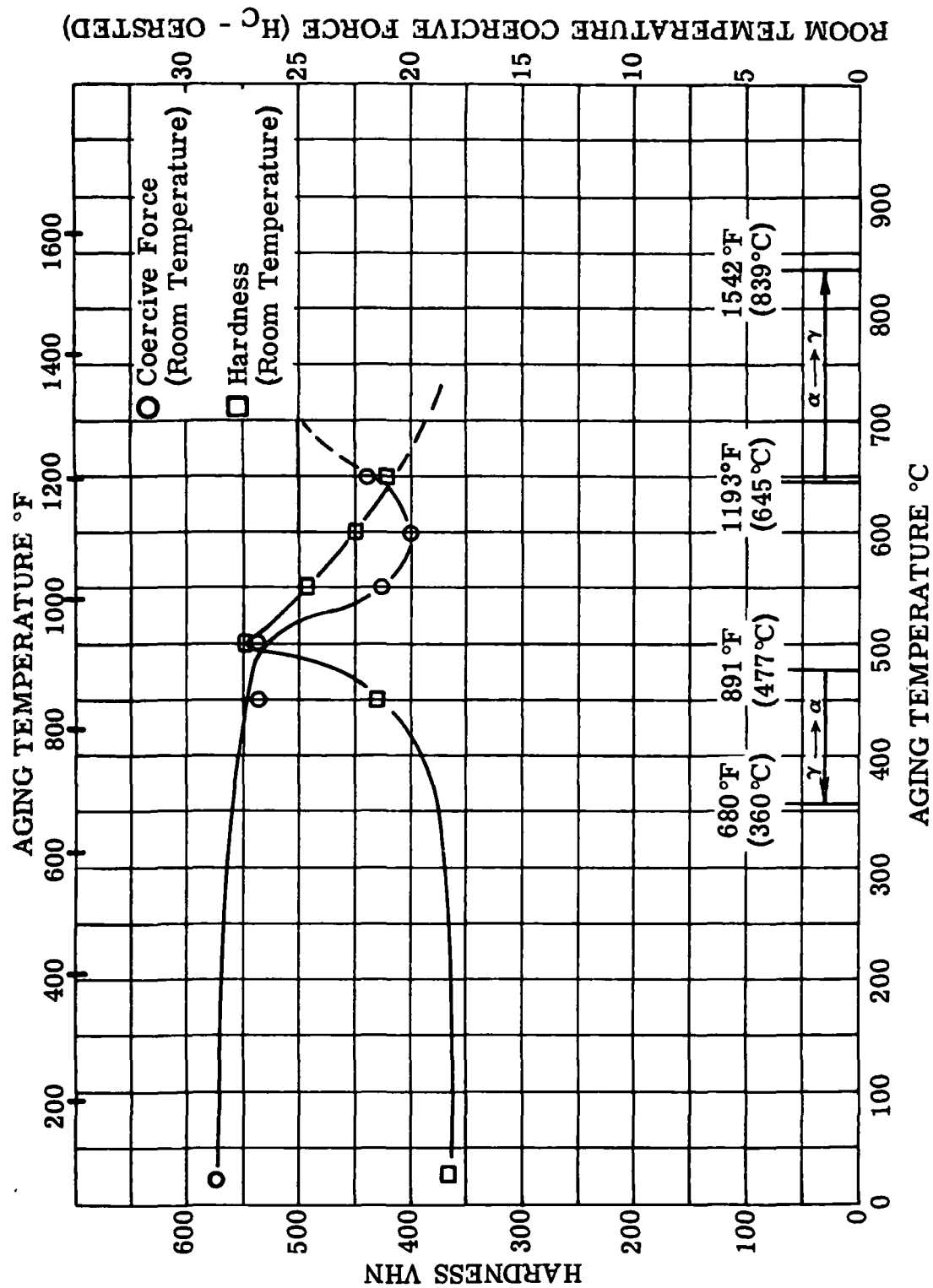


Figure II-2. Room Temperature Hardness and Coercive Force of Alloy 1-A-2

FIGURE II-2. Hardness and Coercive Force of Alloy 1-A-2 (58Fe-15Ni-25Co-2Nb) at Room Temperature After Aging One Hour at Temperature Between 450°C and Maximum Temperature in 50°C Increments

The saturation measurements showed that all additions decreased the saturation at room temperature compared to a ternary alloy of 15 nickel and 25 cobalt, balance iron (Table II-5).

From the alloys tested for response to maraging treatment (Table II-6), it is seen that the effect of niobium addition (Alloy 1-A-2) is similar to the addition of titanium in maraging steels if atomic percent additions are compared. Vanadium and chromium apparently have no effect on strengthening. The other series A alloying additions have not been tested.

Testing the series of alloys containing 12 nickel (Alloys 1-A-11 to 1-A-20) showed the influence of cobalt on saturation (Figure II-1). Thirty to forty percent cobalt give the highest saturation at 1112°F (600°C). Alloys with twenty to thirty percent cobalt show slightly reduced values of saturation. The dilatometer test showed that, with the exception of the binary alloy (88 Fe - 12 Ni) and the alloy with 45 Co, all alloys appear to have sufficient stability of the alpha phase (Table II-4).

The testing of the cobalt base alloys is in its initial stage. One could generally see that the strengths will not reach comparable levels to the alloys where maraging can be applied. (Comparison of hardness of alloys are shown in Table II-6.) By the same reason, the obtained values of the coercivity are considerably lower. The three alloys (1-B-13 through 1-B-15) tested show that increase in the nickel content will lead to increased strengthening if the same amount of aluminum and titanium are added.

3. Program for the Next Quarter

The tests of the alloys already melted will be completed. The phases formed in the precipitation hardening process must be identified and their stability must be ascertained during isochronal aging at a suitable temperature. After completing the results, and pending approval, a new series of alloys will be started. The tests of the new alloys shall fulfill two different goals. One will be to determine the suitable composition range of the primary components and the second is to find the best combination of suitable alloying elements to obtain the desired magnetic and mechanical properties.

B. TASK 2 - INVESTIGATION FOR RAISING THE ALPHA TO GAMMA TRANSFORMATION TEMPERATURE IN COBALT-IRON ALLOYS.

1. Summary of Technical Progress

a.) Eighteen cobalt-iron alloys were selected for evaluation (Table II-7).

b.) Eight alloys have been melted in the form of ingots by the levitation melting technique.

2. Discussion

a. BACKGROUND

Evaluating the results of previous research work, including that performed at the Westinghouse Research Laboratories, and a survey of the binary cobalt-iron phase diagram led to the selection of the basic alloy compositions shown in Table II-7.

The highest transformation temperature of 1805°F (985°C) will be obtained with an alloy containing 45 Co. The highest value of saturation (B_s) at temperatures above 1292°F (700°C) are measured in Fe-50 Co ($B_s = 10,000$ gauss at 1742°F (950°C)). So a basic composition Fe:Co at 1:1 appears most suitable for initial efforts to raise the transformation temperature by addition of alloying elements.

Addition of one weight percent of the elements known to raise the alpha to gamma transformation temperature in pure iron will be checked for their influence on transformation temperature and saturation of the 1:1 cobalt-iron alloy. The first 12 alloys were selected based on this argument. The alloys with columbium and tantalum contain only 0.5 weight percent of the alloying elements because it is questionable whether the solubility limit will extend above this value in the temperature region to be tested.

Chromium and zinc close the gamma region of iron, although the transformation temperature of iron will initially be lowered by the addition of either element. The addition of gold to iron decreases the transformation temperature but slightly.

TABLE II-7. Alloys Selected for Alpha-Gamma Transformation Study

Number	Nominal Alloy Composition (Weight Percent)
2-0	50.0 Fe - 50.0 Co
2-1	49.5 Fe - 49.5 Co - 1 Al
2-2	49.5 Fe - 49.5 Co - 1 As
2-3	49.5 Fe - 49.5 Co - 1 Be
2-4	49.5 Fe - 49.5 Co - 1 Ge
2-5	49.5 Fe - 49.5 Co - 1 Mo
2-6	49.5 Fe - 49.5 Co - 1 P
2-7	49.5 Fe - 49.5 Co - 1 Sb
2-8	49.5 Fe - 49.5 Co - 1 Si
2-9	49.5 Fe - 49.5 Co - 1 Sn
2-10	49.5 Fe - 49.5 Co - 1 Ti
2-11	49.5 Fe - 49.5 Co - 1 V
2-12	49.5 Fe - 49.5 Co - 1 W
2-13	49.75 Fe - 49.75 Co - 0.5 Nb
2-14	49.75 Fe - 49.75 Co - 0.5 Ta
2-15	49.5 Fe - 49.5 Co - 1 Cr
2-16	49.5 Fe - 49.5 Co - 1 Zn
2-17	49.5 Fe - 49.5 Co - 1 Au
2-18	49.5 Fe - 49.5 Co - 1 Mn

In the case that all alloys tested do not indicate a potential for extending the temperature range where an alloy with magnetization saturation superior to pure cobalt can be used, a second series may be suggested.

It is anticipated that a second series of alloys would include additions of 0.5 weight percent rare earth elements. It is not known whether alloying is possible and what effect it might have on transformation temperature and saturation. It is known that the number of Bohr magnetons in some of these alloys are very high. Gadolinium, terbium, dysprosium, holmium, erbium, and thulium have effective Bohr magneton numbers of eight or higher. A unique kind of interaction of the electrons might be expected if these elements prove soluble in the Fe-Co matrix.

b. **EXPERIMENTAL PROCEDURE:** (See Section II. A. 2. a.)

c. **RESULTS**

The alloys numbered 2-0, 2-1, 2-3, 2-5, 2-7, 2-17, and 2-18 were melted as 15 gram ingots by the levitation melting technique described in Section II. A.

The melting of alloys 2-2, 2-6, and 2-16 has been postponed until master melts are available because of the difficulties anticipated during melting of compounds having high-vapor pressures at the melting temperatures.

Alloys 2-3 and 2-17 will be remelted. There are indications that splashes during casting may have caused inhomogeneities in these ingots.

Testing of alloys on this program has not started.

3. Program for Next Quarter

Dilatometer tests to define the transformation temperature will be conducted.

C. **TASK 3 - DISPERSION-STRENGTHENED MAGNETIC MATERIALS FOR APPLICATION IN THE 1200-1600°F RANGE.**

1. Summary of Technical Progress

- a.) Purchase Orders for 18 prealloyed cobalt base and cobalt-iron base atomized powders were completed and submitted for approval to the NASA Contracting Officer prior to placing the orders with suppliers.
- b.) Specifications for cobalt and cobalt-iron base composite powders containing dispersions of alumina and thoria, and a summary of quotations received were prepared and submitted for review to the NASA Project Manager before preparing Purchase Orders. Later these will require approval by the NASA Contracting Officer.
- c.) Specifications for cobalt and cobalt-iron base extrusions containing dispersions of alumina and thoria, and a summary of quotations received were prepared before writing Purchase Orders.
- d.) A number of discussions were held with possible suppliers and various reports reviewed. Dr. F. R. Morrall of the Cobalt Information Center visited Westinghouse to discuss projects on the dispersion-strengthening of cobalt which his organization had sponsored elsewhere.

2. Discussion

a. **PREALLOYED ATOMIZED POWDERS**

The 18 prealloyed atomized powders to be used in the program are listed in Table II-8. The end product which will be fabricated from each powder and tested is extruded rod. In the first phase of this task, which involves an initial evaluation of various compositions, the amount of dispersed phase is of the order of 10 percent by volume.

The atomizing process will provide an extremely rapid quench of the molten powder particles to the solid state ⁽¹⁾. This insures that the constituent particles of high-melting point, which solidify first, will be dispersed and not have time to grow above submicron size before the cobalt or cobalt-iron matrix has solidified.

(1) - Towner, R. J., "Atomized Powder Alloys of Aluminum", Metal Progress, May 1958, p. 70

TABLE II-8. Atomized Powders

Atomized Powder Number	Nominal Composition (Weight Percent)
1	Pure cobalt
2	Pure cobalt - 2.0 boron
3	Pure cobalt - 1.0 boron - 2.2 titanium
4	Pure cobalt - 1.0 boron - 4.2 zirconium
5	Pure cobalt - 1.0 boron - 4.2 columbium
6	Pure cobalt - 1.0 boron - 8.3 tantalum
7	Pure cobalt - 4.0 cerium
8	Pure cobalt - 2.5 aluminum
9	Pure cobalt - 1.3 beryllium
10	27 cobalt - 73 iron
11	26.5 cobalt - 71.5 iron - 2.0 boron
12	26.1 cobalt - 70.7 iron - 1.0 boron - 2.2 titanium
13	25.6 cobalt - 69.2 iron - 1.0 boron - 4.2 zirconium
14	25.6 cobalt - 69.2 iron - 1.0 boron - 4.2 columbium
15	24.5 cobalt - 66.2 iron - 1.0 boron - 8.3 tantalum
16	25.9 cobalt - 70.1 iron - 4.0 cerium
17	26.3 cobalt - 71.0 iron - 2.7 aluminum
18	26.6 cobalt - 72.0 iron - 1.4 beryllium

Powder Nos. 1 and 10, pure cobalt and 27 percent cobalt-73 percent iron, do not contain alloy additions and will serve as bases for comparison with the other compositions. The two percent boron additions in powder Nos. 2 and 11 will be present partially in solid solution and partially as dispersed particles of metal boride compound. The change in solid solubility of boron with temperature is slight, so that the amount of metal boride particles out of solution will be approximately the same at room and elevated temperatures.

The boron additions in powder Nos. 3 through 6 and 12 through 15 are for the purpose of formation of submicron boride particles containing titanium, zirconium, columbium, and tantalum which are stable at high temperatures and resist agglomeration. Powder Nos. 7 and 16 contain cerium and will provide another dispersion-hardening agent, the intermetallic compound CeCo_5 . The aluminum and beryllium containing powders, Nos. 8, 9, 17, and 18, will be given internal oxidation treatments to form Al_2O_3 and BeO dispersed particles at elevated temperatures where the aluminum and beryllium are in solid solution. Also, the cerium containing alloys Nos. 7 and 16 may be considered for internal oxidation treatments.

Based on technical capability for meeting our specifications, the number of compositions quoted on, price, and delivery time, purchase orders were prepared for buying 14 powders from Hoeganaes Sponge Iron Corporation and four powders from Domtar Chemicals Ltd. All four vendors who quoted would undertake to atomize these compositions on an experimental effort basis. Discussions were held with suppliers concerning ways of insuring homogeneity of composition of powder particles from start to end of a run. Homogeneity is assured by the following production methods the vendors will use: (1) induction melting and holding which exerts a stirring action on the melt, and (2) the short production time (about 5 minutes) to atomize a 100 lb. melt. Sampling of the powder stream during atomizing to check composition is not possible in the supplier's equipment.

b. COMPOSITE POWDERS

The compositions of the powders for which quotations were solicited from suppliers are listed in Table II-9. The amount of dispersed oxide is constant, 10 percent by volume. The iron powders Nos. 5 through 8 can be mixed with the cobalt powders to produce cobalt-iron base compositions. Another approach is to purchase cobalt-

TABLE II-9. Composite Powders

Composite Powder Number	Nominal Composition (Weight Percent)	Particle Size of Oxide (Microns)
1	95.25 Co + 4.75 Al ₂ O ₃	0.01-0.06
2	95.25 Co + 4.75 Al ₂ O ₃	0.1-0.6
3	88.8 Co + 11.2 ThO ₂	0.01-0.06
4	88.8 Co + 11.2 ThO ₂	0.1-0.6
5	94.7 Fe + 5.30 Al ₂ O ₃	0.01-0.06
6	94.7 Fe + 5.30 Al ₂ O ₃	0.1-0.6
7	87.6 Fe + 12.4 ThO ₂	0.01-0.06
8	87.6 Fe + 12.4 ThO ₂	0.1-0.6
9	25.6 Co + 69.2 Fe + 5.20 Al ₂ O ₃	0.01-0.06
10	25.6 Co + 69.2 Fe + 5.20 Al ₂ O ₃	0.1-0.6
11	23.7 Co + 64.2 Fe + 12.1 ThO ₂	0.01-0.06
12	23.7 Co + 64.2 Fe + 12.1 ThO ₂	0.1-0.6

iron powders Nos. 9 through 12 directly. Selection of the final approach and compositions to be purchased will be justified in terms of the technical capability of the supplier, processing technique used to make the material, price and delivery time.

The oxide particles will be dispersed in metal particles or the oxide and metal phases will be intermixed so that when the powders are fabricated into extrusions an even distribution of the submicron oxide particles in the metal matrix will result.

The summary of quotations received on composite powders in accordance with our specifications shows that four vendors can supply some or all of the powders made by the following processes: (1) coating of suspended oxide core particles by precipitation of metal from aqueous solution, (2) high intensity arc co-vaporization of mixed oxides of cobalt or cobalt-iron, aluminum, and thorium followed by hydrogen reduction of cobalt or cobalt-iron oxides to metal, (3) a proprietary semi-metallic powder process, and (4) mechanical mixing of metal and alumina powders, and thermal decomposition of thorium salt on metal powder to introduce thoria.

c. EXTRUSIONS OF DISPERSION-STRENGTHENED COBALT
 AND COBALT-IRON ALLOYS

Quotations from suppliers have been received on the compositions listed in Table II-10. All compositions contain 10 percent by volume of oxide with the exception of No. 9 which contains two percent by volume. Selection of the final compositions will be based on technical capability of the vendor and processing technique used to make the material.

Suppliers of extrusions of dispersion-strengthened cobalt and cobalt-iron have proposed the following processes in accordance with our specifications: (1) mechanical mixing of metal and alumina powders, and thermal decomposition of a thorium salt on metal powder, and (2) co-dissolving the desired elements in a volatile solvent, flash-drying, grinding and reduction of matrix oxides. After the above powders are prepared, compacting, sintering, and extrusion are performed.

TABLE II-10. Supplier Extrusions

Extrusion Number	Nominal Composition (Weight Percent)	Particle Size of Oxide (Microns)
1	95.25 Co + 4.75 Al ₂ O ₃	0.01-0.06
2	95.25 Co + 4.75 Al ₂ O ₃	0.1-0.6
3	88.8 Co + 11.2 ThO ₂	0.01-0.06
4	88.8 Co + 11.2 ThO ₂	0.1-0.6
5	25.6 Co + 69.2 Fe + 5.20 Al ₂ O ₃	0.01-0.06
6	25.6 Co + 69.2 Fe + 5.20 Al ₂ O ₃	0.1-0.6
7	23.7 Co + 64.2 Fe + 12.1 ThO ₂	0.01-0.06
8	23.7 Co + 64.2 Fe + 12.1 ThO ₂	0.1-0.6
9	97.74 Co + 2.26 ThO ₂	0.01-0.06

3. Program for Next Quarter

- a.) After purchase orders for prealloyed atomized powders have been approved by the NASA Contracting Officer, the orders will be placed with vendors.
- b.) Purchase orders for composite powders and extrusions of dispersion-strengthened cobalt and cobalt-iron alloys will be submitted for approval to the NASA Contracting Officer.
- c.) Once purchased materials are received, processing and testing will be initiated.

D. TASK 4 - CREEP TESTING.

1. Summary of Technical Progress

a.) A reliable vacuum-creep capsule design previously developed by Westinghouse was modified and the design improved for use on NAS 3-6465.

b.) Nivco alloy (72Co-23Ni and other elements) was selected for the 5,000 to 10,000-hour vacuum creep tests.

c.) Sufficient Nivco bar was obtained from Westinghouse programs to conduct the vacuum creep tests. Official approval to test Nivco alloy was obtained from NASA. Creep test data will be included in subsequent reports as it becomes available after the start of testing.

2. Discussion

Figure II-3 represents the vacuum creep capsule design to be used in the elevated temperature tests on magnetic materials for rotor application. The design was the result of a Westinghouse program where various vacuum chamber designs, pumping systems, and extensometers were evaluated to determine a system which could provide meaningful data at reasonable cost. Included in the extensometer evaluation were measurements of strain using mechanical methods located at various positions along the specimen and the use of strain gages. A slab of Nivco alloy bar was obtained from Westinghouse programs for testing on NAS 3-6465. The composition of the alloy to be tested is listed below in weight percent.

<u>C</u>	<u>Co</u>	<u>Ni</u>	<u>Mn</u>	<u>Si</u>	<u>Ti</u>	<u>Al</u>	<u>Zr</u>	<u>Ingot No.</u>
0.01	Bal	22.5	0.32	0.15	2.15	0.28	0.95	AC-232

Creep sample blanks were cut from the slab which had been solution annealed at 1725°F in air for one hour and water quenched. The blanks were rough machined and will be aged for 25 hours at 1225° ± 5°F.

As part of the program, a series of vacuum measurements were made on the proposed capsule configuration to evaluate pressure losses in the design. These included both hot and cold capsule vacuum measure-

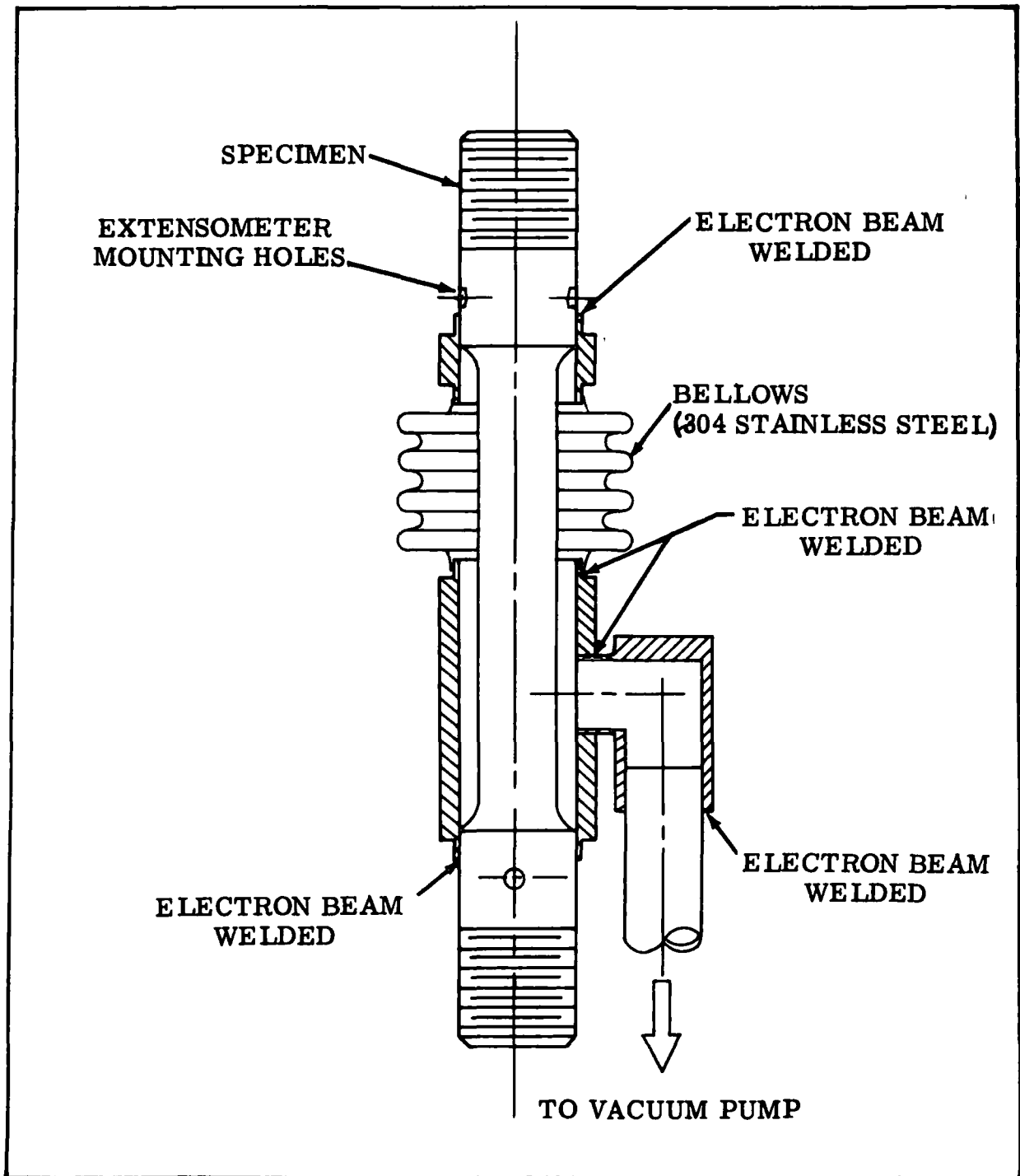


FIGURE II-3. Encapsulated Vacuum Creep Specimen

ments at a position on the upper end of the capsule (position of highest pressure) as well as at a position upstream from the vacuum pumps. Because the total capsule is heated, the ionization gage could not be located at the capsule top end; therefore, data were obtained by brazing a 12-inch extension onto the capsule and locating the ionization gage at the top of this extension outside of the hot zone. After a 48-hour bakeout, a cold-trapped, oil diffusion pumped capsule at 860°F achieved a pressure of 8.2×10^{-7} torr at the capsule extensions.

A similar series of vacuum measurements were made using a four liter per second ion pump. After 22 hours at 570°F, the pressure at the capsule extension was observed to be 6.1×10^{-7} torr and at the pumping port it was observed as 2.1×10^{-7} torr. After 23 hours and an increase in temperature to 1050°F, the pressure was 5.5×10^{-7} torr at the pump port and 3×10^{-6} torr at the top of the capsule extension. A new ion-pumping system of twice the capacity is being procured and will be available by April 1 to use in the test program. It should allow improved vacuum conditions during testing over that observed with the older ion pump. In addition, the removal of the capsule extension will reduce the chamber volume being pumped.

As part of a Westinghouse program to determine bellows reliability, one type of 304 stainless steel bellows has been held at 1350°F in air for more than 4000 hours. Twice monthly the bellows is removed from the furnace and leak checked using a mass spectrometer. As of this date, the bellows has remained leak tight. A second bellows was removed from an actual creep test capsule at 2056 hours for metallographic examination. An examination failed to reveal any degree of deterioration in the bellows. No evidence of intergranular penetration by oxygen was present.

3. Program for Next Quarter

- a.) Creep testing at 1000°F and 1100°F will be initiated.

SECTION III

PROGRAM II - HIGH TEMPERATURE CAPACITOR FEASIBILITY

This program will study the feasibility of building a lightweight capacitor suitable for operation up to 1100°F. It will utilize high-purity dielectric materials and specialized fabrication methods. The ultimate application is in lightweight, high-temperature, power-conditioning equipment suitable for space application. The design goal is a capacitor of 100 to 200 microfarad volts per cubic inch. This range is comparable with conventional low-temperature capacitors.

A. SUMMARY OF TECHNICAL PROGRESS.

- 1.) Major equipment items purchased by Westinghouse for machining and electroding single wafer capacitors are operating satisfactorily. The items include a precision wafering machine, an ultrasonic cutter and machine tool, a vibratory lapping/polishing machine, an ultrasonic cleaner and a three electrode or "triode" sputtering system.
- 2.) Lucalox rods have been reproducibly sliced to a thickness of 10 mils. These wafers were lapped and polished to 6 mils. Grain pull outs have not been completely eliminated and represent a potential problem area particularly for wafers lapped to thicknesses less than 6 mils.
- 3.) A one inch square wafer of pyrolytic boron nitride (Boralloy, High Temperature Materials, Inc.) sliced from a 1 x 1 x 1/8 inch block of "as received" material was lapped to a thickness of 1.5 mils. In this thickness range, the material is very flexible, translucent and free of pin holes or pull outs.
- 4.) An order has been placed for six groupings of differently processed BeO wafers lapped to 10 mils. The vendor, Consolidated Ceramics and Metallizing Corp., will machine the wafers from commercial stock ranging in purity from 99.8 to 99% BeO and one group of wafers will be hot pressed. The other materials are sintered from dry pressed and isostatically pressed BeO.

5.) Five suppliers are being contacted concerning the availability of ultra high purity sputtering target materials of rhodium and platinum (for capacitor electrodes).

6.) A wafer holding fixture is being fabricated for sputtering noble metal electrodes through glass masks. Preliminary designs were made for a small vacuum test furnace with tantalum radiation shields. The furnace will be set up in the CV-18 evaporator (Consolidated Vacuum Corporation) which has been evacuated to about 5×10^{-8} torr with a liquid nitrogen cooled baffle.

B. DISCUSSION.

1. Dielectric Materials Selection

Four dielectric materials are presently being considered for single layer capacitor evaluation in terms of fabricability and electrical properties. Based on these results, a final selection will be made. It is planned to obtain actual chemical analysis on the material selected for preparing multi-layer and monolithic type capacitors.

Included in the group of materials initially selected are aluminum oxide (single crystal and polycrystalline), pyrolytic boron nitride and beryllium oxide. It will be necessary to perform precision machining, lapping and polishing operations on these as received materials to achieve the desired test capacitor thicknesses and surface finishes for comparative evaluation. Electrical data will be obtained on single wafer capacitors electroded by sputtering platinum and/or rhodium films on opposing surfaces. A flow diagram is shown in Figure III-1 to illustrate this approach.

A tabulation of significant properties for a total of ten different materials considered as possible candidates is shown in Table III-1. Included in the table are all of the materials which are known to be available commercially that have superior electrical properties at elevated temperatures. Table III-2 shows a typical spectrochemical analysis for each specific material selected for preliminary study. It is noteworthy that although all the materials listed have very low impurity contents, pyrolytic boron nitride (Boralloy, High Temperature Materials, Inc.) has a total metal impurity level of less than 0.0003%. The oxygen content of this material is not given; however, it is generally considered that the oxygen as B_2O_3 in pyrolytically deposited boron nitride is substantially lower than it is for the hot pressed material.

2. Single Wafer Fabrication

In addition to a materials property tabulation, Table III-1 summarizes the present status of the program as indicated in the appropriate columns.

The major equipment items obtained for preparing single wafer capacitors are:

- a.) WMSA Precision Wafering Machine manufactured by Micro-mesh Manufacturing Corporation. The machine is equipped with a variable spindle speed motor and two variable longitudinal table speed ranges (0 to 1/4 and 0 to 3 inches per minute).

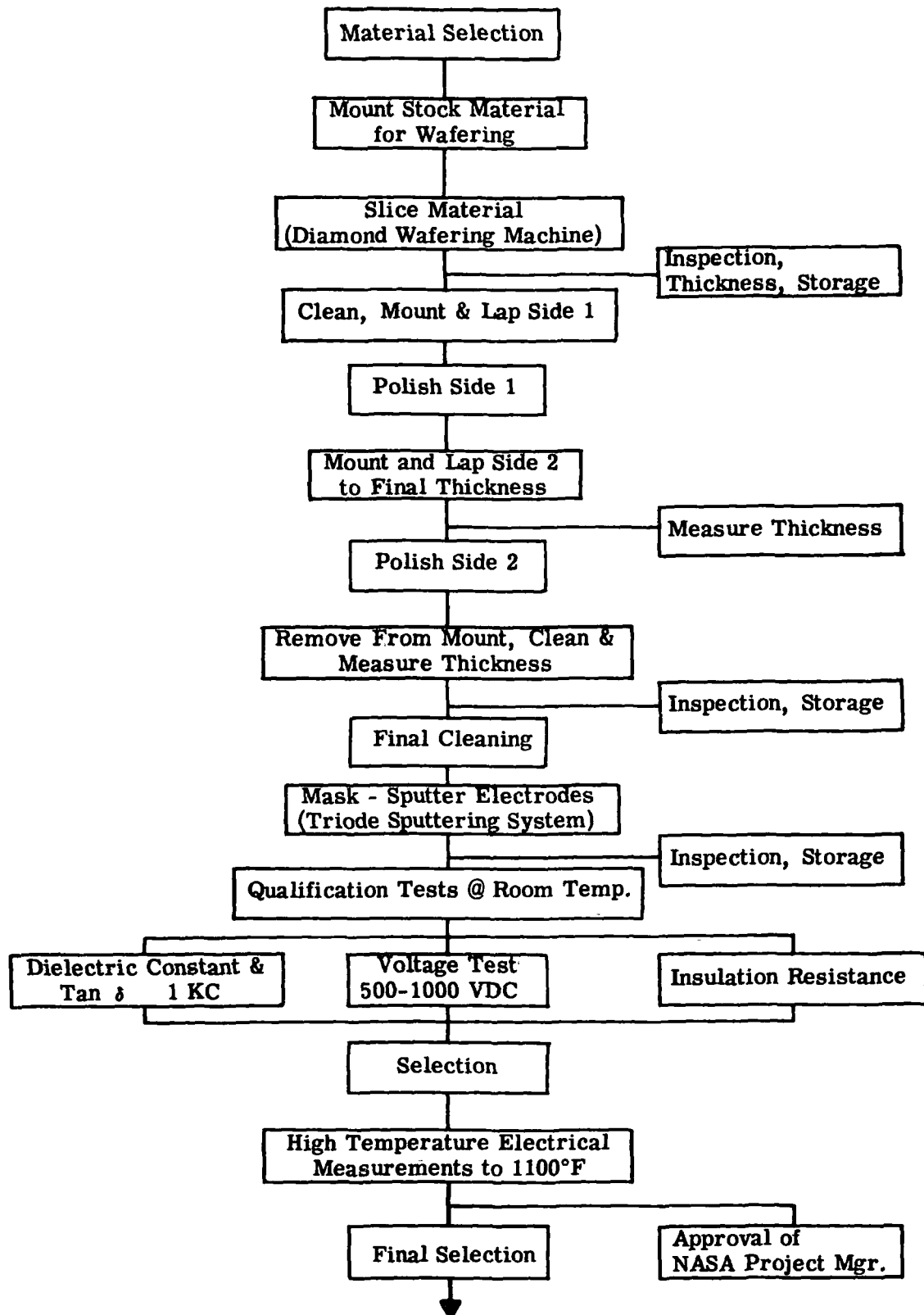


FIGURE III-1. Preparation and Evaluation of Single Layer Capacitors

TABLE III-1. Tabulation of Dielectric Material Properties and Present Status

Material	Microstructure	Process Method	Mfg. or Source and Trade Name	Form (As Received)	Purity	Density - % of Theoretical	Present Status
Al ₂ O ₃	Single Crystal	Verneuil (flame fusion)	Linde Div. Union Carbide Corp.	"Boule" ~ 3/4" Diameter x 3" long	< 100 ppm impurities (2)	~ 100	Two boules on hand
Al ₂ O ₃	Polycrystalline	Pressed and Sintered	General Electric "Lucalox"	Rods 3/4 to 1-1/4" diameter	99.9% Al ₂ O ₃ (2)	~ 100	0.006" thick wafers - sliced, lapped & polished
Al ₂ O ₃	Polycrystalline	Hot Pressed	(W)AED (experimental)	Disks	Starting Material < 100 ppm impurities	~ 100	Several 0.370 X 0.010" disks available
BN	Hexagonal layer High degree of orientation	Pyrolytic decomposition of BCl ₃ and NH ₃	High Temperature Materials Inc. "Boralloy"	Plates 1 X 1 X 1/8" 2 X 2 X 1/4"	Total impurities \cong 100 ppm (2)	~ 98	0.001 to 0.002" wafers - sliced, lapped & polished
BN	Polycrystalline	Hot Pressed	Carborundum Company	None Ordered	97% BN	> 93.4	Not selected as candidate material
BN	Polycrystalline	Hot Pressed	National Carbon Company	None Ordered	~ 95 to 97% BN	~ 90	Not selected as candidate material
BeO	Polycrystalline	Pressed and Sintered	Coors BD 99.5	(1)	99.5% BeO	~ 95	(1)
BeO	Polycrystalline	Pressed and Sintered	American Lava Corp. Alsimag 754	(1)	99.5% BeO		(1)
BeO	Polycrystalline	Hot Pressed	Atomics International (experimental)	Not known at present	Starting material - Minox AAA (2)	~ 99.7	Sample requested
MgO	Single Crystal	Cooled Melt	Norton Co. "Magnorite"	None Ordered	99.9% MgO	~ 100	Not selected as candidate material

1. An order has been placed with Consolidated Ceramics Inc. for machined disks of
2. See Table III-2 for spectrochemical analysis of materials.

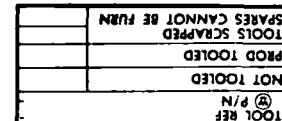
ELECTRICAL PROPERTIES				Remarks	Data Reference
DC Resistivity Ohm-Cm (Approximate Values)	Tan δ	Dielectric Constant	Electric Strength (Voltage Breakdown)		
400°C - 7×10^{12} 600°C - 7×10^{10}	@ 10^5 cps \perp to optic axis 400°C - 0.0002 600°C - 0.001	@ 10^5 cps 60° to optic axis K@20°C - 9 K@900°C - 10	1700 volts/mil @ 60 cps, 20 mil thick sample (Room temper- ature)	No electrical breakdown strength at elevated temper- ature. Tan δ and dielectric constant data not given for 60 cps to 50 kc freq. vs. temp.	Linde Industrial Crystals Bulletin F-814-C F-917-B
400°C - 1×10^{13} 600°C - 3×10^{10}	@ 9720 mc 20°C - 0.000025	@ 9720 mc 20°C - 9.9	1700 volts/mil @ 20°C, 20 mil thick sample	Low frequency (60 cps to 50 kc) Tan δ and dielectric con- stant not given. High temper- ature voltage breakdown not given.	Company Data Bulletin
400°C - 5×10^{12} 600°C - 9×10^{10}	Not Measured	Not Measured	Not Measured		AF33(615)1360
400°C - 10^{14} 600°C - 10^{13}	@ 4kmc, "a" direction RT to 649°C, ap- proximate 0.0004 over the temperature range	K3.4 "c" direction @ RT, not measured at elevated temperature	4000 volts/mil in "c" direction, d-c 10 to 20 mil sample (not measured at elevated temper- ature)	Low frequency data not avail- able (vs. temperature) (60 cps to 50 kc)	Materials in De- sign Eng., Feb 1964 (M. Bosche & D. Schiff)
420°C - 1.7×10^{11} 550°C - 8.3×10^8 660°C - 3.4×10^7	@ 10^3 cps 400°C - 0.012 600°C - 0.14	@ 10^3 cps 400°C - 4.5 600°C - 6.5	1450 volts/mil @ RT, sample 10 mils thick	Not selected as candidate material because of low density.	Company Data Bulletin
482°C - 5×10^8 1000°C - 1×10^7	@ 10^3 cps 300°C - 0.015 575°C - 1.0	@ 10^3 cps 21°C - 4.4 300°C - 4.52	500-1000 volts/ mil, thickness not known	Not selected as candidate material because of low density.	Company Data Bulletin H-8745
300°C - $>10^{15}$ 500°C - 5×10^{13} 700°C - 1.5×10^{10}	No low frequency data given	@ 1 mc 6.7	700 volts/mil average RMS at RT, 10 mil sample thickness		Company Data Bulletin
400°C - 10^{13} 600°C - 10^{12}	No low frequency data given	~7	None given		Company Data Bulletin
No measurements	No measurements	No measure- ments	No measurements	Sample requested.	Private com- munication from R. L. McKisson
600°C - 1.6×10^{12} 1000°C - 10^8	@ 100 cps 25°C - 0.0003	@ 100 cps 25°C - 9.65	No data given	Not selected as candidate material because of hydro- scopic characteristics.	Norton Co. Memo Feb. 1, 1963

O from various commercial sources

TABLE III-2. Spectrochemical Analysis of Candidate Dielectrics

MATERIAL	ELEMENT											
	Al	Fe	Mg	Ti	Mn	V	Na	Cu	Ni	Ca	Cr	Ga
Lentalox (General Electric Co.)	Percent											
	Major	0.002	0.15	(1)	(1)	(1)	(1)	(1)	(1)	0.004	(1)	(1)
Linde Sapphire (Single Crystal Al_2O_3)	Parts Per Million											
	(1)	< 2	< 2	(2)	(2)	(2)	(2)	(2)	(2)	9	(2)	(2)
Boralloy Boron Nitride (3) (Pyrolytic Bn) (High Temperature Materials Inc.)	Percent (%)											
	(1)	(1)	(1)	(1)	(1)	(1)	(1)	0.0001	(1)	0.0001	(1)	(1)
Minox AAA BeO powder used to prepare hot pressed materials by Atomics International	Parts Per Million											
	75	35	50	1	1	--	75	1	7	80	15	--
<p>(1) Not detected.</p> <p>(2) These elements not listed.</p> <p>(3) Total impurities less than 0.003%. Thirty-four additional elements listed as not detected.</p>												

											REFERENCE SOURCE
Si	Mo	B	Zr	Cd	Pb	Ba	Bi	Co	Sm	Li	
											ASD TR-61-628, Part II Studies of the Brittle Behavior of Ceramic Materials, April 1963 - Contract AF33(616)7465
0.03	(2)	(2)	(2)	(2)	(2)	(2)	(2)	(2)	(2)	(2)	
ppm)											Letter from Mr. B G Benak (Linde Co)
6	(2)	(2)	(2)	(2)	(2)	(2)	(2)	(2)	(2)	(2)	
											Company Data Sheet Dated February 1, 1965
0.0001	(1)	Major	(1)	(1)	(1)	(1)	(1)	(1)	(1)	(1)	
m)											Letter from R. L. McKisson (Atomics International)
25	< 3	< 1	< 30	< 1	< 2	< 5	< 5	< 1	< 5	< 5	

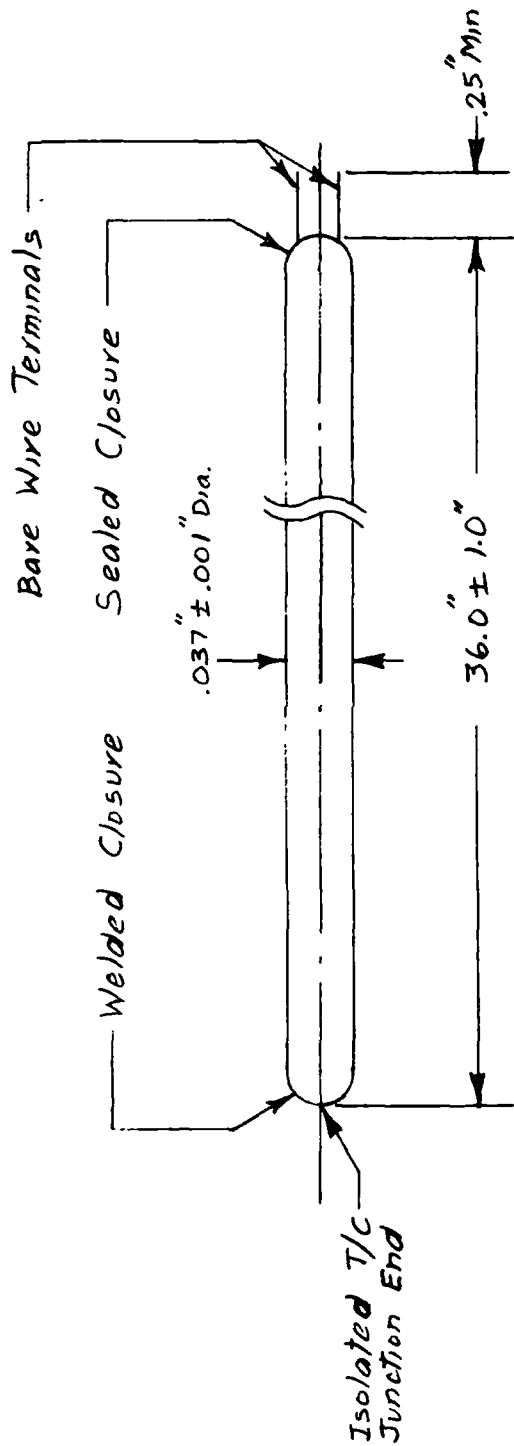


SPACER ~ MAKE FROM MINIMUM 99% PURITY HIGH DENSITY Al_2O_3 (ALUMINUM)

[illegible]

1-800-855-8888

OMR-EDSK 326786



Note: Thermocouples are to be kept straight at all times.

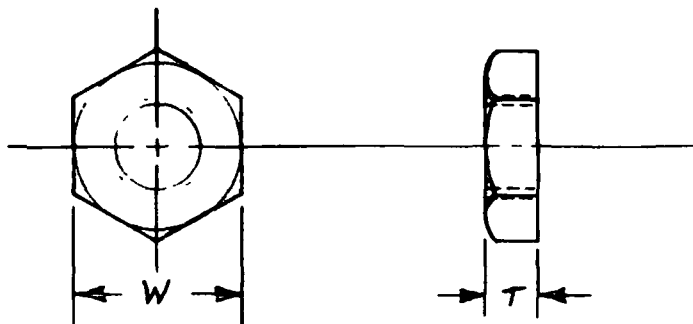
SINGLE DIAMETER SHEATHED THERMOCOUPLE

Reference (W) D-Spec # 709747

EDSK 326796

WJ 2-18-68

WESTINGHOUSE ELECTRIC CORPORATION



Material - Hastelloy Alloy B Rod

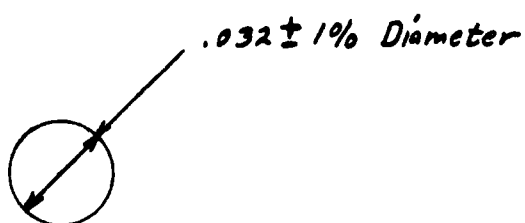
Dash No.	Thread	W $\pm .005$	T $\pm .008$	Stock
1	#8 (.164)-32 NC-2B	.338	.187	1/2 Dia
2	1/4-28 UNF-2B	.428	.219	5/8 Dia

NUT - HEXAGONAL

W.L. Grant 4-21-65 EDSK 327503

WESTINGHOUSE ELECTRIC CORPORATION

Conductor
20% Nickel-Clad Silver



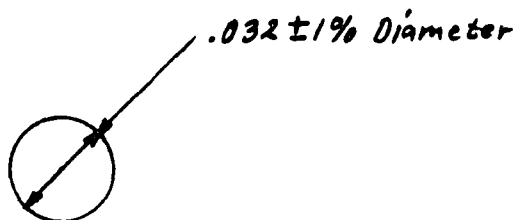
Dimensions in inches
Scale 20:1

EDSK 326533

H.E. Keneipp 12-21-64

WESTINGHOUSE ELECTRIC CORPORATION

Conductor
28% Inconel-Clad Silver



Dimensions in inches
Scale 20:1

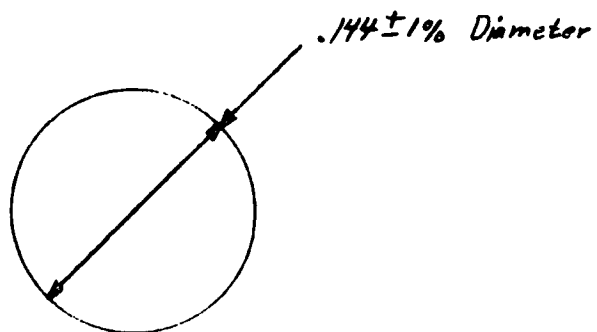
Rev. A - 12-29-64
28% was 20% *mf*

EDSK 326534 A

H.E. Keneipp 12-21-64

WESTINGHOUSE ELECTRIC CORPORATION

Conductor
20% Nickel-Clad Silver



Dimensions in inches
Scale 10:1

EDSK 926535

H. E. Keneipp 12-21-64

APPENDIX A

STATOR MATERIALS SUMMARY

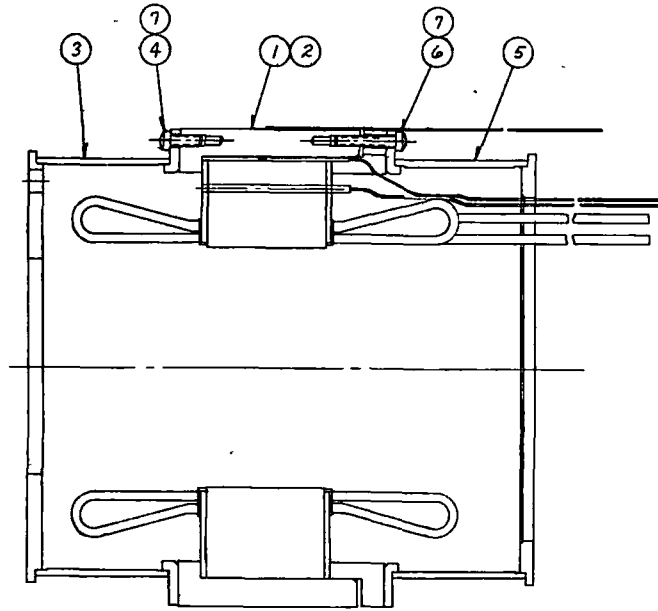
APPENDIX A

STATOR ASSEMBLY

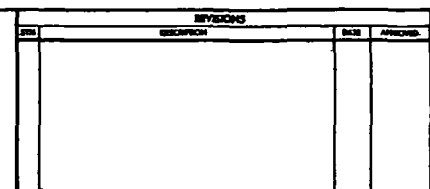
DRAWING NO.	TITLE	MATERIAL
EDSK 326681	Stator Assembly	-
EDSK 326618	Stack, Stator, Wound	-
EDSK 326602	Stack, Stator, Wound (Practice) (1)	-
EDSK 326550	Details	See Drawing
EDSK 326603	Details	See Drawing
EDSK 326620	Details	See Drawing
EDSK 326622	Details	See Drawing
EDSK 326619	Forging	Hiperco 27 Alloy (27 Co - Fe)
EDSK 326793	Forging, Cut Up	Hiperco 27 Alloy (27 Co - Fe)
EDSK 326549	Punching Stator	Hiperco 27 Alloy (27 Co - Fe)
EDSK 326686	Insulator, Tube	99 Al ₂ O ₃
EDSK 326680	Insulator, Tube	99 Al ₂ O ₃
EDSK 326626	Details, Slot	99 Al ₂ O ₃
EDSK 326625	Insulation, Liner	99 Al ₂ O ₃
EDSK 326794	Insulation, Tube	99 Al ₂ O ₃
EDSK 326795	Thermocouple	99 Al ₂ O ₃
EDSK 326796	Thermocouple	(Inconel Cladding, Platinel II)
P15C8278-12	Pin, Spring	wire system, Al ₂ O ₃ Insulation
EDSK 327502	Screw	CRES-AMS 5506
EDSK 326531	Conductor	Hastelloy Alloy B
EDSK 326532A	Conductor	Nickel Clad Silver (20% Clad area) (2)
		Inconel Clad Silver (28% Clad area) (2)

(1) - Used in a program to define conductor winding methods.

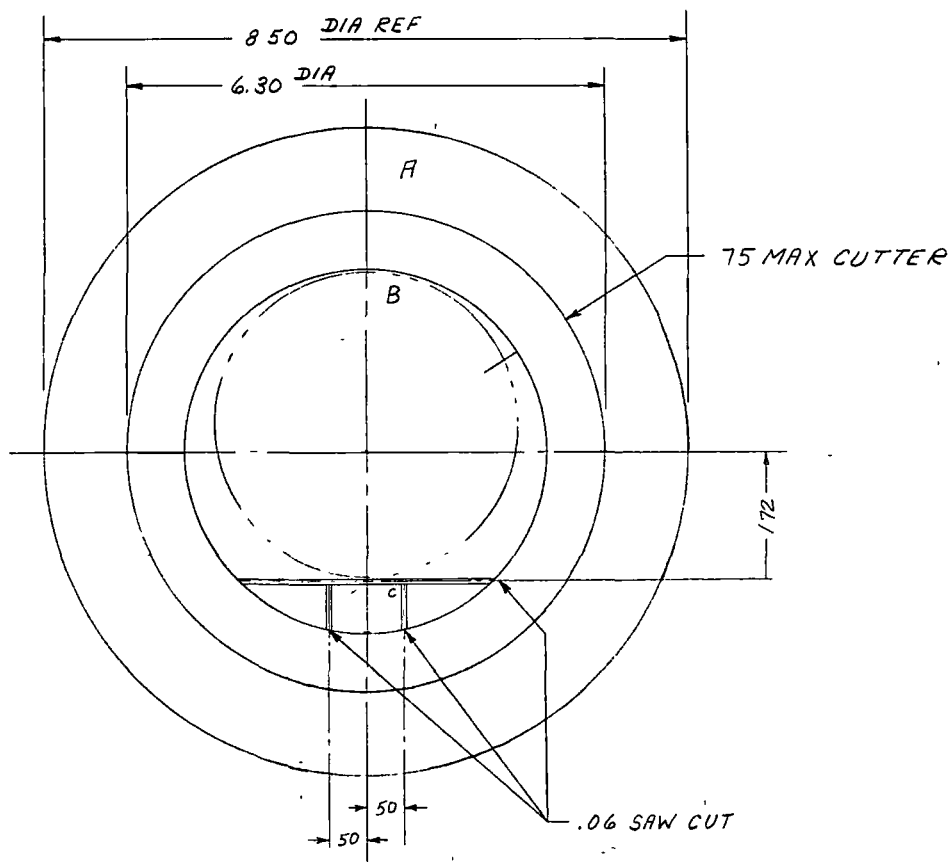
(2) - All conductors insulated with Anaconda's Anadur, a refractory-oxide-filled glass insulation.



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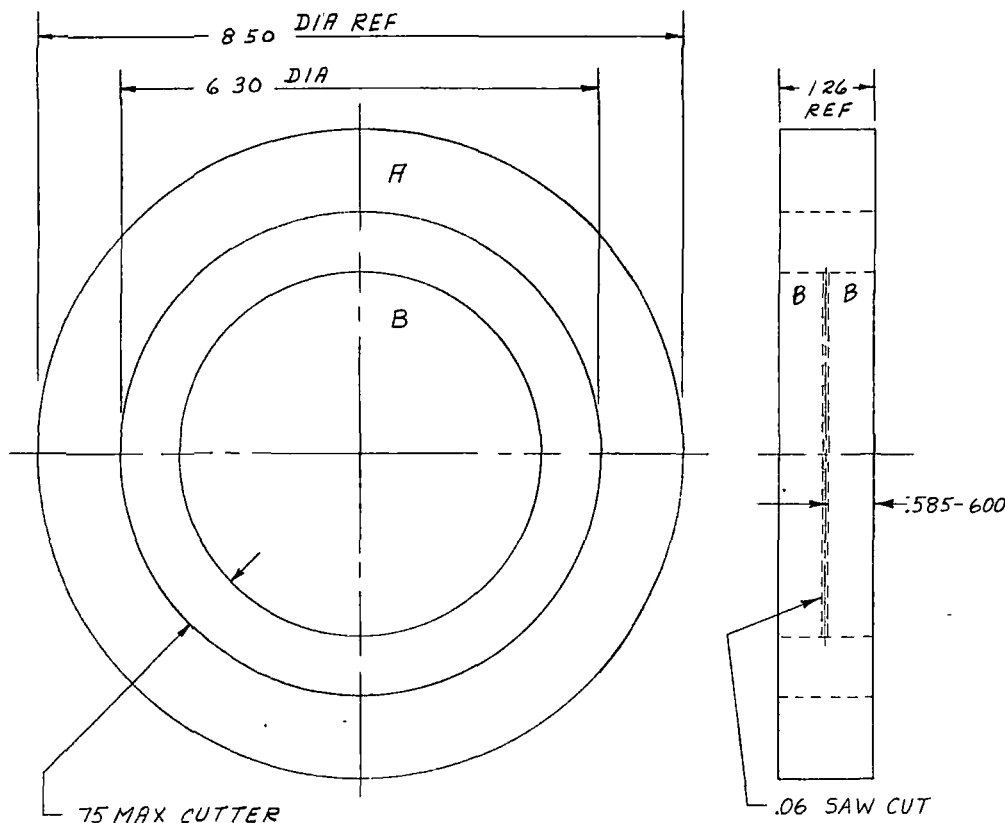


CUT UP FORGING EDSK 326619 ITEM NO 3 AS SHOWN. PART "A" TO BE USED TO MAKE FRAME EDSK 326603 ITEM NO 4, PART "B" TO BE USED TO MAKE HOUSING EDSK 326638 ITEM NO 2 PART "C" TO BE USED TO MAKE BAR EDSK 326688 ITEM NO 6

ITEM

5-27-65	INTERNAL INFO & DATA
REVISOR	REVISOR
WIDTH FROM .38 TO	WIDTH FROM .38 TO
75 MAX REVISED	75 MAX REVISED
PICTURE TO MATCH	PICTURE TO MATCH
1	WL Grant 5-1-65
ITEM 2 - RT VIEW	ITEM 2 - RT VIEW
ADDED SAW CUT	ADDED SAW CUT
4.585-600 DIM	4.585-600 DIM
2	WL Grant 5-26-65

REVISIONS			
SYM	DESCRIPTION	DATE	APPROVED

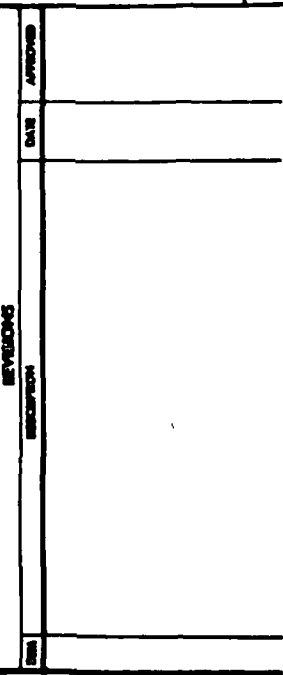


CUT UP FORGING EDSK 326619 ITEM NO 2 AS SHOWN.
PART "A" TO BE USED TO MAKE RETAINING RING EDSK 326603 ITEM NO 5,
AND PART "B" TO BE USED TO MAKE RING EDSK 326688 ITEM NO 5

ITEM NO 2

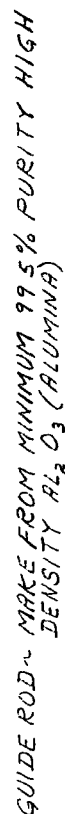
SPECIFICATION REFERENCE	GOVERNMENT SPECIFICATION	NAME OF PROCESS	QTY REQD	SYM	NOMENCLATURE OR DESCRIPTION	PART NO OR IDENT NO	SPECIFICATION	MATERIAL OR NOTE	INTERNAL INFORMATION	ITEM NO
LIST OF MATERIAL OR PARTS LIST										
TOLERANCE AND SPECS REF 1A-155 UNLESS OTHERWISE SPECIFIED DIMENSIONS ARE IN INCHES DO NOT SCALE DIMS FRACTIONS DECIMALS ANGLES 1/16 1/32 1/64 .005 .010 .015 1/8 1/4 1/2 3/4 1 1 1/2 2 3 4 5 6 8 10 12 14 16 18 20 22 24 26 28 30 32 34 36 38 40 42 44 46 48 50 52 54 56 58 60 62 64 66 68 70 72 74 76 78 80 82 84 86 88 90 92 94 96 98 100			DFTM	H90	1/23/65	AEROSPACE ELECTRICAL DEPARTMENT WESTINGHOUSE ELECTRIC CORP LIMA, OHIO, U.S.A.				
PROCESS TO BE APPLIED AS INDICATED ON DRAWING			DFTM	47	12565					
GOVT SPEC			ENGR	W L Grant	1/25/65	TITLE FORGING, CUT UP				
SPEC			ENGR	W L Grant	1-25-65					
STOCKS			MFR			CODE IDENT NO. SIZE DRAWING NO 83843 D EDSK 326793				
WESTINGHOUSE			TEST							
APPD FOR			APPD FOR			SCALE 1 / 1 WEIGHT SHEET				
BY			BY							

EDSK 326793

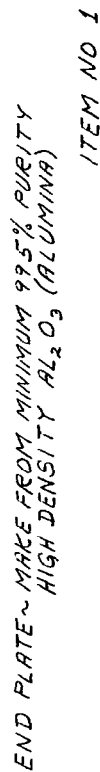
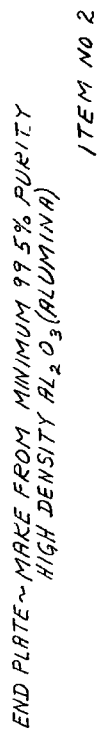


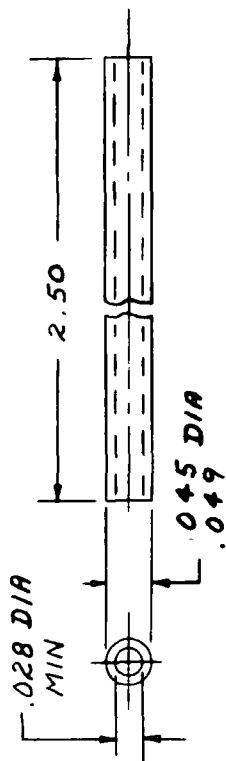
BUSHING ~ MAKE FROM MINIMUM 99.5% PURITY HIGH DENSITY Al_2O_3 (ALUMINA)

FD-5K 326787



UNITED STATES DEPARTMENT OF AGRICULTURE

[illegible]

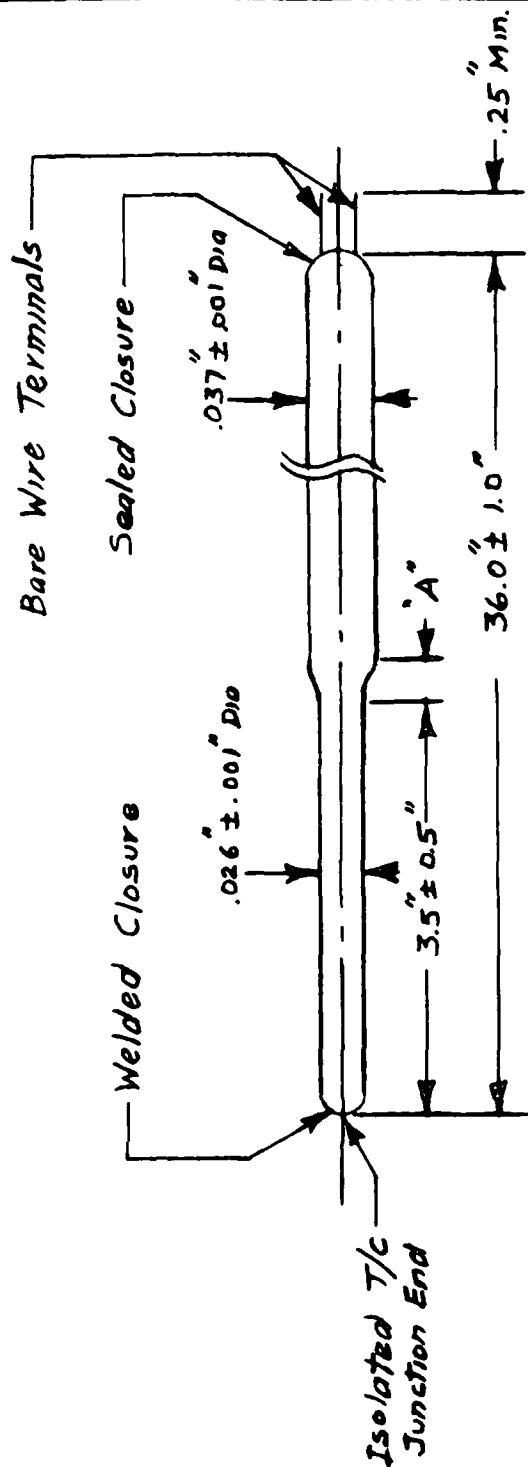


NOTE ~ STRAIGHTNESS TOLERANCE $\frac{1}{16}$ INCH PER FOOT OF LENGTH

INSULATOR ~ MAKE FROM MINIMUM 99% PURITY HIGH DENSITY Al_2O_3 (ALUMINA)

TOOL REP	TOOL NOT TOOK	TOOL TOOK	TOOL SCRAPED	SPARES CANNOT BE USED
Q/P/N				

[illegible]



"A" - Dimension as required
for forming operation -
1.0 inch maximum.
OD surface must be
continuous thru transition.

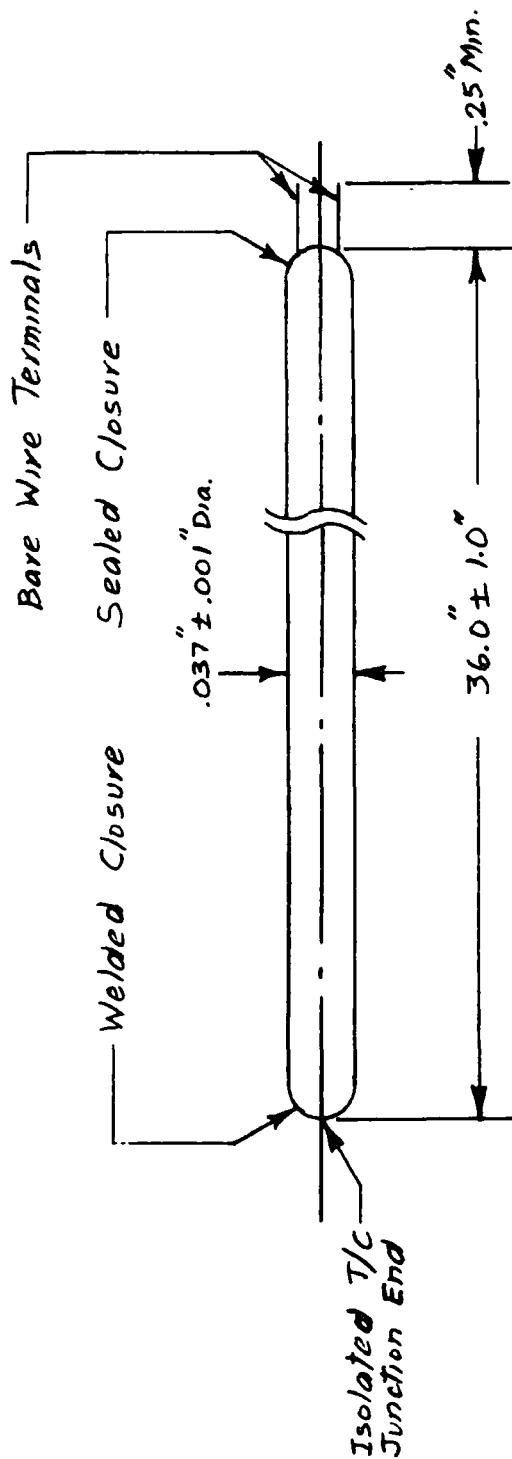
Note: Thermocouples are to be
kept straight at all times.

DUAL DIAMETER SHEATHED THERMOCOUPLE

Reference (W) D-Spec # 709747

EDSK 326795

2-18-65



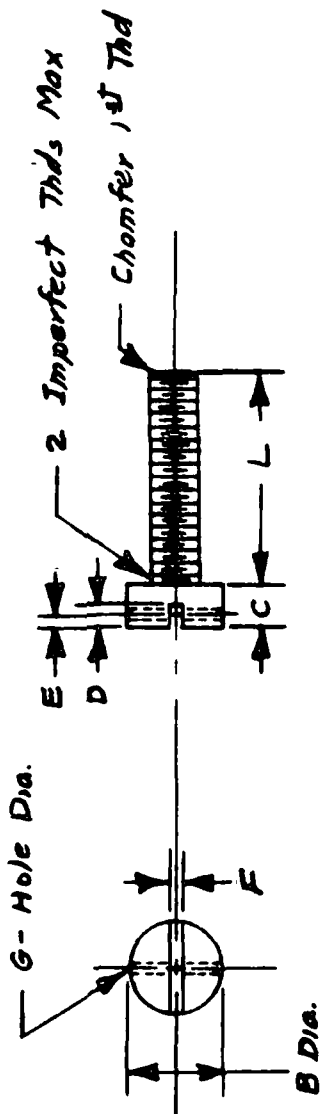
Note: Thermocouples are to be kept straight at all times

SINGLE DIAMETER SHEATHED THERMOCOUPLE
 Reference (W) D-Spec # 709747

EDSK 326796

July 2-18-65

WESTINGHOUSE ELECTRIC CORPORATION



Material - Hastelloy Alloy B Rod

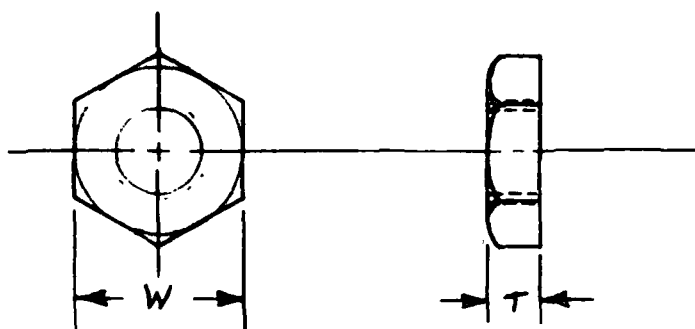
Dash No.	Thread	B Dia	C	D	E	F	G Dia	L	Stock
1	#8 (164)-32 NC-2A	.270	.156	.078	.050	.062	.040	.500	3/8 Dia
2	" " "	.270	.156	.078	.050	.062	.040	.750	3/8 Dia
3	#10 (190)-32 NF-2A	.313	.156	.078	.050	.062	.040	.500	7/16 Dia
4	" " "	.313	.156	.078	.050	.062	.040	1.00	7/16 Dia
5	1/4-28 UNF-2A	.414	.207	.103	.078	.062	.040	.500	1/2 Dia

SCREW - FLAT FILLISTER HEAD

W.L. Grant 4-21-65 EDSK 327502

EDSK 327502

WESTINGHOUSE ELECTRIC CORPORATION



Material - Hastelloy Alloy B Rod

Dash No.	Thread	W ±.005	T ±.008	Stock
1	#8 (.164)-32 NC-2B	.338	.187	1/2 Dia
2	1/4-28 UNF-2B	.428	.219	5/8 Dia

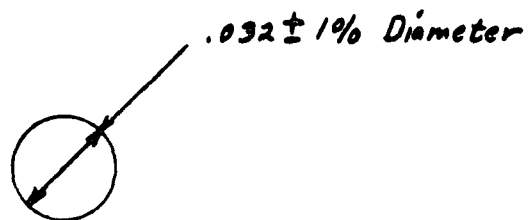
NUT - HEXAGONAL

W.L. Grant 4-21-65 EDSK 327503



WESTINGHOUSE ELECTRIC CORPORATION

Conductor
20% Nickel-Clad Silver



Dimensions in inches
Scale 20:1

EDSK 326533

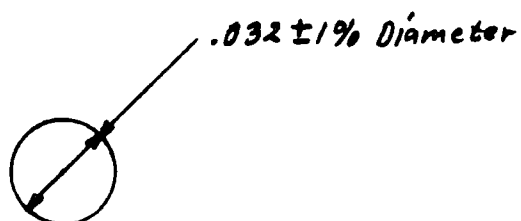
H.E. Keneipp 12-21-64



7
2
1

WESTINGHOUSE ELECTRIC CORPORATION

Conductor
28% Inconel-Clad Silver



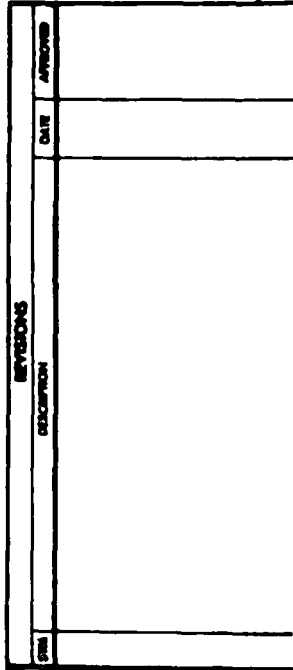
Dimensions in inches
Scale 20:1

EDSK 326534 A

Rev. A - 12-29-64
28% was 20% *ms*

H.E. Keneipp 12-21-64





NOTE ~ STRAIGHTNESS TOLERANCE $1/16$ INCH PER FOOT OF LENGTH

INSULATOR ~ MAKE FROM MINIMUM 99% PURITY HIGH DENSITY Al_2O_3 (ALUMINA)

[illegible][illegible]

SPACER ~ MAKE FROM MINIMUM 99% PURITY HIGH DENSITY Al_2O_3 (ALUMINA)

WEDGE ~ MAKE FROM MINIMUM 99% PURITY HIGH DENSITY Al_2O_3 (ALUMINA)



INSULATOR ~ MAKE FROM MINIMUM 99% PURITY HIGH DENSITY Al_2O_3 (ALUMINA)

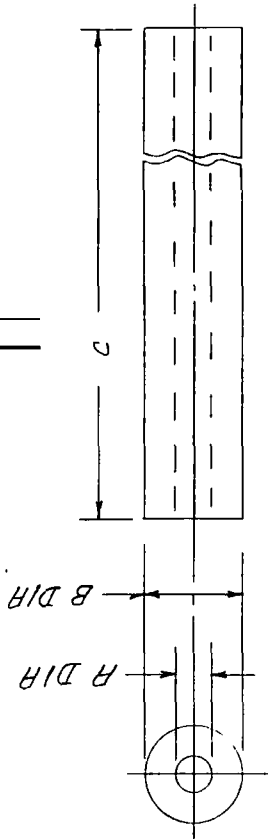
[illegible]

1. THE STATE OF TEXAS, COUNTY OF DALLAS, ss. I, _____, Clerk of the County Court, do hereby certify that the within and foregoing is a true and correct copy of the original as the same appears from the records of the County Court of Dallas County, Texas.

08992E VSD

PART NO	DIA A	DIA B	LENGTH C
-1	0.62	1.25	.70
-2	0.62	.188	2.50
-3	.044	.094	2.50

EDSK 326686

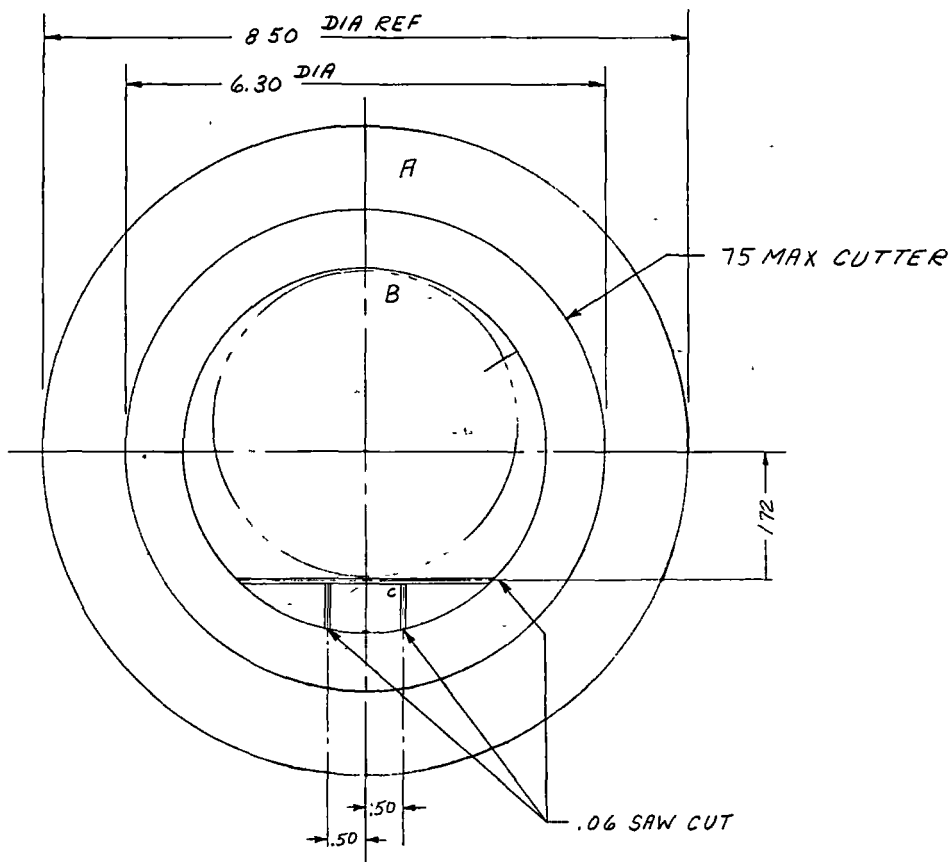


NOTE ~ STRAIGHTNESS TOLERANCE 1/16 INCH PER FOOT OF LENGTH

INSULATOR ~ MAKE FROM MINIMUM 99% PURITY HIGH DENSITY AL₂O₃ (ALUMINA)

5. 47919039 INTERNAL INFO & DATA		3. 47919039 INTERNAL INFO & DATA	
SPECIFICATION REFERENCE GOVERNMENT SPECIFICATION 1 PLACE 2 PLACE 3 PLACE 4 PLACE 5 PLACE 6 PLACE 7 PLACE 8 PLACE 9 PLACE 10 PLACE 11 PLACE 12 PLACE 13 PLACE 14 PLACE 15 PLACE 16 PLACE 17 PLACE 18 PLACE 19 PLACE 20 PLACE 21 PLACE 22 PLACE 23 PLACE 24 PLACE 25 PLACE 26 PLACE 27 PLACE 28 PLACE 29 PLACE 30 PLACE 31 PLACE 32 PLACE 33 PLACE 34 PLACE 35 PLACE 36 PLACE 37 PLACE 38 PLACE 39 PLACE 40 PLACE 41 PLACE 42 PLACE 43 PLACE 44 PLACE 45 PLACE 46 PLACE 47 PLACE 48 PLACE 49 PLACE 50 PLACE 51 PLACE 52 PLACE 53 PLACE 54 PLACE 55 PLACE 56 PLACE 57 PLACE 58 PLACE 59 PLACE 60 PLACE 61 PLACE 62 PLACE 63 PLACE 64 PLACE 65 PLACE 66 PLACE 67 PLACE 68 PLACE 69 PLACE 70 PLACE 71 PLACE 72 PLACE 73 PLACE 74 PLACE 75 PLACE 76 PLACE 77 PLACE 78 PLACE 79 PLACE 80 PLACE 81 PLACE 82 PLACE 83 PLACE 84 PLACE 85 PLACE 86 PLACE 87 PLACE 88 PLACE 89 PLACE 90 PLACE 91 PLACE 92 PLACE 93 PLACE 94 PLACE 95 PLACE 96 PLACE 97 PLACE 98 PLACE 99 PLACE 100 PLACE		MAKE OF PARTS TOOL REF 1 PLACE 2 PLACE 3 PLACE 4 PLACE 5 PLACE 6 PLACE 7 PLACE 8 PLACE 9 PLACE 10 PLACE 11 PLACE 12 PLACE 13 PLACE 14 PLACE 15 PLACE 16 PLACE 17 PLACE 18 PLACE 19 PLACE 20 PLACE 21 PLACE 22 PLACE 23 PLACE 24 PLACE 25 PLACE 26 PLACE 27 PLACE 28 PLACE 29 PLACE 30 PLACE 31 PLACE 32 PLACE 33 PLACE 34 PLACE 35 PLACE 36 PLACE 37 PLACE 38 PLACE 39 PLACE 40 PLACE 41 PLACE 42 PLACE 43 PLACE 44 PLACE 45 PLACE 46 PLACE 47 PLACE 48 PLACE 49 PLACE 50 PLACE 51 PLACE 52 PLACE 53 PLACE 54 PLACE 55 PLACE 56 PLACE 57 PLACE 58 PLACE 59 PLACE 60 PLACE 61 PLACE 62 PLACE 63 PLACE 64 PLACE 65 PLACE 66 PLACE 67 PLACE 68 PLACE 69 PLACE 70 PLACE 71 PLACE 72 PLACE 73 PLACE 74 PLACE 75 PLACE 76 PLACE 77 PLACE 78 PLACE 79 PLACE 80 PLACE 81 PLACE 82 PLACE 83 PLACE 84 PLACE 85 PLACE 86 PLACE 87 PLACE 88 PLACE 89 PLACE 90 PLACE 91 PLACE 92 PLACE 93 PLACE 94 PLACE 95 PLACE 96 PLACE 97 PLACE 98 PLACE 99 PLACE 100 PLACE	
QTY RECD SYM NOMENCLATURE OR DESCRIPTION PART NO OR IDENT NO SPECIFICATION MATERIAL OR NOTE INTERNAL INFORMATION ITEM NO		LIST OF MATERIAL OR PARTS LIST AEROSPACE ELECTRICAL DEPARTMENT WESTINGHOUSE ELECTRIC CORP LIMA, OHIO U.S.A.	
TITLE INSULATOR, TUBE		CODE IDENT NO 83843	
DRAWING NO C EDSK 326686		SCALE N.T.S.	
WEIGHT SHEET		BY	

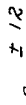
Handwritten signature



CUT UP FORGING EDSK 326619 ITEM NO.3 AS SHOWN. PART "A" TO BE USED TO MAKE EDSK 326603 ITEM NO 4, PART "B" TO BE USED TO MAKE HOUSING EDSK 326688 PART "C" TO BE USED TO MAKE BAR EDSK 326688 ITEM NO 6

SLIP 17039	
INTERNAL INFO & DATA	
REVISED CUTTER MAX WIDTH FROM .38 TO .75 MAX REVISED PICTURE TO MATCH	
1	W.L. Grant 5-1-65
ITEM 2 - RT VIEW ADDED SAW CUT & 585-600 DIM	
2	W.L. Grant 5-26-65

~~SS~~

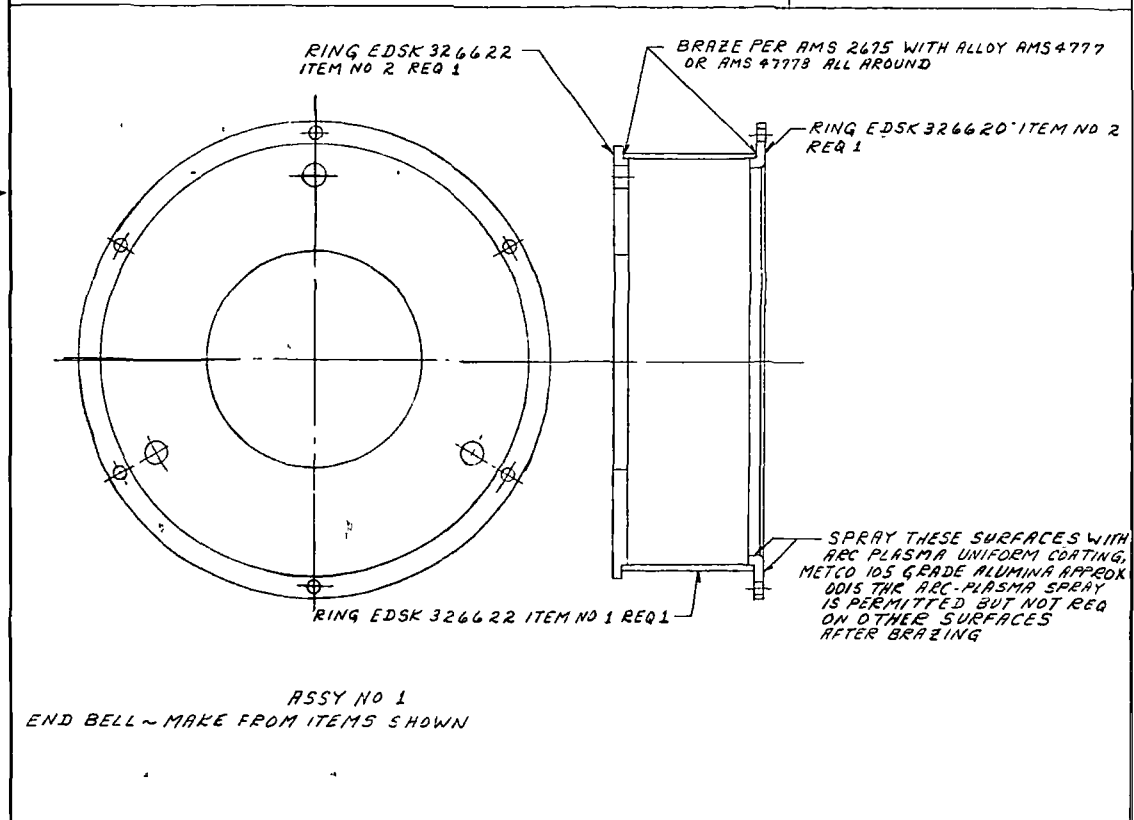
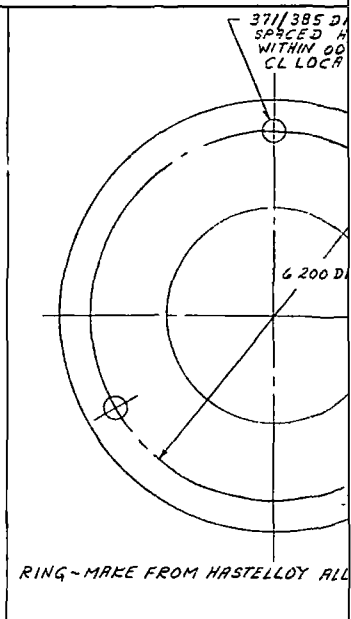
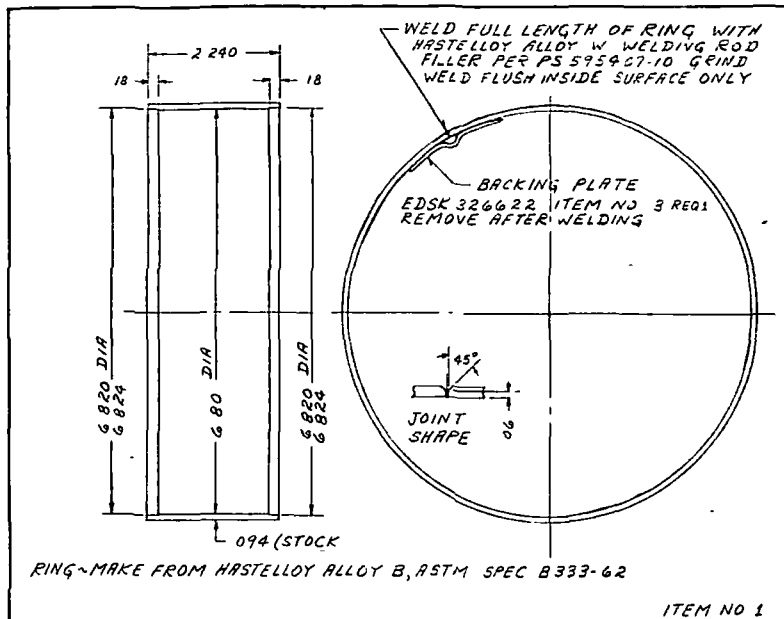


EDSK 326619	DIM "A"
ITEM NO 1	3 40
ITEM NO 2	1 26
ITEM NO 3	4 00

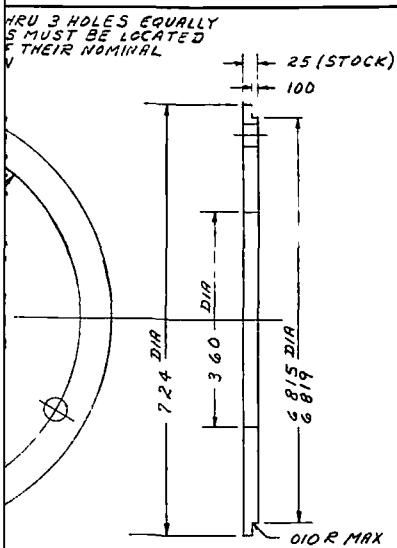
~ NOTE ~
MATERIAL SHALL BE MADE TO MEET WESTINGHOUSE PDS 13225 AB WITH THE FOLLOWING ADDITIONAL REQUIREMENTS 1. THAT CERTIFICATION OF ACCEPTABLY PASSING ULTRASONIC INSPECTION BE PROVIDED 2. THAT THE MATERIAL BE SHIPPED WITHOUT BEING ANNEALED, HOWEVER IN THE ANNEALED CONDITION SAID FORGING SHALL BE CERTIFIED TO MEET THE MECHANICAL AND MAGNETIC PROPERTIES OF PDS 13225 AB

[illegible]

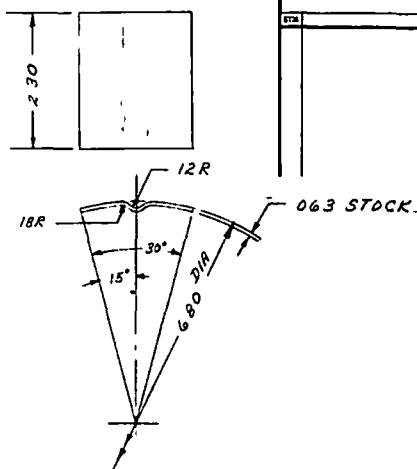
39



5.102132.37
INTERNAL INFO & DATA



ITEM NO 2



BACKING PLATE ~ MAKE FROM STAINLESS STEEL PDS 4562-2

ITEM NO 3

PRAY THESE SURFACES WITH
C-PLASMA UNIFORM COATING,
100 GRADE ALUMINA APPROX
0.15 THK AEC-PLASMA SPRAY
PERMITTED BUT NOT REQ
OTHER SURFACES
TER BRAZING

BRAZE PER AMS 2675 WITH ALLOY AMS 4777
OR AMS 4778 ALL AROUND

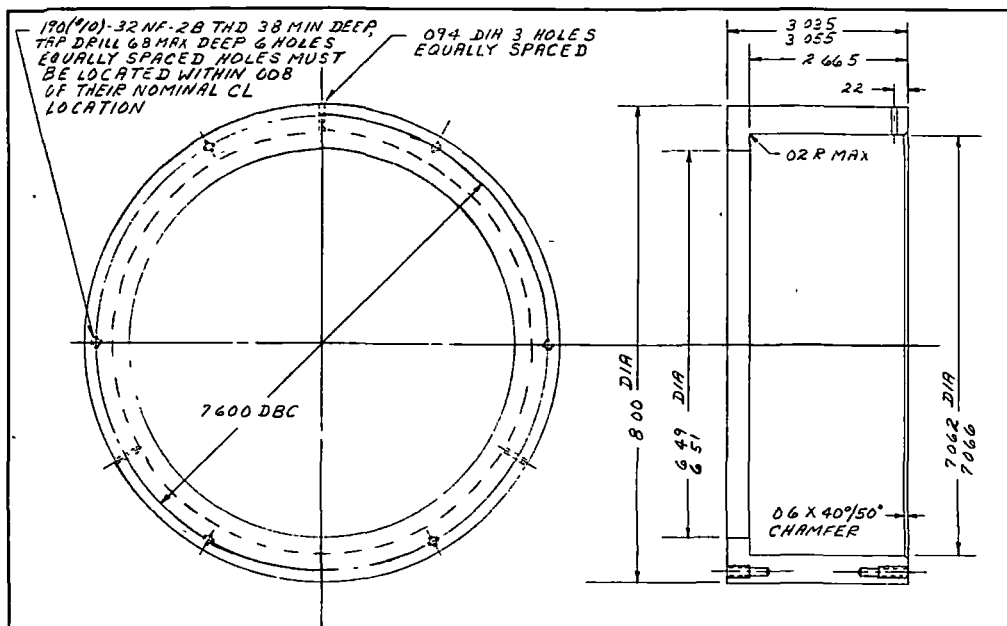
RING EDSK 326620
ITEM NO 3 REQ 1

RING EDSK 326620
ITEM NO 2 REQ 1

RING EDSK 326622
ITEM NO 1 REQ 1

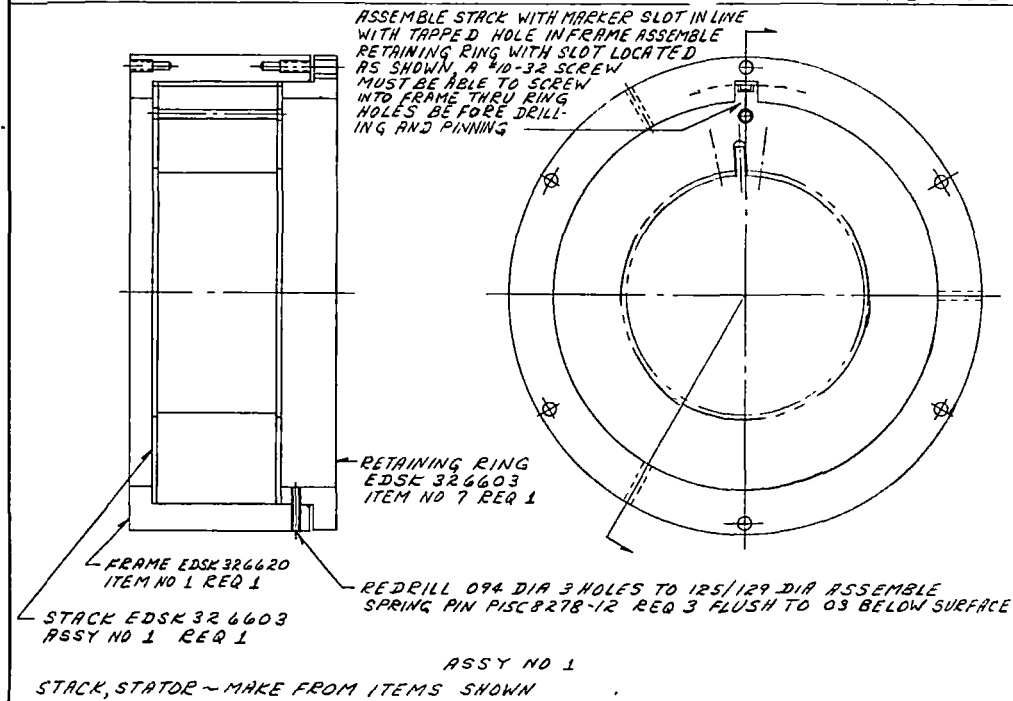
ASSY NO 2
END BELL ~ MAKE FROM ITEMS SHOWN

QTY	SYM	NOMENCLATURE OR DESCRIPTION	PART NO. OR IDENT NO.	SPECIFICATION	MATERIAL OR NOTE	INTERNAL INFORMATION	ITEM NO.
LIST OF MATERIAL OR PARTS LIST							
1	DTM	NGO	1/265				
1	DTM	CT	28565				
1	ENGR	W. Grant	1/265				
1	ENGR	W. Grant	1/265				
TITLE: DETAILS							
CODE IDENT NO. 83843 F EDSK 326622							
SCALE 1/1 WT 2MT SHEET							



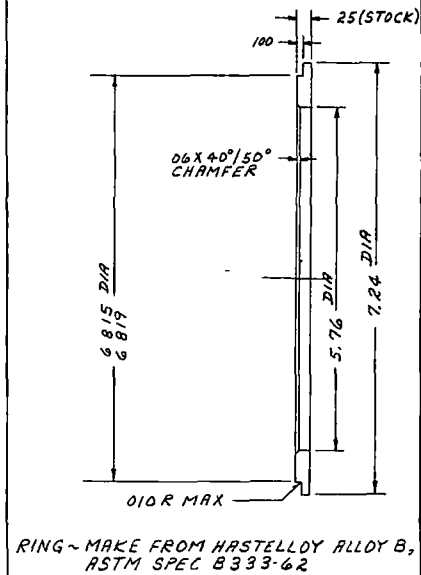
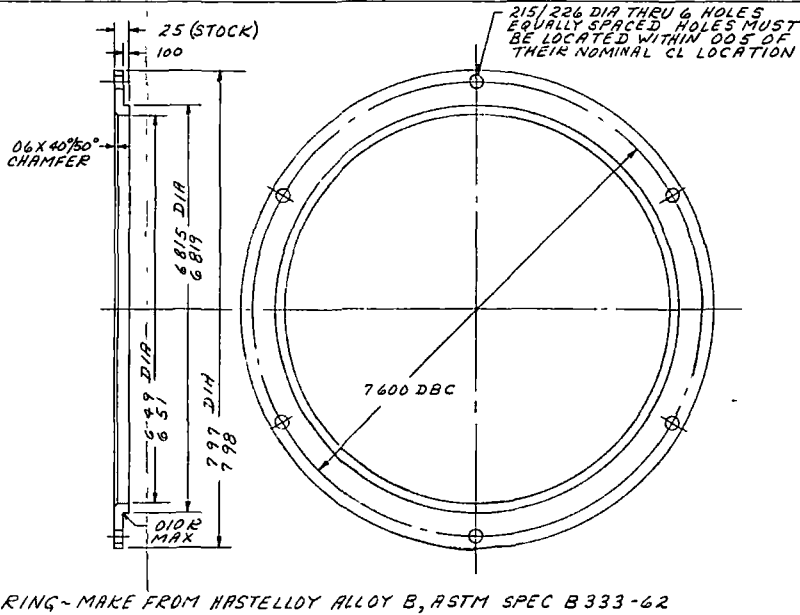
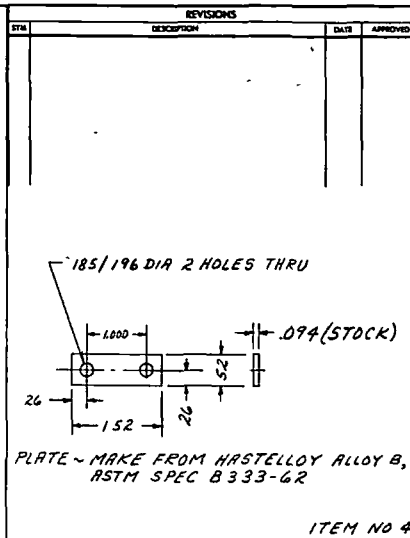
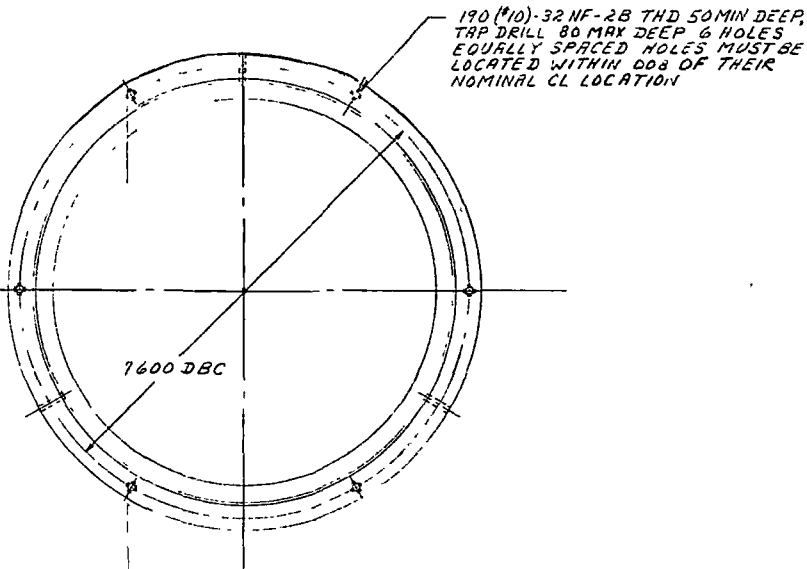
FRAME ~ MAKE FROM EDSK 326603 ITEM NO 4

ITEM NO 1



VIEW
LEGEND

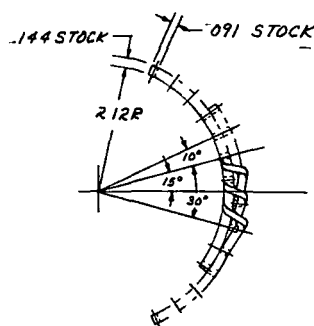
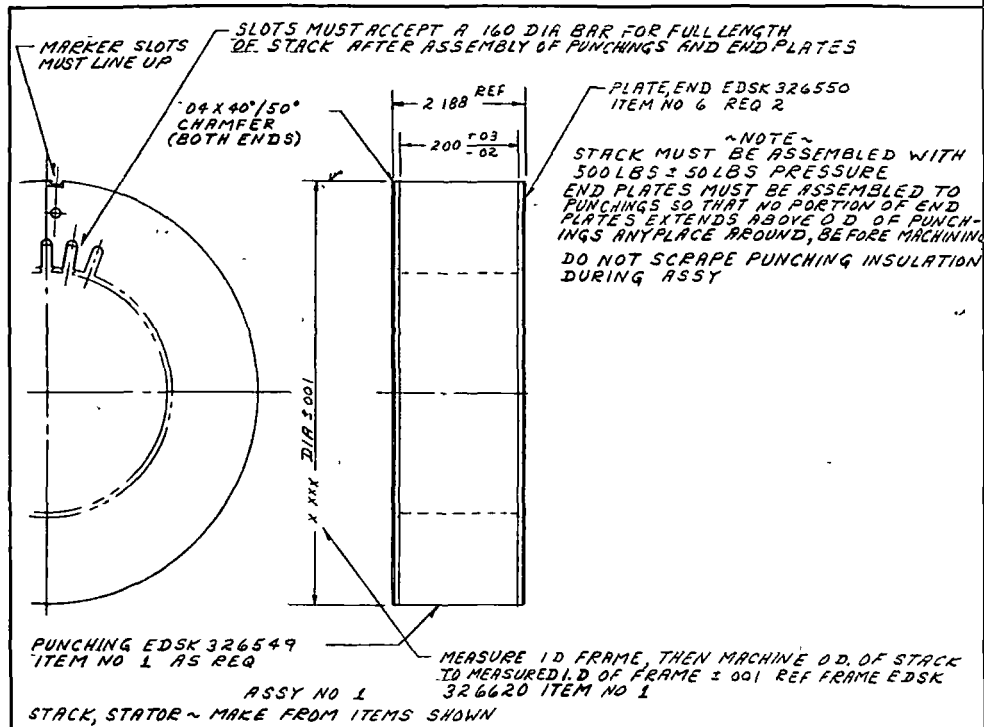
INTERNAL SPAC & DATA



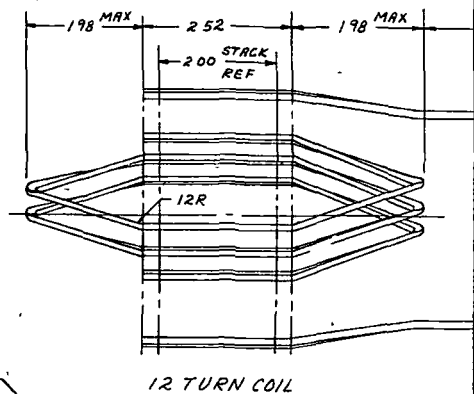
ITEM NO 2

ITEM NO 3

SYMBOL	DESCRIPTION	DATE	APPROVED
1	185/196 DIA 2 HOLES THRU		
2	215/226 DIA THRU 6 HOLES		
3	190 (#10)-32 NF-2B THD 50 MIN DEEP		
4	0.6 X 40°/50° CHAMFER		
5	7600 DBC		
6	0.10 R MAX		
7	0.074 (STOCK)		
8	0.008 OF THEIR NOMINAL CL LOCATION		
9	0.005 OF THEIR NOMINAL CL LOCATION		
10	0.008 OF THEIR NOMINAL CL LOCATION		
11	0.005 OF THEIR NOMINAL CL LOCATION		
12	0.008 OF THEIR NOMINAL CL LOCATION		
13	0.005 OF THEIR NOMINAL CL LOCATION		
14	0.008 OF THEIR NOMINAL CL LOCATION		
15	0.005 OF THEIR NOMINAL CL LOCATION		
16	0.008 OF THEIR NOMINAL CL LOCATION		
17	0.005 OF THEIR NOMINAL CL LOCATION		
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100	0.008 OF THEIR NOMINAL CL LOCATION		



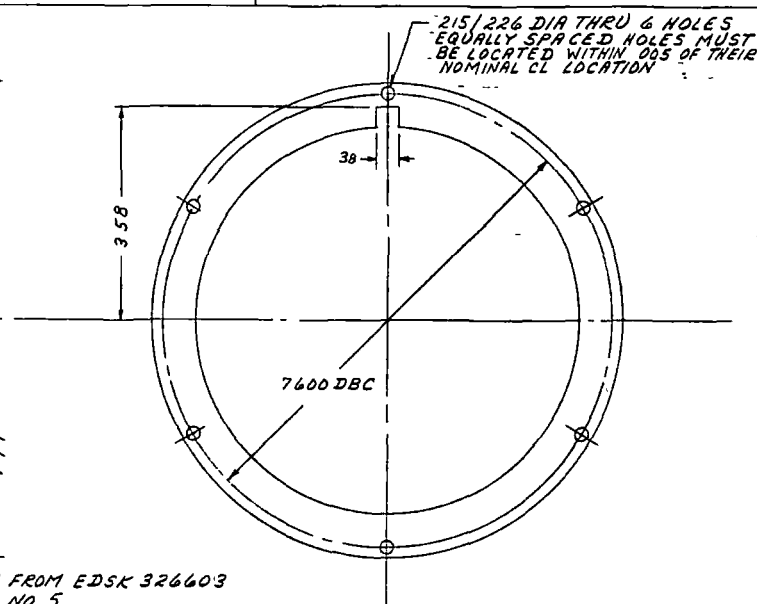
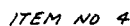
44 MAX DIA PIN
MIN DIA PIN
TO BE DEVELOPED



COIL, STATOR ~ ITEM NO 1 MAKE FROM .091 X .144 COPPER WIRE 12903-7

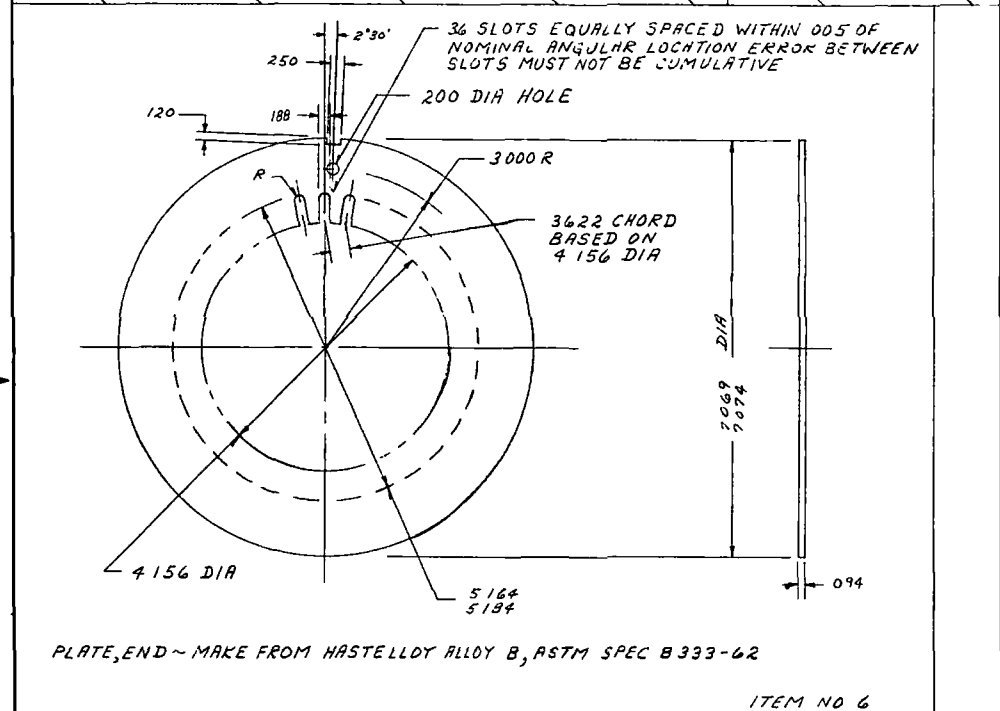
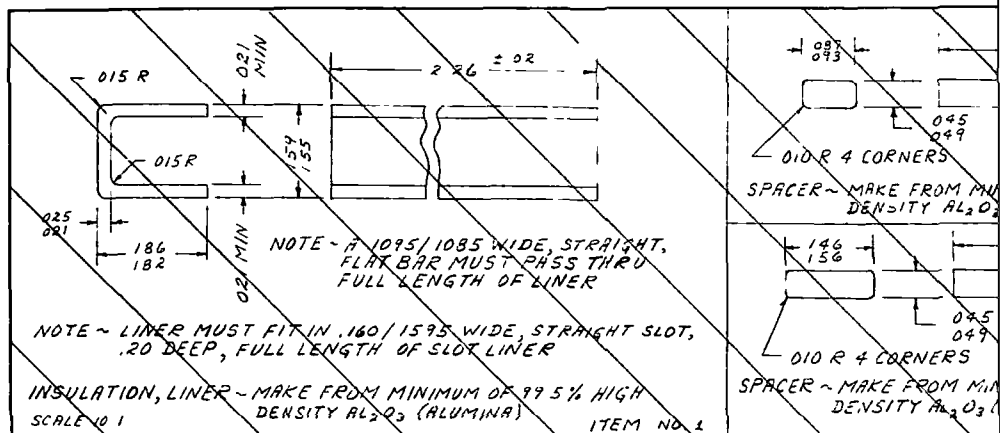
COIL, STATOR ~ ITEM NO 2 MAKE FROM .091 X .144 NICKEL-CLAD SILVER FURNISHED BY ENG

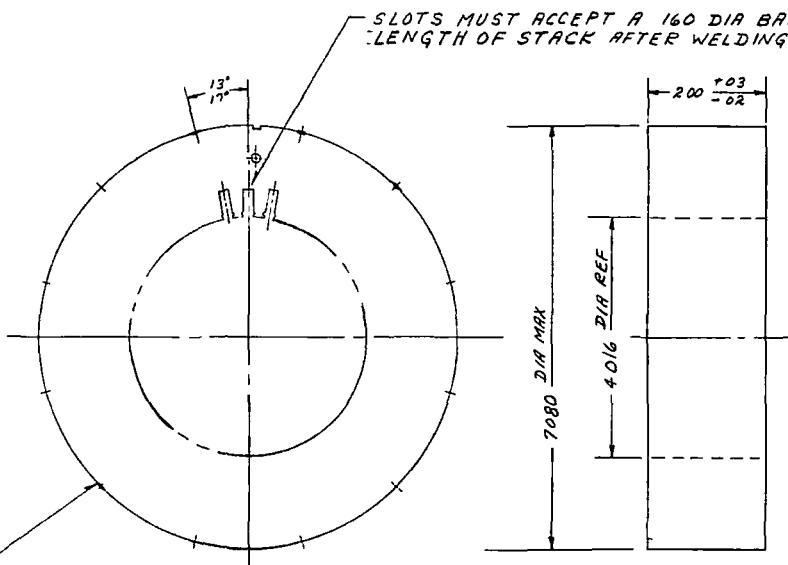
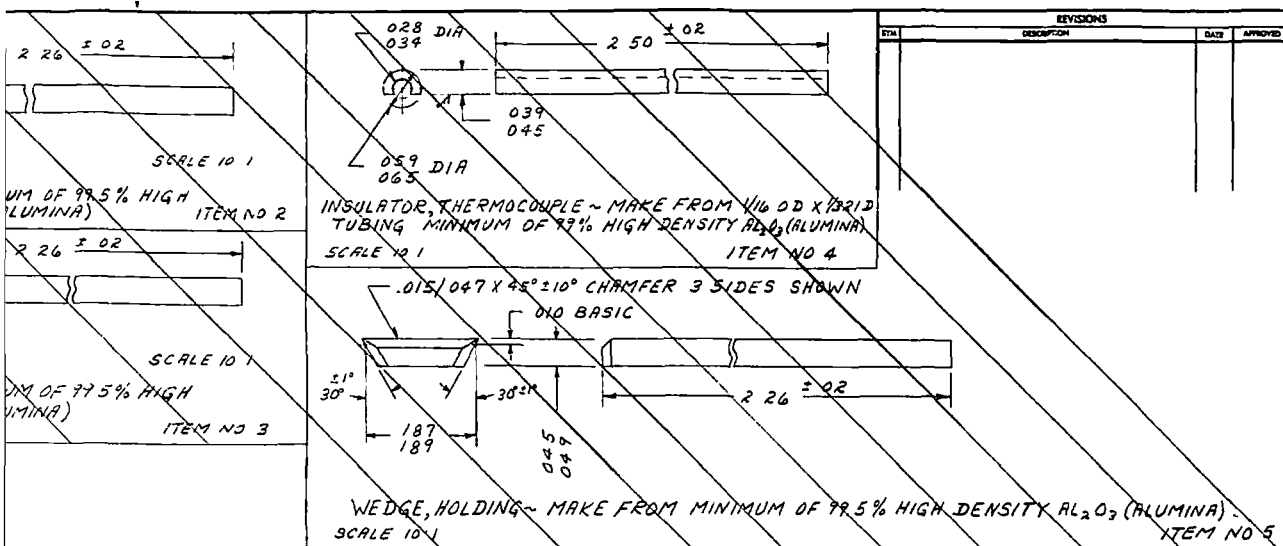
COIL, STATOR ~ ITEM NO.3 MAKE FROM .091 X .144 INCONEL-CLAD SILVER FURNISHED BY ENG



ITEM NO 7

<input type="checkbox"/> SPECIFICATION REFERENCE	<input type="checkbox"/> GOVERNMENT SPECIFICATION	<input type="checkbox"/> NAME OF PROCESS	<input type="checkbox"/> CITY STATE	<input type="checkbox"/> SYM	<input type="checkbox"/> NOMENCLATURE OR DESCRIPTION	<input type="checkbox"/> PART NO. OR IDENT. NO.	<input type="checkbox"/> SPECIFICATION	<input type="checkbox"/> MATERIAL OR NOTE	<input checked="" type="checkbox"/> INTERNAL INFORMATION	<input type="checkbox"/> ITEM NO.
LIST OF MATERIAL OR PARTS LIST										
TITLE WESTINGHOUSE ELECTRIC CORP LIMA, OHIO, U.S.A.			AEROSPACE ELECTRICAL DEPARTMENT WESTINGHOUSE ELECTRIC CORP LIMA, OHIO, U.S.A.							
DETAILS										
CODE IDENT. NO. 83843			SIZE F		DRAWING NO. EDSK 326603					
APP. FOR REVISIONS			SCALE 1 1" = 1"		METHOD 1 SHEET					



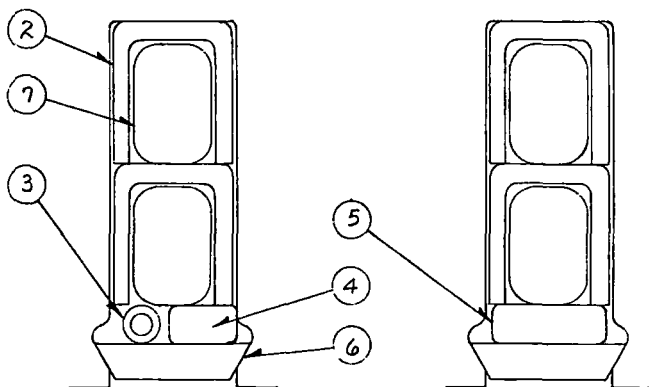
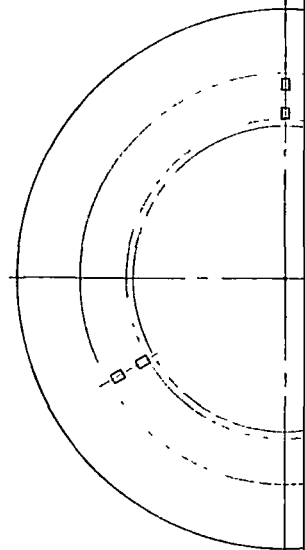
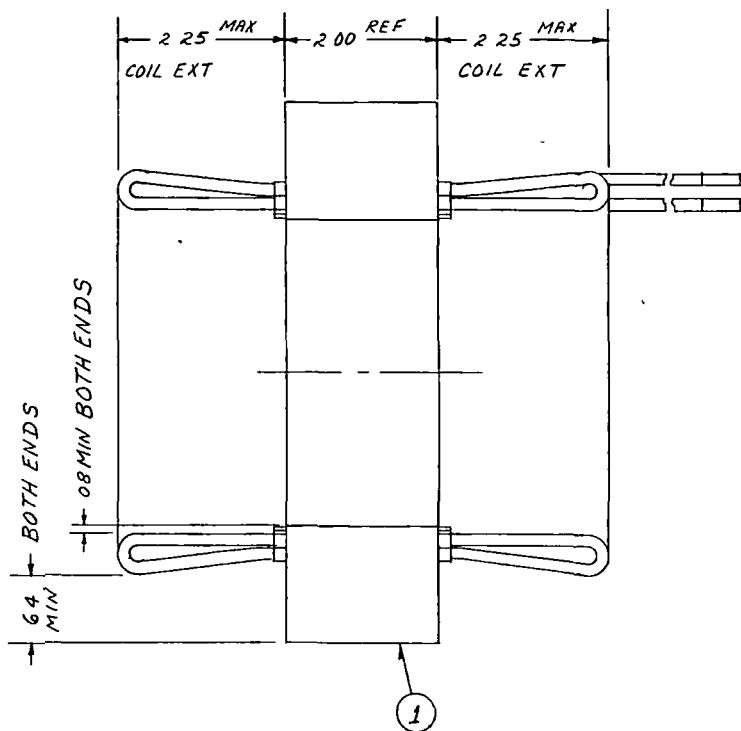


~NOTE~
STACK MUST BE ASSEMBLED WITH
500 LBS ± 50 LBS PRESSURE

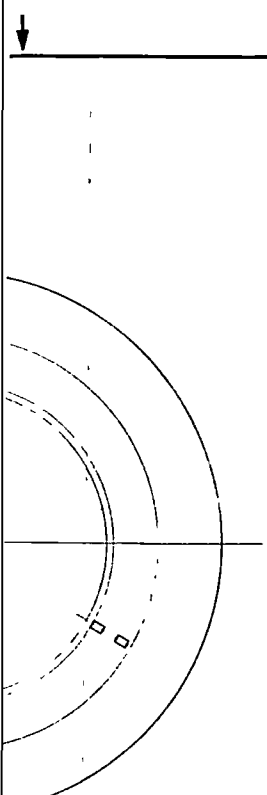
ARC WELD PER PS291466-1 12 BEADS EQUALLY SPACED WITHIN ± 25
FULL LENGTH OF STACK WELD BEADS MAY BE .03 MAX ABOVE O.D
OF STACK AND MAX WIDTH OF WELD .38

ASSY NO 1
STACK, STATOR ~ MAKE FROM PUNCHING EDSK 326549 ITEM NO 2 AS REQ

QTY	SYM	DESCRIPTION	PART NO.	SPECIFICATION	MATERIAL	INTERNAL	ITEM
REQD			OR IDENT NO.		OR NOTE	INFORMATION	NO.
LIST OF MATERIAL OR PARTS LIST							
		ITEM H90	12/16/66				
		ITEM E7	12/16/66				
		ENGINEER	12/16/66				
		ENGINEER	12/16/66				
		APPROVED					
		TEST					
		APPROVED					
		BY					
COORDINATOR				COORDINATOR	COORDINATOR	COORDINATOR	COORDINATOR
DRAWING NO.				83843	F	EDSK 326550	
SCALE				1/10	TWEIGHT		



8	LYP19037
9	INTERNAL INFO & DATA
10	ITEMS 3 & 4 SPECIFIED
11	ITEM 5 SPECIFIED
12	LYP19037 MEREILL
13	2-2-05
14	2-3-05
15	W.L. Grant
16	W.L. Grant
17	W.L. Grant
18	W.L. Grant
19	W.L. Grant
20	W.L. Grant
21	W.L. Grant
22	W.L. Grant
23	W.L. Grant
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97	W.L. Grant
98	W.L. Grant
99	W.L. Grant
100	W.L. Grant



REVISIONS			
SYM	DESCRIPTION	DATE	APPROVED

NOTE 1 - USE POTTING COMPOUND ITEM 8 AS REQUIRED TO FILL VOIDS IN SLOTS, CHANNELS AND AROUND WEDGES AND SPACERS EXTEND POTTING OUT APPROX HALF WAY TO THE COIL TURNS

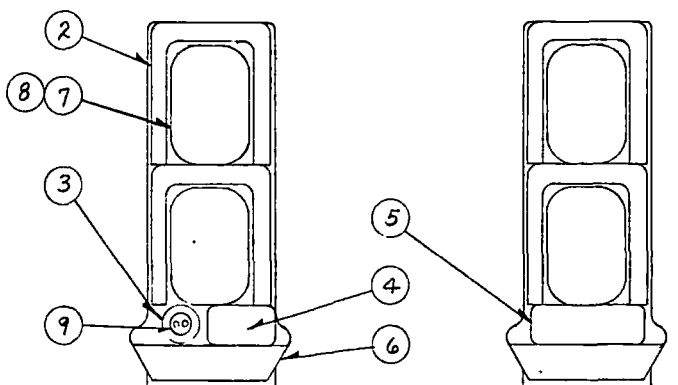
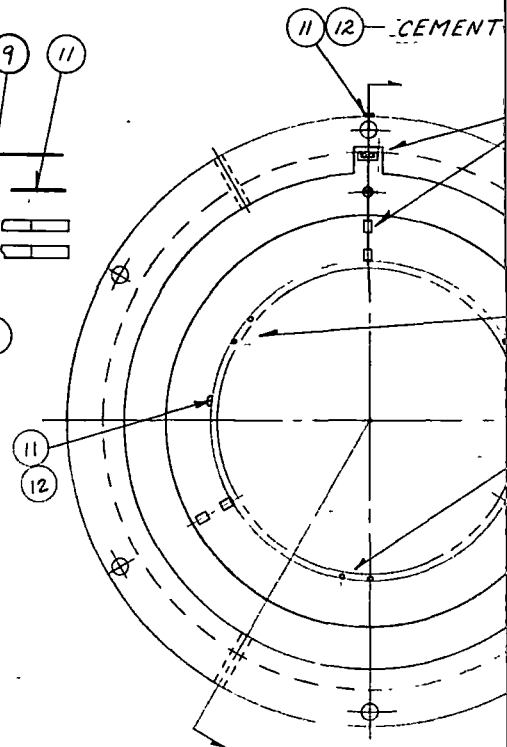
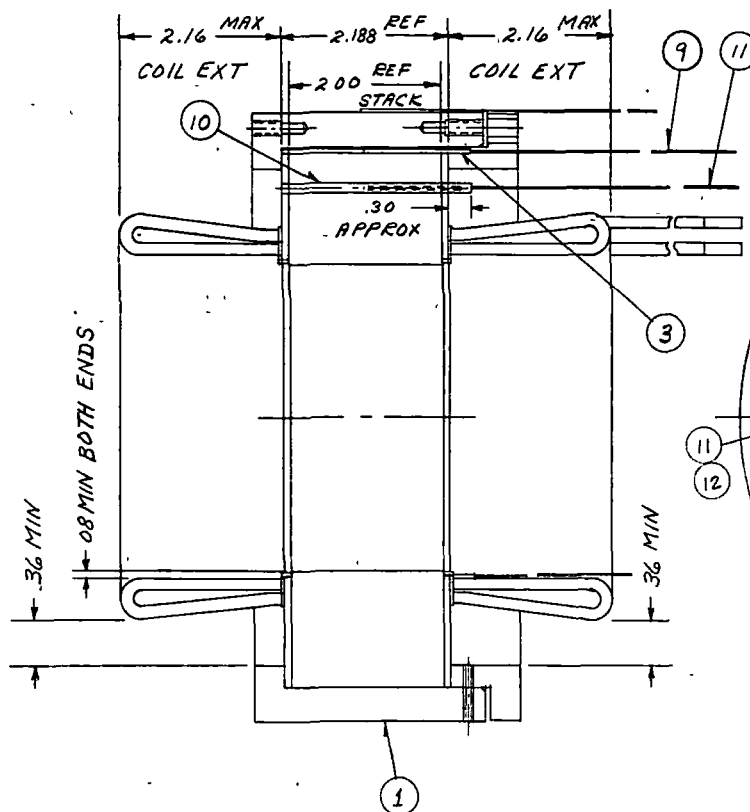
NOTE 2 - AFTER STATOR ASSEMBLY USE THE FOLLOWING SCHEDULE FOR POTTING COMPOUND BAKE-OUT: OVERNIGHT AT 125-150°F FOLLOWED BY 2 HOURS AT 450-500°F

AR B	POTTING COMPOUND	W-839 (WESTINGHOUSE)	8
3	COIL, STATOR	EDSK 326603 ITEM NO. 1	7
36	WEDGE	EDSK 326626 ITEM NO. 3	6
30	SPACER	EDSK 326626 ITEM NO. 2	5
6	SPACER	EDSK 326626 ITEM NO. 1	4
6	INSULATOR	EDSK 326680	3
72	INSULATION LINER	EDSK 326625	2
1	STACK, STATOR	EDSK 326550 ASSY NO 1	1

595405-1

SPECIFICATION REFERENCE GOVERNMENT SPECIFICATION NAME OF PROCESS

QTY REQD	SYM	NOMENCLATURE OR DESCRIPTION	PART NO. OR IDENT NO	SPECIFICATION	MATERIAL OR NOTE	INTERNAL INFORMATION	ITEM NO
LIST OF MATERIAL OR PARTS LIST							
TOLERANCES UNLESS OTHERWISE SPECIFIED		AEROSPACE ELECTRICAL DEPARTMENT					
DIMENSIONS ARE IN INCHES UNLESS OTHERWISE SPECIFIED		WESTINGHOUSE ELECTRIC CORP					
FRACTIONS DECIMALS ANGLES		LIMA, OHIO, U.S.A.					
FRACTIONS DECIMALS ANGLES		TITLE					
PROCESS TO BE APPLIED AS INDICATED ON DRAWING		STACK, STATOR WOUND					
GOVT SPEC		CODE IDENT NO. SIZE DRAWING NO.					
SPEC		83843 D EDSK 326602					
STANDARD		SCALE 1/10 1 WEIGHT SHEET					
BY WESTINGHOUSE		APPROVED FOR					
BY SUPPLIER		BY					



~NOTES~

1-ALL THERMOCOUPLE TIPS TO BE APPROX IN CENTER OF STACK IN STACK

2-USE POTTING COMPOUND ITEM 13 AS REQ CHANNELS AND AROUND WEDGES AND SPAC APPROXIMATELY HALF WAY TO THE COIL

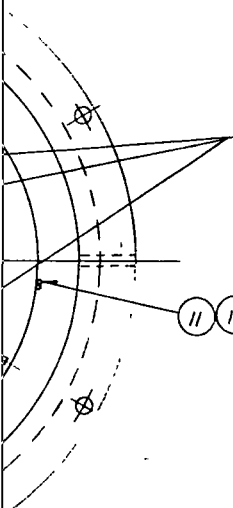
3-AFTER STATOR ASSEMBLY USE THE FO COMPOUND BAKE-OUT AND SETTING OF OVERNIGHT AT 125-150°F, 2 HOURS AT 800-900°F, 1/2 HOUR AT 1200-1250°F

S. LYP 19037
INTERNAL INFO & DATA
CHANGED ITEM 11
FROM 6 TO 8 ADDED
ITEM 13 & NOTES 2
& 3 ADDED SPEC REF
W.L. Grant 5-3-65

TERMOCOUPLES TO O.D. WITH PYROCERAM

REVISIONS			
SYM	DESCRIPTION	DATE	APPROVED

BRING LEADS OUT IN APPROX
POSITION SHOWN IN RELATION
TO SLOT SHOWN



APPROX LOCATION OF THERMOCOUPLES IN SLOTS

11 12 CEMENT THERMOCOUPLES TO COILS WITH PYROCERAM

ASSEMBLED
WHEN USED

ED TO FILL VOIDS IN SLOTS,
25 EXTEND POTTING OUT
ND TURNS.

DRAWING SCHEDULE FOR POTTING
INDUCTOR ANADUR INSULATION.
50-500°F, 2 HOURS AT

AR	AR	B	POTTING COMPOUND	W-839 (WESTINGHOUSE)	13
AR	AR	B	PYROCERAM		12
8	8		THERMOCOUPLE	EDSK 326796	11
1	1		INSULATOR	EDSK 326794	10
8	8		THERMOCOUPLE	EDSK 326795	9
3	-		COIL, STATOR	EDSK 326603 ITEM NO. 3	8
-	3		COIL, STATOR	EDSK 326603 ITEM NO. 2	7
36	36		WEDGE	EDSK 326626 ITEM NO. 3	6
30	30		SPACER	EDSK 326626 ITEM NO. 2	5
6	6		SPACER	EDSK 326626 ITEM NO. 1	4
8	8		INSULATOR	EDSK 326680	3
72	72		INSULATION, LINER	EDSK 326625	2
1	1		STACK, STATOR	EDSK 326620 ASSY NO 1	1

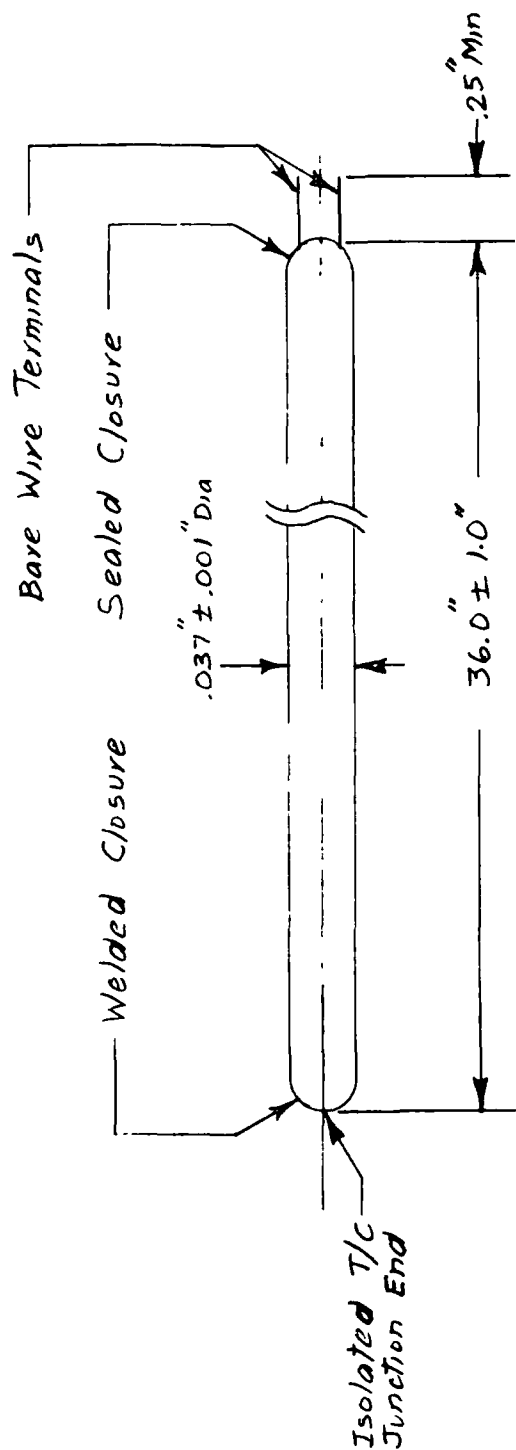
DWG EDSK 326618

595405-1

SPECIFICATION REFERENCE	GOVERNMENT SPECIFICATION	NAME OF PROCESS
----------------------------	-----------------------------	--------------------

QTY REQD PK-2	QTY REC'D PK-1	SYM	NOMENCLATURE OR DESCRIPTION	PART NO OR IDENT NO	SPECIFICATION	MATERIAL OR NOTE	INTERNAL INFORMATION	ITEM NO		
LIST OF MATERIAL OR PARTS LIST										
TOLERANCE AND UNITS PER LA 225 UNLESS OTHERWISE SPECIFIED DIMENSIONS ARE IN INCHES, NOT SCALE DWG FRACTIONS ON DECIMALS ANGLES IN DEGREES PROCESS TO BE APPLIED AS INDICATED ON DRAWING				DFTM HGO 1-8-65 DFTM LT 1-25-65 ENGR W L Grant 1-25-65 MFR TEST PROJ ENGR APPD APPD FOR BY					AEROSPACE ELECTRICAL DEPARTMENT WESTINGHOUSE ELECTRIC CORP LIMA, OHIO, U.S.A. TITLE STACK, STATOR WOUND CODE IDENT NO 83843 SIZE D DRAWING NO EDSK 326618 SCALE 1/10 WEIGHT SHEET	

TOOL REF
NOT TOOL
PROD TOOL
TOOLS SCRAPED
SPARES CANNOT BE FURN
NEET ASSP



Note: Thermocouples are to be kept straight at all times

SINGLE DIAMETER SHEATHED THERMOCOUPLE

Reference (W) D-Spec # 709747

EDSK 326796

Wg 2-18-65

.250 DIA			
PART NO	A	CATALOG NO	GOVT PART NO
PISC8278-1	7/8	79-048-250-0875	MS171652
-2	3/4	79-048-250-075	MS171650
-3	2	79-048-250-2000	MS171662
-4	3/4	79-048-250-3250	MS171667
-5	13/8	79-048-250-1375	MS171657
-6	1 1/2	79-048-250-1500	MS171658

.125 DIA			
PART NO	A	CATALOG NO	GOVT PART NO
PISC8278-9	3/8	79-028-125-0375	MS171524
-10	7/16	52-028-125-0437	
-11	7/16	79-028-125-0437	MS171525
-12	3/4	79-028-125-0750	MS171530
-13	1	79-028-125-1000	MS171534
-14	13/8	79-028-125-1375	MS171537
-15	1 1/4	79-028-125-0250	MS171522
-16	13/4	79-028-125-1750	MS171540
-17	5/8	52-028-125-0625	
-60	9/16	79-028-125-0562	MS171527

.156 DIA			
PART NO	A	CATALOG NO	GOVT PART NO
PISC8278-35	1/2	79-032-156-0500	MS171556
-36	7/16	52-032-156-0437	
-37	1/2	79-032-156-1500	MS171568




.094 DIA			
PART NO	A	CATALOG NO	GOVT PART NO
PISC8278-45	1/2	79-022-094-0500	MS171496
-46	3/8	79-022-094-0375	MS171494
-47	5/16	79-022-094-03125	MS171493
-48	1/4	79-022-094-0250	MS171492

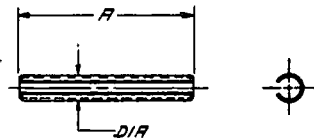
.062 DIA			
PART NO	A	CATALOG NO	GOVT PART NO
PISC8278-18	3/8	79-012-062-2375	MS171434
-19	5/16	79-012-062-0312	MS171433
-20	1/4	79-012-062-0250	MS171432
-21	3/16	79-012-062-0188	MS171431
-22	7/16	52-012-062-0437	
-23	1/2	79-012-062-0500	MS171436
-24	7/8	52-012-062-0125	

.188 DIA			
PART NO	A	CATALOG NO	GOVT PART NO
PISC8278-26	1 1/4	79-040-187-0687	MS171589
-27	3/4	79-040-187-0813	MS171591
-28	3/8	79-040-187-0375	MS171584
-29	1/8	79-040-187-1125	MS171595
-30	1	79-040-187-1000	MS171594
-31	13/8	79-040-187-1375	MS171597
-32	7/16	79-040-187-0438	MS171585

.312 DIA			
PART NO	A	CATALOG NO	GOVT PART NO
PISC8278-40	1	52-062-312-1000	

.219 DIA			
PART NO	A	CATALOG NO	GOVT PART NO
PISC8278-51	1/2	79-048-219-0500	MS171616

- NOTE -
 ORDERING INFORMATION FOR  INTERNAL
 COLUMN ORDER  PART NO.
 FOR GOVT COLUMN ORDER GOVT PART NO. IF
 AVAILABLE OTHERWISE SPECIFY  PART
 NO. AND NOTE "ENDOR ITEM SEE SPECIFICATION
 CONTROL DRAWING."



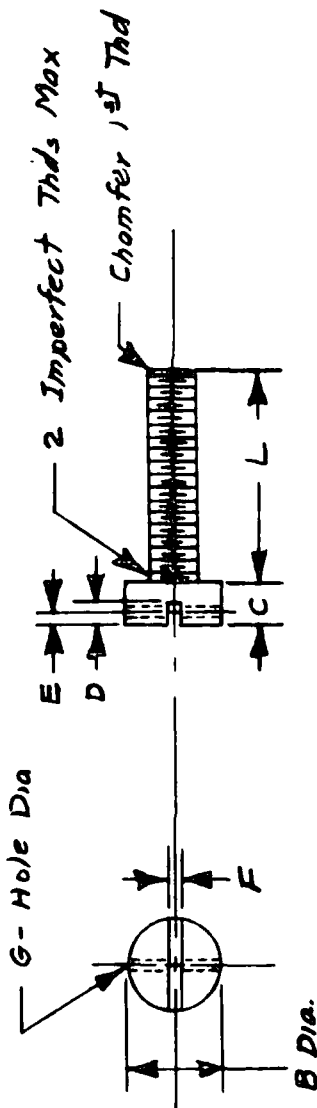
5. # PD18075	INTERNAL INFO & DATA
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100. PER REV AB	99. PER REV AB

REQUIREMENTS

- QUALIFICATION TESTS
NOT APPLICABLE

PIN-ELASTIC STOP NUT CORPORATION UNION, NEW JERSEY CODE IDENT 72962			BOUGHT OUTSIDE				
QTY REQD	SYM	NOMENCLATURE OR DESCRIPTION	PART NO OR IDENT NO	SPECIFICATION	MATERIAL OR NOTE	INTERNAL INFORMATION	ITEM NO
LIST OF MATERIAL OR PARTS LIST							
TOLERANCE AND SPECS REF P 231 UNLESS OTHERWISE SPECIFIED DIMENSIONS ARE IN INCHES, DO NOT SCALE DIMS TOLERANCES ON FRACTIONS ANGLES .2514 .005 .12 .500 .005 .12 PROCESS TO BE APPLIED AS INDICATED ON DRAWING			DFTM <i>EB</i> <i>11/13/61</i> DFTM <i>J. Calhoun</i> <i>11/15/61</i> ENGR <i>R. Milburn</i> <i>11/15/61</i> ENGR ENGR MGR TEST		AEROSPACE ELECTRICAL DEPARTMENT WESTINGHOUSE ELECTRIC CORP LIMA, OHIO U.S.A.		
COINT SPEC (NONE)			TITLE PIN, SPRING STANDARD APPARATUS				
STRINGS @ WESTINGHOUSE BUS BULK MATERIAL @ SPEC/ISSUES COINT DIMS @ IN PROCESS PART @ SPEC/ISSUES PART @ SPEC/ISSUES PART			CODE IDENT NO 83843		SIZE D		
APPD FOR BY			DRAWING NO P15C8278		SCALE <i>NONE</i> WEIGHT SHEET		

WESTINGHOUSE ELECTRIC CORPORATION



Material - Hastelloy Alloy B Rod

Dash No.	Thread	B Dia	C	D	E	F	G Dia	L	Stock
1	#8 (164)-32 NC-2A	.270	.156	.078	.050	.062	.040	.500	3/8 Dia
2	" " "	.270	.156	.078	.050	.062	.040	.750	3/8 Dia
3	#10 (190)-32 NF-2A	.313	.156	.078	.050	.062	.040	.500	7/16 Dia
4	" " "	.313	.156	.078	.050	.062	.040	1.00	7/16 Dia
5	1/4-28 UNF-2A	.414	.207	.103	.078	.062	.040	.500	1/2 Dia

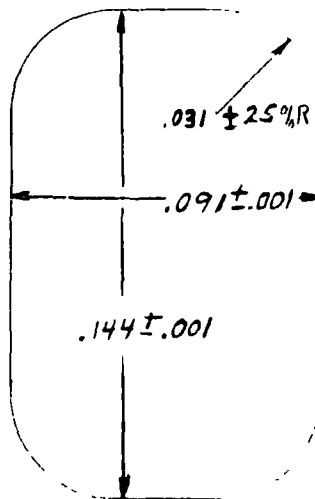
SCREW - FLAT FILLISTER HEAD

W. L. Grant 4-21-65 EDSK 327502

EDSK 327502

WESTINGHOUSE ELECTRIC CORPORATION

Conductor
20% Nickel-Clad Silver



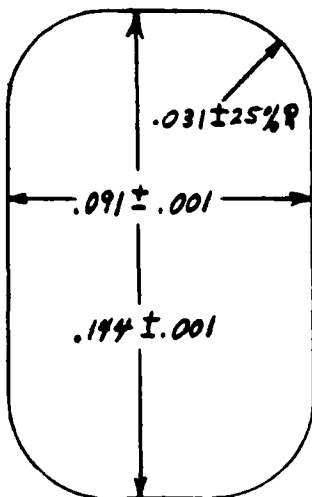
Dimensions in inches
Scale 20:1

EDSK 326531

H.E. Kellepp 12-17-64

WESTINGHOUSE ELECTRIC CORPORATION

Conductor
28 % Inconel-Clad Silver



Dimensions in inches
Scale 20:1

EDSK 326532 A

REV. A - 12-29-64
28% was 20% w/

HE/Keneipp 12-17-64

APPENDIX B

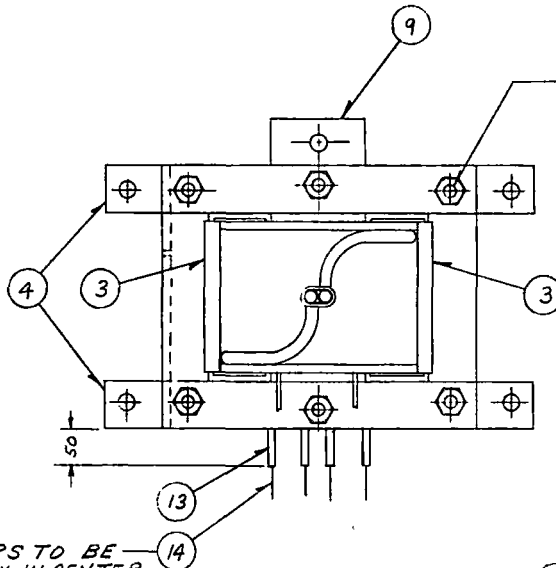
TRANSFORMER MATERIALS SUMMARY

APPENDIX B

TRANSFORMER ASSEMBLY

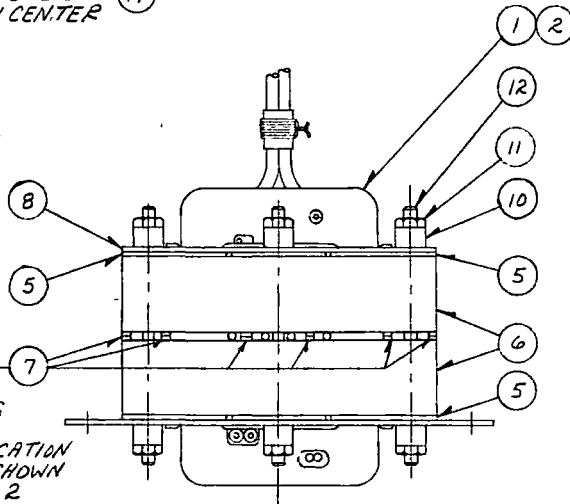
DRAWING NO.	TITLE	MATERIAL
EDSK 326761	Transformer Assembly	-
EDSK 326760	Winding, Transformer	-
EDSK 326762	Details	See Drawing
EDSK 326756	Punching, Transformer	Hiperco 27 Alloy (27 Co - Fe)
EDSK 326626	Details, Slot	99 Al ₂ O ₃
EDSK 326686	Insulation, Tube	99 Al ₂ O ₃
EDSK 326757	Channel	99.5 Al ₂ O ₃
EDSK 326758	Spool	99.5 Al ₂ O ₃
EDSK 326759	End Plates	99.5 Al ₂ O ₃
EDSK 326786	Spacer	99 Al ₂ O ₃
EDSK 326796	Thermocouple	Inconel Cladding, Platinel II (wire system, Al ₂ O ₃ Insulation)
EDSK 327503	Nut	Hatelloy Alloy B
EDSK 326533	Conductor	Nickel Clad Silver (20% Clad area) ⁽¹⁾
EDSK 326534A	Conductor	Inconel Clad Silver (28% Clad Area) ⁽¹⁾
EDSK 326535	Conductor	Nickel Clad Silver (20% Clad area) ⁽¹⁾
EDSK 326536A	Conductor	Inconel Clad Silver (28% Clad area) ⁽¹⁾

(1) - All conductors insulated with Anaconda's Anadur, a refractory-oxide-filled glass insulation.



NOTE ~ AFTER FINAL ASSEMBLY AND TEST, HELIARC SPOT WELD STUDS - 12 PLACES

THERMOCOUPLE TIPS TO BE LOCATED APPROX IN CENTER OF STACK



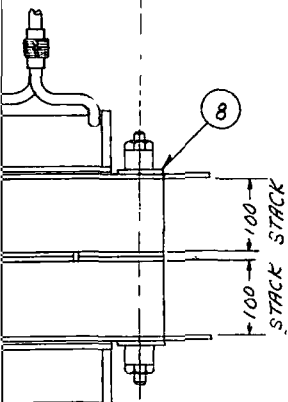
CUT ONE SPACER INTO TWO AND USE AS REQ TO HOLD STACKS APART 1 AND 1/2 SPACERS AT EACH LOCATION LOCATE APPROX AS SHOWN 6 PLACES ACROSS 2 LAYERS OF SPACERS

NOTE ~ USE DUCO CEMENT TO HOLD SPACERS (ITEM NO 7) AND INSULATORS (ITEM NO 13) IN PLACE AT CENTER OF STACK DURING ASSY AFTER CEMENT SETS, POT AS REQ BETWEEN SPACERS AND INSULATORS

NOTE ~ DO INSUL

S. LYP 190.2.3	INTERNAL INFO & DATA
ITEM 7 SPEC EDSK 3	
26682 ITEM 2 REQD 9	
ITEM 11 SPEC 2 REQD 4	
ITEM 13 SPEC 2 REQD 4	
26680 ITEM 14 SPEC	
EDSK 326795 ITEM 15	
"W" TIDED TO 839, PIC.	
TUBE CHANGED TO SHOW	
2 LAYERS OF SPACERS	
(ITEM 7), 5-LYP190-2.3	
ME-FEILL 2-4-65	
SPACER 2-3-65	
1 W.L. Grant	
ADDED WELDING NOTE	
ADDED HOLE TO IT 9	
CHANGED ITEM 11 TO	
EDSK CALL-OUT	
2 W.L. Grant 5-1-65	

ACCEPTANCE
PUTS TO



T SCRAPE PUNCHING
ACTION DURING ASSY

REVISIONS			
SYM	DESCRIPTION	DATE	APPROVED

RR	RR	B	POTTING COMPOUND WESTINGHOUSE W-839	15
4	4		THERMOCOUPLE EDSK 326796	14
4	4		INSULATOR EDSK 326686-3	13
6	6		STUD EDSK 326762 ITEM NO. 7	12
12	12		NUT EDSK 327503-1	11
12	12		SPACER EDSK 326786	10
2	2		STRAP, GROUND EDSK 326762 ITEM NO 4	9
2	2		PLATE EDSK 326762 ITEM NO. 2	8
18	18		SPACER EDSK 326626 ITEM NO 1	7
RR	RR		PUNCHING EDSK 326756	6
4	4		PLATE, SUPPORT EDSK 326762 ITEM NO 3	5
2	2		PLATE, MOUNTING EDSK 326762 ITEM NO 1	4
2	2		CHANNEL EDSK 326757	3
1	-		WINDING EDSK 326760 PN-2	2
-	1		WINDING EDSK 326760 PN-1	1

EDSK 326761

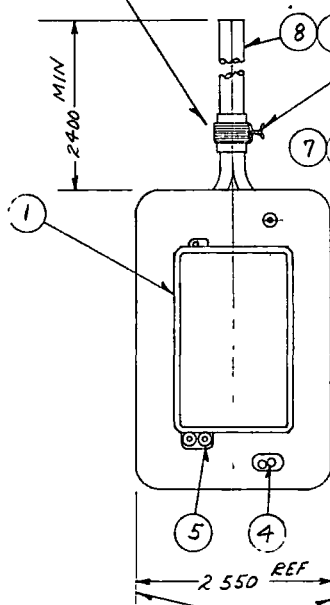
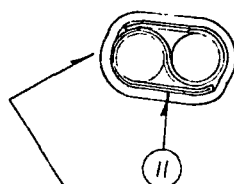
① SPECIFICATION REFERENCE	GOVERNMENT SPECIFICATION	NAME OF PROCESS
------------------------------	-----------------------------	--------------------

QTY REQD PN-2	QTY REQD PN 1	SYM	NOMENCLATURE OR DESCRIPTION	PART NO OR IDENT NO	SPECIFICATION	MATERIAL OR NOTE	② INTERNAL INFORMATION	ITEM NO
LIST OF MATERIAL OR PARTS LIST								
TOLERANCES AND SPECS UNLESS OTHERWISE SPECIFIED DIMENSIONS ARE IN INCHES, DO NOT SCALE DWG FRACTIONS DECIMALS ANGLES PROCESS TO BE APPLIED AS INDICATED ON DRAWING				DFTM HGO 11/9/65 DFTM LT 1-25-65 ENGR HLG 1/25/65 ENGR W L Grant 1-25-65 MFR TEST PROJ ENGR APPD APPD FOR BY				
GOVT SPEC SPEC SYMBOLES W=WESTINGHOUSE SC=SPEC/STANDARD CONY DWG Y=FORWARD FILE U=VERY IMPORTANT				AEROSPACE ELECTRICAL DEPARTMENT WESTINGHOUSE ELECTRIC CORP LIMA, OHIO U.S.A. TITLE TRANSFORMER CODE IDENT NO. 83843 SIZE D DRAWING NO EDSK 326761 SCALE 1/1 WEIGHT SHEET				

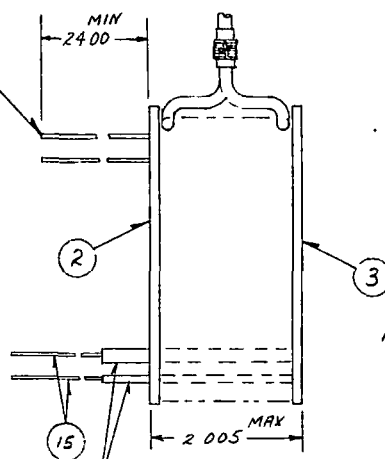
100% REF
100% P.N.
NOT TOOLED
PROD TOOLED
DOES SCREWED
SIZES CANNOT BE FURN
LAST ASSY

ENLARGED VIEW SHOWING 2 LAYERS OF INSULATION ITEM 11 WRAPPED AROUND LEADS CUT INSULATION TO LENGTH AT ASSEMBLY

0.044 DIA MAX WIRE WINDING, INSULATION TO



8 9 10 TIE LEADS TOGETHER WITH ITEM 10 TWIST WIRES AFTER BANDING



WINDING IN 4 LAYERS 37 TURNS PER L OF #20 032 BARE .044 MAX LAYER OF INSULATION ITEM WIRE 4 LAYERS OF INSULATION WINDINGS SECONDARY WINDING OF #7 .144 BARE 156 MAX IN USE SMALL PIECES OF TAPE TO HOLD INSULATION IN PLACE

NOTE ~ CEMENT PLATES ITEM SPOOL ITEM NO 1 WITH

WINDINGS MUST NOT EXTEND ABOVE SIDES OF PLATES

THERMOCOUPLE INSULATORS TO EXTEND THRU W AND BUTT UP AGAINST END PLATE, THERMOCOUPLE ASSEMBLED APPROX IN CENTER OF WINDING

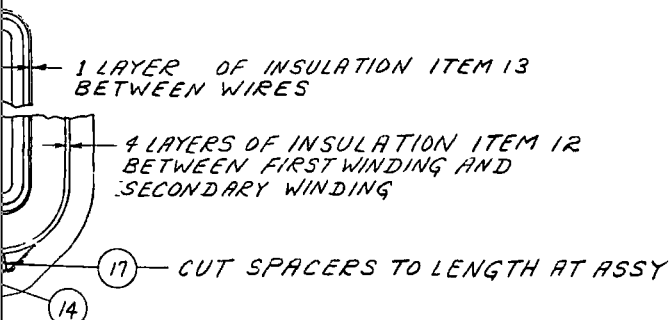
NOTE ~ PRIOR TO POTTING (ITEM 16) INSULATORS AND SPACERS, BAKE OUT CONDUCTOR INSULATION AT $1225 \pm 25^\circ\text{F}$ FOR $\frac{1}{2}$ HOUR

NOTE ~ POT SPACERS, INSULATORS AND OUTSIDE SURFACE OF SECONDARY WINDING

3	LYP 19038
4	INTERNAL INFO & DATA
5	W ADDED TO 839, ITEM 16
6	S LYP 19038 MEEBELL
7	2.2-65
8	CALLER 2-3-65
9	WL Grant 2-3-65
10	1 LAYER WAS 3
11	LAYERS-2 PLACES
12	ADDED POTTING
13	NOTE AND BAKE
14	OUT NOTE ADDED
15	POTTING SPEC REF
16	2 WL Grant 5-3-65

GO ACROSS SPOOL BEFORE START OF WIRE WITH ONE LAYER OF ITEM 14, CUT LENGTH AT ASSEMBLY

OVERLAP INSULATION ITEM 13 AT TOP ON EACH ROW OF WIRE, CUT INSULATION TO LENGTH AT ASSEMBLY



FORMATION
WIRE, 1 LAYER 26 TURNS
INSULATED WIRE WITH 1
3 BETWEEN LAYERS OF
ON ITEM 12 BETWEEN
NG 1 LAYER 10 TURNS
INSULATED WIRE

ITEM NO 18 AS REQ
DURING WINDING

2 AND 3 TO
CO CEMENT

WINDING
TIPS TO BE

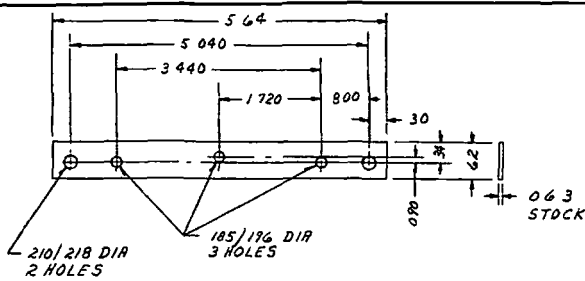
AR	AR	B	TAPE, ADHESIVE	0025X.500 9948-3	18
4	4		SPACER	EDSK 326626 ITEM NO 1	17
AR	AR		POTTING COMPOUND WESTINGHOUSE W839		16
4	4		THERMOCOUPLE	EDSK 326796	15
2	2		INSULATION	EDSK 326762 ITEM NO 8	14
4	4		INSULATION	EDSK 326762 ITEM NO. 5	13
1	1		INSULATION	EDSK 326762 ITEM NO. 3	12
1	1		INSULATION	EDSK 326762 ITEM NO 9	11
AR	AR	B	WIRE, SST	032 DIA 7860-2	10
AR	-	B	WIRE #7 (.144 DIA) INCONEL - CLAD SILVER FURNISHED BY ENGINEER		9
-	AR	B	WIRE #7 (.144 DIA) NICKEL - CLAD SILVER FURNISHED BY ENGINEER		8
AR	-	B	WIRE #20 (.032 DIA) INCONEL - CLAD SILVER FURNISHED BY ENGINEER		7
-	AR	B	WIRE #20 (.032 DIA) NICKEL - CLAD SILVER FURNISHED BY ENGINEER		6
2	2		INSULATOR	EDSK 326686 PN-2	5
2	2		INSULATOR	EDSK 326686 PN-3	4
1	1		END PLATE	EDSK 326759 ITEM NO 1	3
1	1		END PLATE	EDSK 326759 ITEM NO 2	2
1	1		SPOOL	EDSK 326758	1

595405-1

SPECIFICATION REFERENCE	GOVERNMENT SPECIFICATION	NAME OF PROCESS
----------------------------	-----------------------------	--------------------

QTY REQD PV 2	QTY READ PN-1	SYM	NOMENCLATURE OR DESCRIPTION	PART NO OR IDENT NO	SPECIFICATION	MATERIAL OR NOTE	INTERNAL INFORMATION	ITEM NO.
LIST OF MATERIAL OR PARTS LIST								
TOLERANCE AND SPECS UNLESS OTHERWISE SPECIFIED DIMENSIONS ARE IN INCHES DO NOT SCALE DWG FRACTIONS = 1/16 = 1/32 = 1/64 = 1/8 = 1/4 = 1/2 = 3/4 = 1 = 1 1/2 = 2 = 3 = 4 = 6 = 8 = 10 = 12 = 14 = 16 = 18 = 20 = 22 = 24 = 26 = 28 = 30 = 32 = 34 = 36 = 38 = 40 = 42 = 44 = 46 = 48 = 50 = 52 = 54 = 56 = 58 = 60 = 62 = 64 = 66 = 68 = 70 = 72 = 74 = 76 = 78 = 80 = 82 = 84 = 86 = 88 = 90 = 92 = 94 = 96 = 98 = 100			DFTM H40	1/18/65	AEROSPACE ELECTRICAL DEPARTMENT WESTINGHOUSE ELECTRIC CORP LMA OHIO U.S.A.			
PROCESS TO BE APPLIED AS INDICATED ON DRAWING			DFTM 27	12565	TITLE WINDING, TRANSFORMER			
COPY SPEC			ENGR W L Grant	12565				
SPEC			MFR					
SYMBOLS = WESTINGHOUSE = SPEC SOURCE CONT DWG = EXTENSION PART = VERY IMPORTANT			TEST					
			PROJ ENGR APPD					
			APPD FOR					
			BY					
			CODE IDENT NO.	83843	SIZE	D	DRAWING NO	EDSK 326760
			SCALE	1/2	WEIGHT		SHEET	

EDSK 326760



PLATE, MOUNTING ~ MAKE FROM HASTELLOY ALLOY B,
ASTM SPEC B 333-62

ITEM NO 1

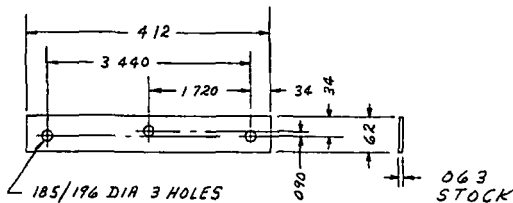
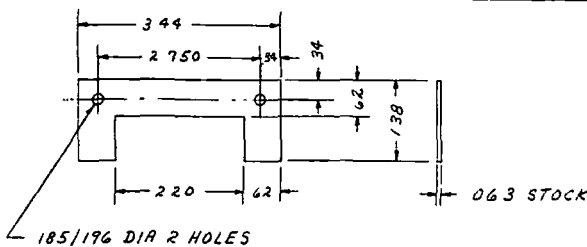


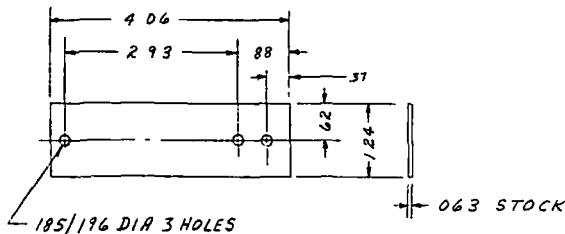
PLATE ~ MAKE FROM HASTELLOY ALLOY B,
ASTM SPEC B 333-62

ITEM NO 2



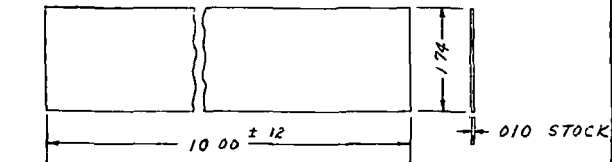
PLATE, SUPPORT ~ MAKE FROM HASTELLOY ALLOY B,
ASTM SPEC B 333-62

ITEM NO 3



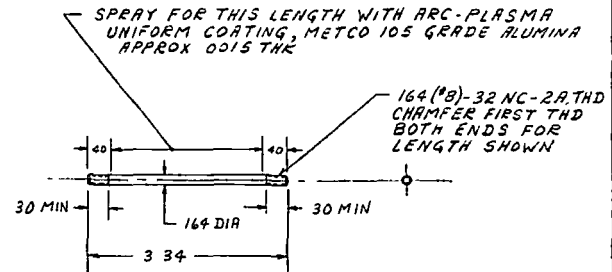
STRAP, GROUND ~ MAKE FROM HASTELLOY ALLOY B,
ASTM SPEC B 333-62

ITEM NO 4



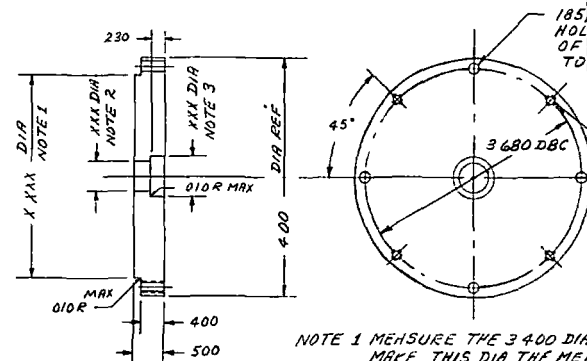
INSULATION ~ MAKE FROM BURNIL CM-2 SYNTHETIC
MICA PAPER

ITEM NO 5



STUD ~ MAKE FROM HASTELLOY ALLOY B, ASTM
SPEC B 335-58T

ITEM NO 7

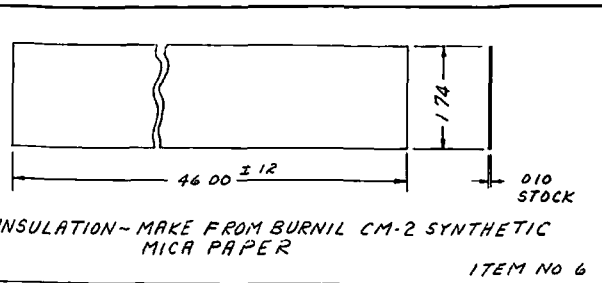


NOTE 1 MEASURE THE 3 400 DIA
MAKE THIS DIA THE MEAS
NOTE 2 MEASURE THE 460 DIA
DIA THE MEASURED D
NOTE 3 MEASURE THE 700 DIA
DIA THE MEASURED D

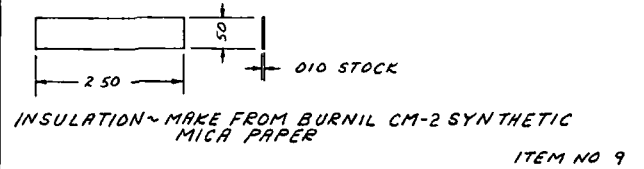
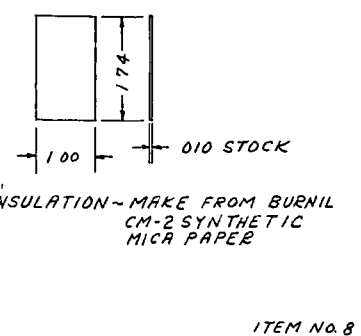
NOTE ~ DIA NOTE 2 AND DIA NOTE 3 MUST BE CONCENTRIC W
DIA NOTE 3 MUST BE CONCENTRIC WITH DIA NOTE 1 W

END BELL ~ MAKE FROM RING EDSE 326688 ITEM NO

5. LCP 170 38
TO: INTERNAL INFO & DATA
ITEM 11A ADDED 37
DIM HOLE ITEM 5-
WAS CM-2 WAS CM-1
ES- CM-2 WAS CM-1
1 Wt. Cont 5-1-65

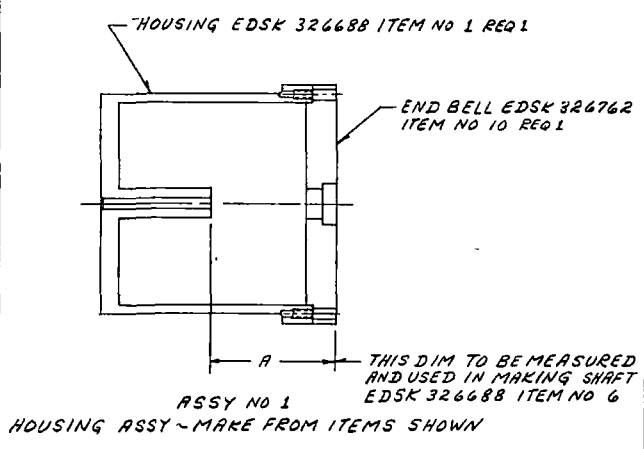


REVISIONS			
STN	DESCRIPTION	DATE	APPROVED



4 DIA 4 HOLES EQUALLY SPACED
MUST BE LOCATED WITHIN 0.05
IN NOMINAL CL LOCATION DBC
CONCENTRIC WITH DIA AT NOTE 1

190(40)-32 NF-2B THD THRU 4 HOLES
EQUALLY SPACED HOLES MUST BE
LOCATED WITHIN 0.08 OF THEIR
NOMINAL CL LOCATION



HOUSING EDSK 326688 ITEM NO 1
WRE D DIA + 0.00 - 0.02

BUSHING EDSK 326787 MAKE THIS
+ 0.40 ± 0.05

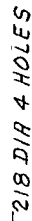
BUSHING EDSK 326787 MAKE THIS
+ 0.01 + 0.02

IN 0.005 TIR

IN 0.01 TIR

ITEM NO 10

CITY		STN	NOMENCLATURE OR DESCRIPTION		PART NO. OR IDENT NO.	SPECIFICATION	MATERIAL OR NOTE	INTERNAL INFORMATION	ITEM NO.
LIST OF MATERIAL OR PARTS LIST									
1. DRAWING NO. 83843 2. QUANTITY 1 3. PARTS LIST 4. PARTS LIST 5. PARTS LIST 6. PARTS LIST 7. PARTS LIST 8. PARTS LIST 9. PARTS LIST 10. PARTS LIST 11. PARTS LIST 12. PARTS LIST 13. PARTS LIST 14. PARTS LIST 15. PARTS LIST 16. PARTS LIST 17. PARTS LIST 18. PARTS LIST 19. PARTS LIST 20. PARTS LIST 21. PARTS LIST 22. PARTS LIST 23. PARTS LIST 24. PARTS LIST 25. PARTS LIST 26. PARTS LIST 27. PARTS LIST 28. PARTS LIST 29. PARTS LIST 30. PARTS LIST 31. PARTS LIST 32. PARTS LIST 33. PARTS LIST 34. PARTS LIST 35. PARTS LIST 36. PARTS LIST 37. PARTS LIST 38. PARTS LIST 39. PARTS LIST 40. PARTS LIST 41. PARTS LIST 42. PARTS LIST 43. PARTS LIST 44. PARTS LIST 45. PARTS LIST 46. PARTS LIST 47. PARTS LIST 48. PARTS LIST 49. PARTS LIST 50. PARTS LIST 51. PARTS LIST 52. PARTS LIST 53. PARTS LIST 54. PARTS LIST 55. PARTS LIST 56. PARTS LIST 57. PARTS LIST 58. PARTS LIST 59. PARTS LIST 60. PARTS LIST 61. PARTS LIST 62. PARTS LIST 63. PARTS LIST 64. PARTS LIST 65. PARTS LIST 66. PARTS LIST 67. PARTS LIST 68. PARTS LIST 69. PARTS LIST 70. PARTS LIST 71. PARTS LIST 72. PARTS LIST 73. PARTS LIST 74. PARTS LIST 75. PARTS LIST 76. PARTS LIST 77. PARTS LIST 78. PARTS LIST 79. PARTS LIST 80. PARTS LIST 81. PARTS LIST 82. PARTS LIST 83. PARTS LIST 84. PARTS LIST 85. PARTS LIST 86. PARTS LIST 87. PARTS LIST 88. PARTS LIST 89. PARTS LIST 90. PARTS LIST 91. PARTS LIST 92. PARTS LIST 93. PARTS LIST 94. PARTS LIST 95. PARTS LIST 96. PARTS LIST 97. PARTS LIST 98. PARTS LIST 99. PARTS LIST 100. PARTS LIST			DFTM H90 DFTM 1-21-67 ENGR 4/11/67 ENGR W. L. G. 1-21-67 TEST PROJ ENGR APPD APPD FOR BY		CODE IDENT NO. 83843 SIZE F DRAWING NO. EDSK 326762 SCALE WEIGHT SHEET				



PUNCHING~MAKE FROM HIPERCO 27 008 THK 13202AC
DIMENSIONS PER LAMINATION TYPE 138-E1

ENCLOSURE



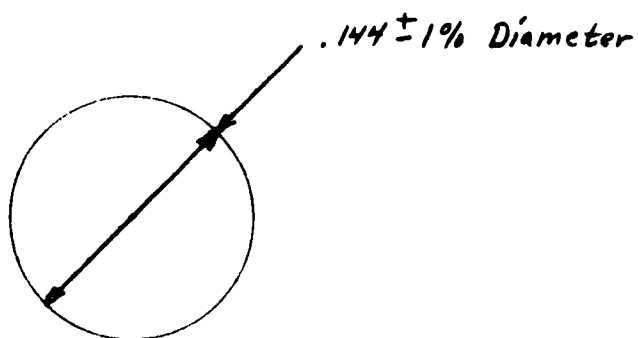


FD5K 326758

3	470 DIM W45 2 440	2	470 DIM W45 2 440
2	356 DIM W45 3 299	1	356 DIM W45 3 299
1	176 DIM W45 2 176	5	176 DIM W45 2 176
	5=176038 M3E911L		5=176038 M3E911L
	47103E 2-3-65		47103E 2-3-65
	W L Gmmt 2 2-65		W L Gmmt 2 2-65
	DELETED 1,600 DIM - 2 PLACES		DELETED 1,600 DIM - 2 PLACES
	ADDED P17 NMT DELETED		ADDED P17 NMT DELETED
	2 470 DIM - 2 PLACES - ADDED		2 470 DIM - 2 PLACES - ADDED
	P17 NMT ADDED MAX TO		P17 NMT ADDED MAX TO
	2 550 DIM - 2 PLACES		2 550 DIM - 2 PLACES
	2 W L Gmmt 5-1-65		2 W L Gmmt 5-1-65

WESTINGHOUSE ELECTRIC CORPORATION

Conductor
28% Inconel-Clad Silver



Dimensions in inches
Scale 10:1

EDSK 326536 A

Rev A 12-28-64
28% was 20% only

H.E. Keneipp 12-31-64

APPENDIX C

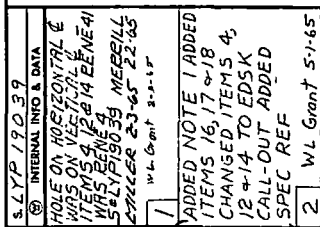
SOLENOID MATERIALS SUMMARY

APPENDIX C

SOLENOID ASSEMBLY

DRAWING NO.	TITLE	MATERIAL
EDSK 326788	Solenoid Assembly	-
EDSK 326689	Winding, Solenoid	-
EDSK 326762	Details	See Drawing
EDSK 326688	Details	See Drawing
EDSK 326620	Details	See Drawing
EDSK 326619	Forging	Hiperco 27 Alloy (27 Co - Fe)
EDSK 326793	Forging, Cut Up	Hiperco 27 Alloy (27 Co - Fe)
EDSK 326787	Bushing	99.5 Al ₂ O ₃
EDSK 326687	Solenoid Armature, Guide Rod	99.5 Al ₂ O ₃
EDSK 326686	Insulator, Tube	99 Al ₂ O ₃
EDSK 326685	Insulator, Tube	99 Al ₂ O ₃
EDSK 326684	End Plate	99.5 Al ₂ O ₃
EDSK 326680	Insulator, Tube	99 Al ₂ O ₃
EDSK 326795	Thermocouple	(Inconel Cladding, Platinel II)
EDSK 326796	Thermocouple	(Wire system, Al ₂ O ₃ Insulation)
EDSK 327502	Screw	Hastelloy Alloy B
EDSK 327503	Nut	Hastelloy Alloy B
EDSK 326533	Conductor	Nickel Clad Silver (20% Clad area) ⁽¹⁾
EDSK 326534A	Conductor	Inconel Clad Silver (28% Clad area) ⁽¹⁾

(1)- All conductors insulated with Anaconda's Anadur, a refractory-oxide-filled glass insulation.



REVISIONS			
SYM	DESCRIPTION	DATE	APPROVED

NOTE 1- WINDING POTTING PROCEDURE APPLY FOUR STRIPS OF POTTING COMPOUND W-839 3/8-1/2 INCH WIDE AND LOCATED APPROX 90° APART COATING MUST BE THICK ENOUGH TO PROVIDE CONTACT BETWEEN WINDING AND HOUSING AFTER ASSEMBLY AVOID AREA OCCUPIED BY THERMOCOUPLES ITEM 15

AR	AR	B	POTTING COMPOUND	W-839 (WESTINGHOUSE) (SEE NOTE 1)	18
AR	AR	B	PYROCERAM		17
2	2		THERMOCOUPLE	EDSK 326796	16
2	2		THERMOCOUPLE	EDSK 326795	15
1	1		NUT	EDSK 327503-2	14
1	1		WEIGHT	EDSK 326688 ITEM NO 4	13
4	4		SCREW	EDSK 327502-2	12
1	1		PLATE	EDSK 326688 ITEM NO 3	11
1	1		STAINLESS STEEL O-RING	CAT NO U-8212-00688-NPN UNITED AIRCRAFT PRODUCTS, INC DAYTON, OHIO	10
1	1		BUSHING	EDSK 326787	9
1	1		SHAFT	EDSK 326688 ITEM NO 7	8
1	-		WINDING	EDSK 326689 PN-2	7
-	1		WINDING	EDSK 326689 PN-1	6
AR	AR	B	WIRE, SST	.020 DIA 7860-2	5
2	2		SCREW	EDSK 327502-1	4
1	1		PLATE	EDSK 326620 ITEM NO 4	3
1	1		GUIDE ROD	EDSK 326687	2
1	1		HOUSING ASSY	EDSK 326762 ASSY NO 1	1

595405-1

<input checked="" type="checkbox"/> SPECIFICATION REFERENCE	<input type="checkbox"/> GOVERNMENT SPECIFICATION	<input type="checkbox"/> NAME OF PROCESS
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QTY REQD	SYM	NOMENCLATURE OR DESCRIPTION	PART NO OR IDENT NO	SPECIFICATION	MATERIAL OR NOTE	INTERNAL INFORMATION	ITEM NO
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LIST OF MATERIAL OR PARTS LIST

TOLERANCE: MAG SPEC PER 1.1, 2.1		DFTM	1/21/65	<input checked="" type="checkbox"/> AEROSPACE ELECTRICAL DEPARTMENT WESTINGHOUSE ELECTRIC CORP LIMA, OHIO U.S.A.
UNLESS OTHERWISE SPECIFIED DIMENSIONS ARE IN INCHES, DO NOT SCALE DWG		DFTM	1/21/65	
FRACTIONS: 1/16, 1/32, 1/64, 3/32, 1/2, 3/4, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, 20, 21, 22, 23, 24, 25, 26, 27, 28, 29, 30, 31, 32, 33, 34, 35, 36, 37, 38, 39, 40, 41, 42, 43, 44, 45, 46, 47, 48, 49, 50, 51, 52, 53, 54, 55, 56, 57, 58, 59, 60, 61, 62, 63, 64, 65, 66, 67, 68, 69, 70, 71, 72, 73, 74, 75, 76, 77, 78, 79, 80, 81, 82, 83, 84, 85, 86, 87, 88, 89, 90, 91, 92, 93, 94, 95, 96, 97, 98, 99, 100		ENGR	W L Grant	TITLE SOLENOID
PROCESS TO BE APPLIED AS INDICATED ON DRAWING		ENGR	W L Grant	
GOVT SPEC		MFR		CODE IDENT NO 83843
SYMBOLS W = WESTINGHOUSE S = SPEC/SOURCE CONT DWG V = VENDOR PART I = VERY IMPORTANT		TEST		
APPROVED FOR		PROJ ENGR APPD		DRAWING NO EDSK 326788
BY		APPD FOR		
SCALE 1/1		WEIGHT		SHEET

TOOL REF
(NO P/N)
NOT TOOLED
PROD TOOLED
TOOLES CANNOT
BE FURN
SHEET ASSY

THERMOCOUPLE TIPS TO BE ASSEMBLED 24 APPROX INSIDE INSULATOR FROM THIS END

ALL WINDINGS MUST BE BELOW THIS DIA

1860 TURNS PER WINDING

NOTE ~ CEMENT PLATES ITEM NO 1 AND 3 TO TUBE ITEM NO. 2, CEMENT INSULATORS ITEM NO 7 TO PLATE ITEM NO 3 WITH DUCO CEMENT

SYN		REVISIONS		DATE		APPROVED	
DESCRIPTION		DESCRIPTION		DATE		APPROVED	

QTY	REQD	SYM	NOMENCLATURE OR DESCRIPTION	PART NO	SPECIFICATION	MATERIAL	OR NOTE	INTERNAL INFORMATION	ITEM NO
4	4		THERMOCOUPLE	EDSK 326795					8
2	2		INSULATOR	EDSK 326686	PN-1				7
4	4		INSULATOR	EDSK 326680					6
AR	-	B	WIRE #20 (0.32 DIA) INCONEL-CLAD SILVER FINISHED BY ENGINEER						5
-	AR	B	WIRE #20 (0.32 DIA) NICKEL-CLAD SILVER FINISHED BY ENGINEER						4
1	1		END PLATE	EDSK 326684	ITEM NO 2				3
1	1		INSULATOR TUBE	EDSK 326685					2
1	1		END PLATE	EDSK 326684	ITEM NO 1				1

LIST OF MATERIAL OR PARTS LIST	
DEFIN	HGO
ENGR	W.L. Grant
ENGR	W.L. Grant
TEST	
TEST	
TEST	

TITLE	
WINDING, SOLENOID	

DRAWING NO	
83843	EDSK 326689

CODE IDENT NO	
83843	EDSK 326689

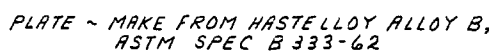
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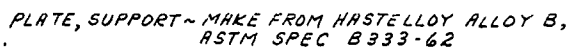
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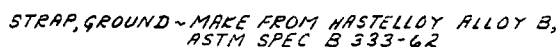
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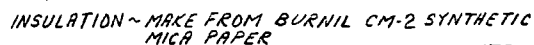
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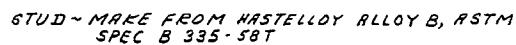
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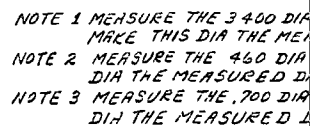
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ITEM NO 5



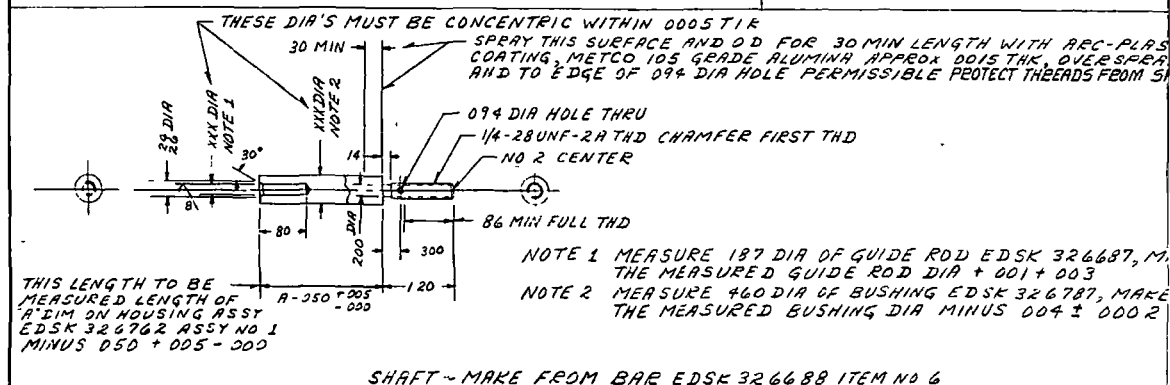
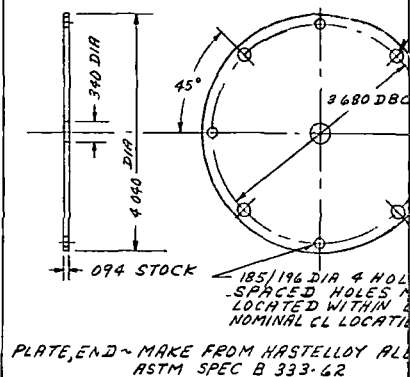
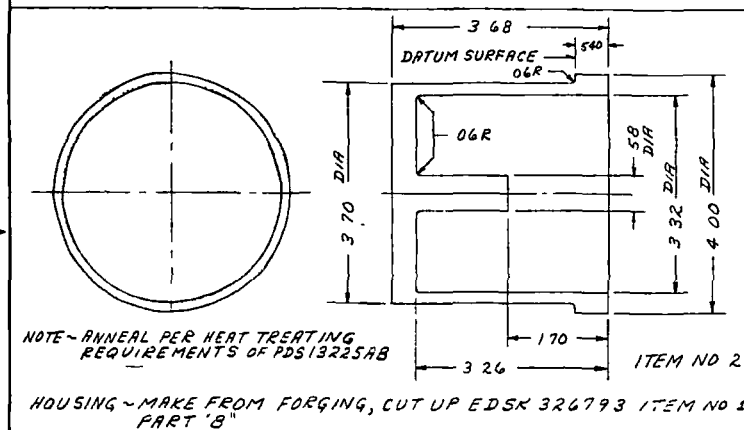
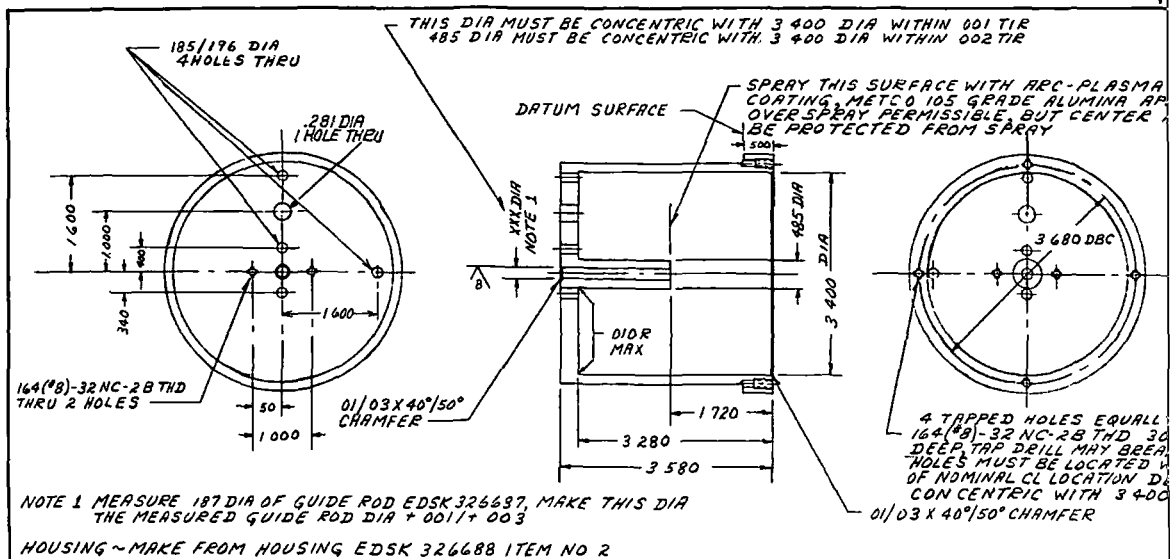
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NOTE~ DIA NOTE 2 AND DIA NOTE 3 MUST BE CONCENTRIC WITH DIA NOTE 1
DIA NOTE 3 MUST BE CONCENTRIC WITH DIA NOTE 1

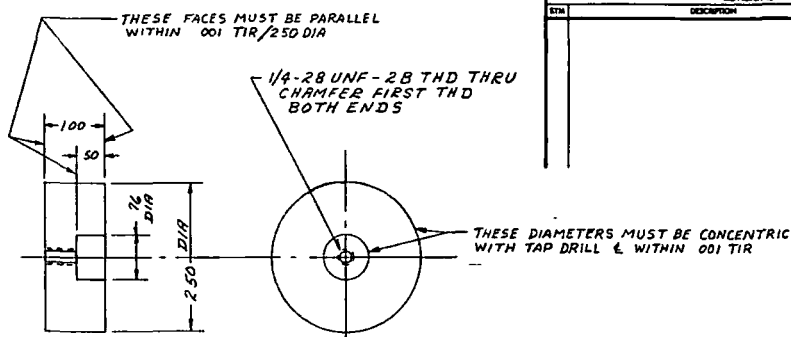
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ALCP 17038
(C) INTERNAL INFO & DATA
ITEM 114 ADDED 37
DIM. HOLE ITEM 5-
WAS '80, C/C
WAS C/S. ITEMS 5, 6
CM-2 WAS CM-1
W/ Grant 5-1-65



ITEM 2	281 DIA HOLE
ITEM 3	164(#8)-32 NC-28 THD
ITEM 4	164(#8)-32 NC-28 THD
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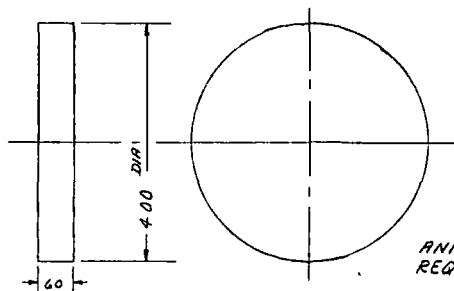
no 1



WEIGHT~ MAKE FROM MALLORY 1000 OR FANSTEEL 77

ITEM NO 4

210/218 DIA 4 HOLES
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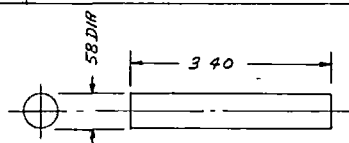
ITEM NO 3

~NOTE~
ANNEAL PER HEAT TREATING
REQUIREMENTS OF PDS 13225AB

RING ~ MAKE FROM FORGING, CUT UP EDSK 326793 ITEM NO 2
PART "B"

ITEM NO 5

9 UNIFORM
IN 200 DIA



~ NOTE ~
ANNEAL PER HEAT TREATING
REQUIREMENTS OF PDS 13225 AB

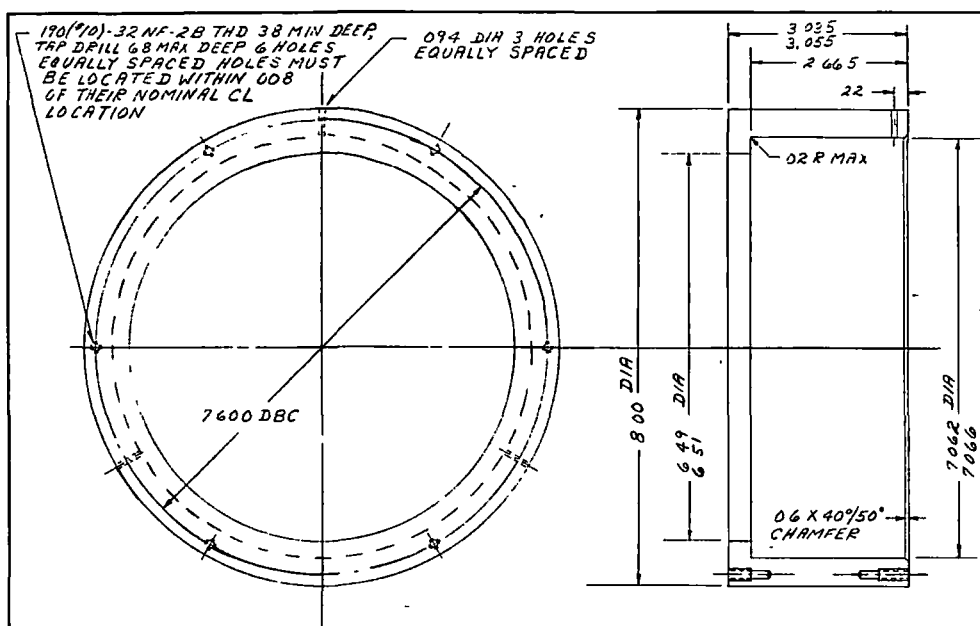
E THIS DIA
HIS DIA

BAR-MAKE FROM FORGING, CUT UP
EDSK 326773 ITEM NO 1 PART C"

ITEM NO 6

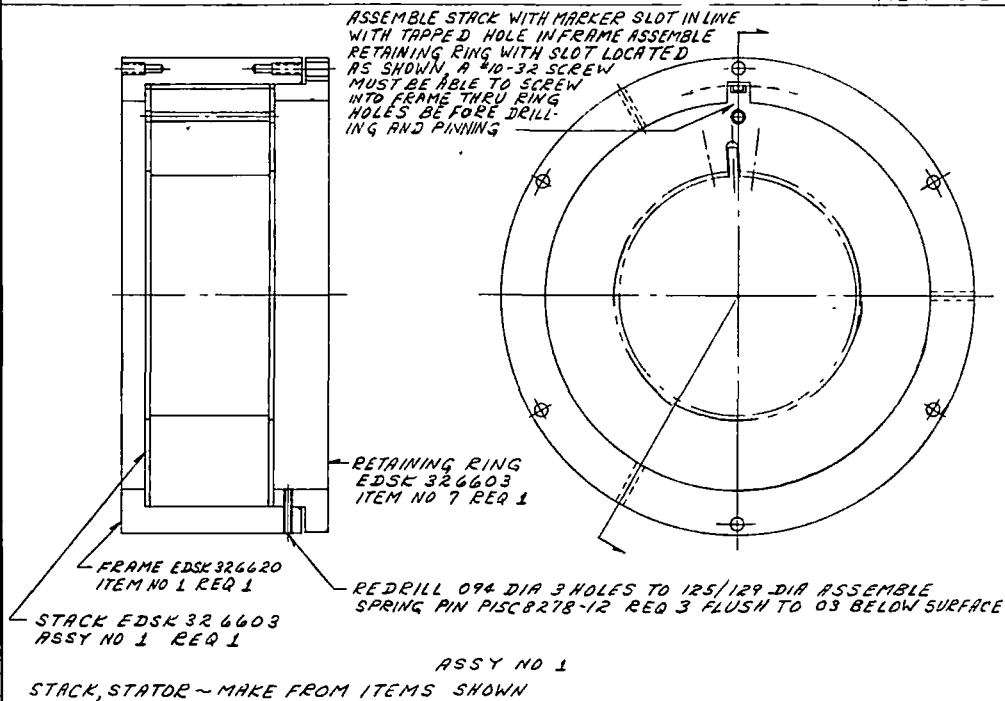
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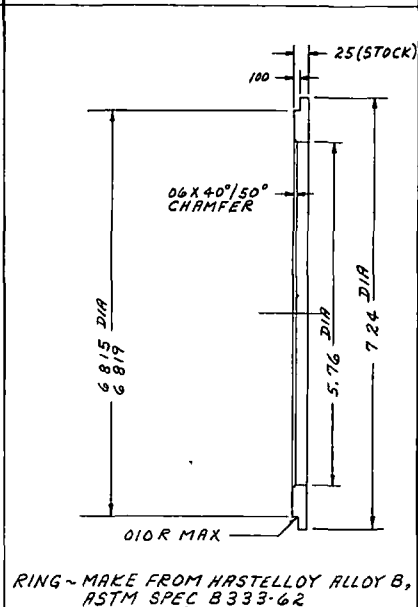
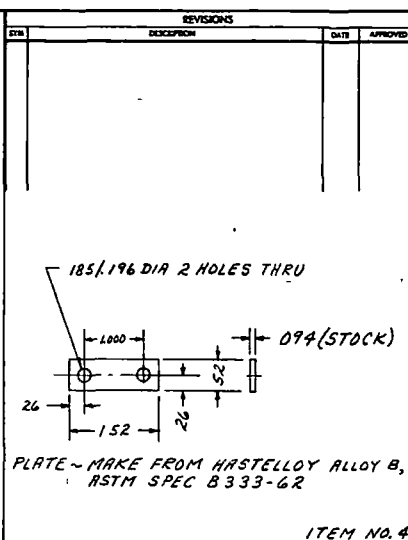
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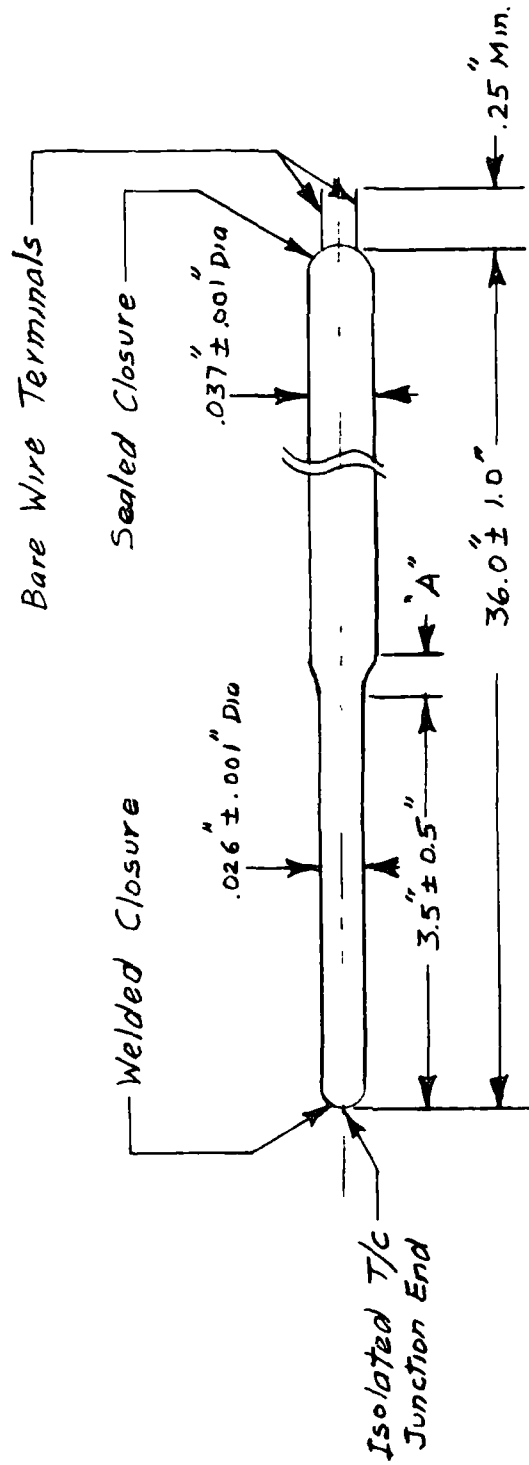
ITEM NO 1



EDSK 326603

EDSK 326603

[illegible]



"A" - Dimension as required for forming operation - 1.0 inch maximum. OD surface must be continuous thru transition.

Note: Thermocouples are to be kept straight at all times.

DUAL DIAMETER SHEATHED THERMOCOUPLE

Reference (W) D-Spec # 709747

EDSK 326795

July 2-1965

- b.) Model 602, Mazur Lapping/Polishing Machine manufactured by Westinghouse Scientific Equipment Division. The machine is equipped with a variable vibrational speed control and timer.
- c.) Model 200 B-5 Bench Type Ultrasonic Machine Tool manufactured by Sheffield Corporation.
- d.) Model F, Industrial Abrasive Unit manufactured by S. S. White Co.
- e.) Model L-300 Ultrasonic Cleaner manufactured by Landsverk Electrometer Co.
- f.) Model AST-100 Low Energy Sputtering System (Plasmavac) including a CV-18 Vacuum Evaporator manufactured by Consolidated Vacuum Corporation. The system has an ultimate vacuum of 3.5×10^{-8} torr. Sputtering operations are performed using a novel three electrode technique.

3. Wafer Slicing

The Lucalox rod was sliced into wafers ranging in thickness from 8 to 15 mils. Six wafers were sliced in the 10 mil thickness range. Pyrolytic boron nitride was wafered into 10 to 12 mil thick slices. A yield of five wafers per 1/8 inch thick block was obtained.

Lucalox rod stock and a 1 x 1 x 1/8 inch block of pyrolytic boron nitride was mounted and sliced as follows:

a. MOUNTING FOR SLICING OPERATIONS

A 3/4 inch diameter x 3 inch long Lucalox rod was cemented to a glass plate with LOC-Wax 20 (Geoscience Instrument Co.) and sliced as shown in Figure III-2. A 1 x 1 x 1/8 inch block of pyrolytic boron nitride was also mounted and sliced in a similar manner.

b. SLICING

Slicing Machine: WMSA Precision Wafering Machine

Diamond Wheel: Metal Bonded Diamond Wheel manufactured by Norton Co., Abrasives Div.
Wheel Diameter - 5 inches

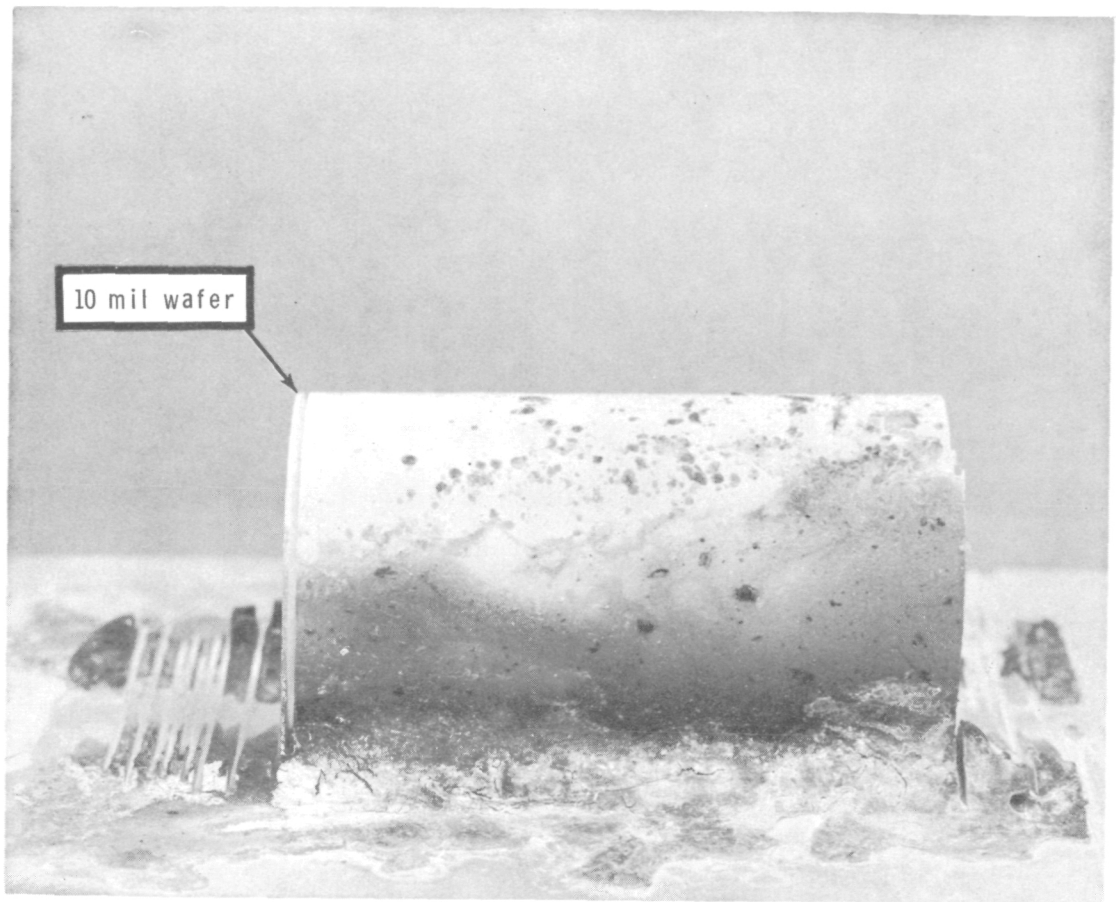


FIGURE III-2. Lucalox Rod Stock With 10 Mil Sliced
Wafer Held in LOC-Wax 20
(Geoscience Instrument Co.)

Wheel Thickness - 0.017 inches
Manufacturer's Identification 1-5 x 0.015 x 5/8,
D220-N100M-1/8, ME 73082

Travel Speed: Wafers were cut at longitudinal table speeds
in the range from 0.059 to 0.111 inches per minute.

Coolant & Wheel RPM: Water was used as a coolant. The
wheel was rotated at 3800 rpm during all cutting
operations.

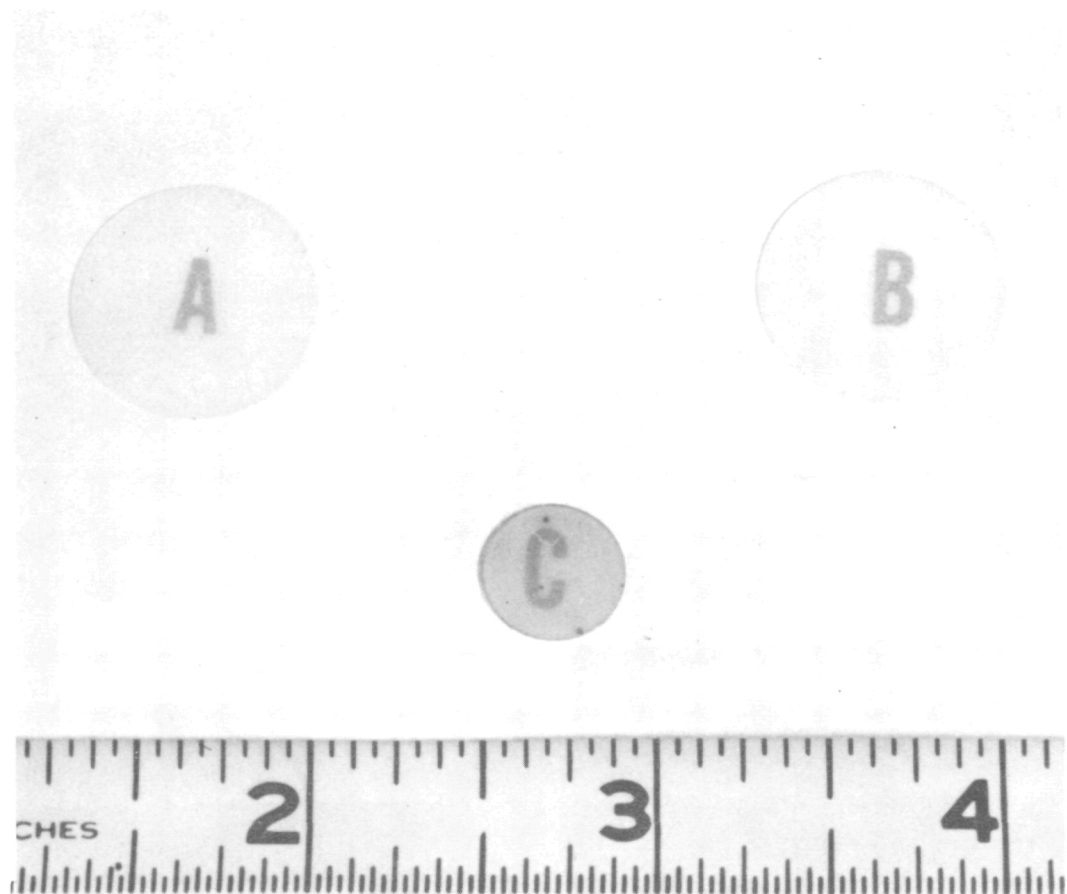
4. Lapping Procedures & Results

The sliced wafers were bonded to a 3 x 3 inch piece of plate glass with a lightly frosted surface to improve adhesion. A black, low melting point wax (Pyseal, Fisher Scientific Co.) was used. The glass plate was heated on a hot plate to the flow temperature of the wax $\sim 158^{\circ}\text{F}$ ($\sim 70^{\circ}\text{C}$). After applying a thin coating of wax, the preheated wafers were positioned on the plate and the excess wax was then squeezed out from under the wafers by applying light and uniform pressure to each individual wafer.

Controlled wafer thickness were achieved by cementing metal (steel) shim strips along two edges of the glass. An epoxy cement was used to permanently bond the shim strips to the glass. A series of glass plates with different thickness shims were made ranging from 2 to 10 mils. Each set of wafers were sequentially lapped by transferring them to thinner shimmed plates until the desired thickness was obtained (2 to 6 mils). Figure III-3 shows two sliced Lucalox wafers lapped to 6 mils. In addition, a 10 mil hot pressed Al_2O_3 disk is also shown. All thickness measurements were made with a hand held micrometer.

Lapping and polishing operations were performed with a Mazur Lapping/Polishing Machine manufactured by Westinghouse Scientific Equipment Department. The machine operation is based on a variable-speed, eccentrically-rotating plate with interchangeable trays that contain different grades of abrasives. Each tray has a plate glass removable base. Lapping and polishing can be done directly on the glass plate or the plate can be covered with a variety of bonded surfacing materials.

Table III-3 shows a lapping and polishing sequence which was found to yield the least number of grain pull outs in Lucalox wafers. In many instances, large areas of the Lucalox wafer surfaces were free from



A & B - 6 Mil Lucalox Wafer Cut and Lapped
From 3/4 Inch Rod Stock

C - 10 Mil Hot Pressed Al_2O_3 Prepared
Under WAED Contract AF33(615)360

Note the Translucence of the Materials.

FIGURE III-3. Dielectric Materials

TABLE III-3. Lapping and Polishing Operations

Lucalox Wafers: 10 to 12 mil starting material, 3/4 inch diameter						
Operation Sequence	Abrasive	Lapping Fluid	Lap Plate Surface	No. of Wafers Per Holder and Applied Pressure	Machine Speed (Dial Setting)	Comments
1	600 Grit Boron Carbide (Norton Co.)	Elgin Watch Company (Dymo)	Plate glass	6 wafers 3 lbs.	2	Lapped to ~6 mils
2	30 micron diamond (Geoscience Instrument Co.)	same	same	same	same	pre-finish 1/2 hr. each side
3	15 micron diamond	same	same	same	same	same
4	same	same	Pre K, bonded surface material (Geoscience Instrument Co.)	same	3	rough polish 1/2 hr. each side
5	6 micron diamond	same	Fine K bonded surfacing material (Geoscience Instrument Co.)	same	4	final polish 1 hr. each side
6	1 micron diamond	same	Fine K	same	4	remove scratches 1 hr. each side
Pyrolytic Boron Nitride: 10 to 12 mil starting material, 1 x 1 inch square						
1	600 Grit Boron Carbide	Elgin Watch Company (Dymo)	Plate glass	One wafer 1 in ² , 1 lb.	2	Lapped to ~3 mils
2	15 micron diamond	same	Pre K	same	2	Lapped to 2 mils
3	6 micron diamond	same	Fine K	same	2	Lapped and polished to final thickness 1 to 1.5 mils. No pin holes visible

pull outs; however, occasional grain voids were found in all wafers when their surfaces were completely scanned at high magnification. In general, the highest concentration of voids or pull outs were found in the central area of a wafer. Wafers that were lapped to 2 mils had voids that penetrated the entire thickness of the wafer resulting in pin holes. It appears, at present, that 5 to 6 mil wafers can be produced free of pin holes and these wafers could be electroded and tested.

Although the measured thickness of these wafers is in the range of 5 to 6 mils their effective thicknesses (due to large grain pull outs) after electroding might be in the 4 to 5 mil range. In addition, surface voids would contribute to high voltage stress concentrations during electric strength tests causing premature breakdown.

A pyrolytic boron nitride wafer has been lapped and polished with relative ease to a measured thickness that varied from 1 to 2 mils across the wafer surface (Table III-3 shows the lapping sequence). This wafer exhibited an unusually high degree of flexibility (about equivalent to mica sheet or paper in the same thickness range). No visible pin holes or pull outs could be seen. At this stage, pyrolytic boron nitride appears very promising.

Mechanical stresses that occur during machining of brittle materials cannot be entirely eliminated but they can be minimized. Table III-4 contains a listing of possible causes of grain pull outs together with appropriate corrective actions.

5. Dielectric Measurements

It is not planned to make extensive high-temperature dielectric measurements until an inventory of qualified, single-wafer capacitors are accumulated. Room temperature measurements will be made to determine test sample qualification as follows:

- a.) Dielectric constant and $\tan \delta$ @ 1 KC
- b.) Insulation resistance
- c.) Voltage Test: 500 to 1000 VDC

Preliminary designs are being considered for constructing a small cold-wall vacuum furnace capable of heating test specimens to 1100°F in a

TABLE III-4. Possible Causes of Grain Pull Outs in Machined Lucalox Wafers

Cause	Corrective Actions
<p>1. a) Excessive mechanical stress as a result of wafering operations b) Surface initiated fractures due to abrasion damage c) Frictional heating</p>	<p>a) Improve coolant efficiency at cutting interface (higher flow rates, direct impingement of coolant at cutting surface) b) Increase cutting wheel rpm c) Decrease feed rate d) Use smaller grain size diamonds in slicing wheel e) Anneal wafers after slicing or heat and quench</p>
<p>2. a) Mechanical stress exceeding cohesive strength of individual crystallites during lapping and polishing operations. b) Abrasion surface damage-excessive stress concentrations at point contacts between abrasive and wafer surface c) Frictional heating-thermal stress gradients</p>	<p>a) Use smaller grain size abrasives for all stock removal and polishing operations b) Determine best compromise between stock removal rates, abrasive size and type, applied pressure and lap speed c) Determine if hardness of lap surface is a significant factor (glass vs. metal and cloth) d) Investigate low amplitude, high frequency lap motion e) Maintain high abrasive concentrations at wafer-lap plate interface and minimize frictional heating</p>
<p>3. a) Non-uniform grain size material with a large number of excessively large grains b) High mean grain size - decreased grain boundary stresses with decreasing grain size c) Non-uniform density distribution from center to outer edges of wafer d) High internal stresses due to crystal anisotropy in expansion coefficients and elastic moduli</p>	<p>a) Obtain smaller mean grain size material b) Investigate hot pressed material with controlled-uniform grain size. The microstructure of hot pressed Al_2O_3 shows that the long axis of individual crystallites lie in a direction parallel to wafer surfaces. Thus, an occasional grain pull out would result in a more shallow surface depression</p>

vacuum in the 10^{-7} - 10^{-8} torr range. The vacuum pumping unit (Model CV18) presently combined with our sputtering unit can evacuate an 18-inch glass bell jar into the 10^{-8} torr range.

C. PROGRAM FOR NEXT QUARTER.

- 1.) Fabrication of single crystal sapphire wafers will be started.**
- 2.) Lucalox and pyrolytic boron nitride machining and polishing studies will be continued.**
- 3.) Sputtered rhodium and platinum electrodes will be applied to lapped and polished wafers of the candidate dielectric materials.**
- 4.) Dielectric tests will be started including qualification tests and high temperature dielectric measurements.**

SECTION I V

PROGRAM III - BORE SEAL DEVELOPMENT AND COMBINED MATERIAL INVESTIGATIONS UNDER A SPACE SIMULATED ENVIRONMENT

The bore seal effort under Task 1 will evaluate promising ceramic-metal sealing systems in potassium and lithium vapor at temperatures to 1600°F for 2000 hours. Elevated temperature seal strength and vacuum tightness will be determined. A four-inch diameter bore seal loaded with potassium will be used in a 5000-hour stator endurance evaluation at temperature and in a high-vacuum environment to confirm data determined on smaller geometries.

Two five-thousand hour tests will be run under Task 2 on a stator which typifies the construction of an inductor alternator or a motor. The first will be run between 800 - 1100°F temperature. The second test will be run with a bore seal at temperatures between 1100 - 1600°F. All will be tested at high vacuum (greater than 10^{-8} torr) under electric and magnetic stresses.

A transformer and two solenoids under Tasks 3 and 4 will be similarly tested under high-vacuum and at elevated temperature. The purpose will be to evaluate the combined effects of electric and magnetic stresses, and high vacuum on combinations of materials suitable for application to advanced space electric power systems. One solenoid test will be under d-c excitation and the other under intermittent excitation so the effects of an invariant electric stress can be investigated.

A. TASK 1 - BORE SEAL DEVELOPMENT.

1. Summary of Technical Progress

a.) The alkali metal loading facility has been undergoing modification to make provisions for manipulative controls through the vacuum chamber walls and for the electron-beam welder which will be used in final closure of the alkali metal corrosion test capsules.

b.) Microhardness surveys were completed across the metal-braze alloy-ceramic seals which survived alkali metal exposure on a previous bore seal program conducted on NAS 3-4162.

c.) Two additional ceramics and six braze alloys have been ordered. Delivery is scheduled for the second quarter.

d.) Tentative design of the model four-inch diameter bore seal has been completed.

2. Discussion

a. FACILITY DESIGN

The chamber which was utilized in performance of the previous contract NAS 3-4162 is being modified to include provisions for loading test capsules with alkali metal in a vacuum of 1×10^{-5} torr and for electron-beam welding of the final closure. This requires manipulative control through the vacuum chamber walls. A schematic layout of the loading chamber is shown in Figure IV-1. The capsule loading techniques which are to be used with the modified equipment have been evaluated. The manipulators for the loading chamber and the electron-beam welder have been delivered. A 1500 ℓ /sec diffusion pump is on order.

The present vacuum system for alkali metal loaded capsule testing furnace is being converted to ion pumps to achieve a vacuum of 10^{-9} torr. The modifications associated with this change are shown in Figure IV-2. The 400 ℓ /sec ion pump and power supply for the dual vacuum environmental test furnaces are on order.

b. BORE SEAL MATERIALS AND SEALING STUDY

The ceramic bodies (low silica 99.8% beryllia and low silica yttria), refractory metal member (low oxygen columbium-1% zirconium),

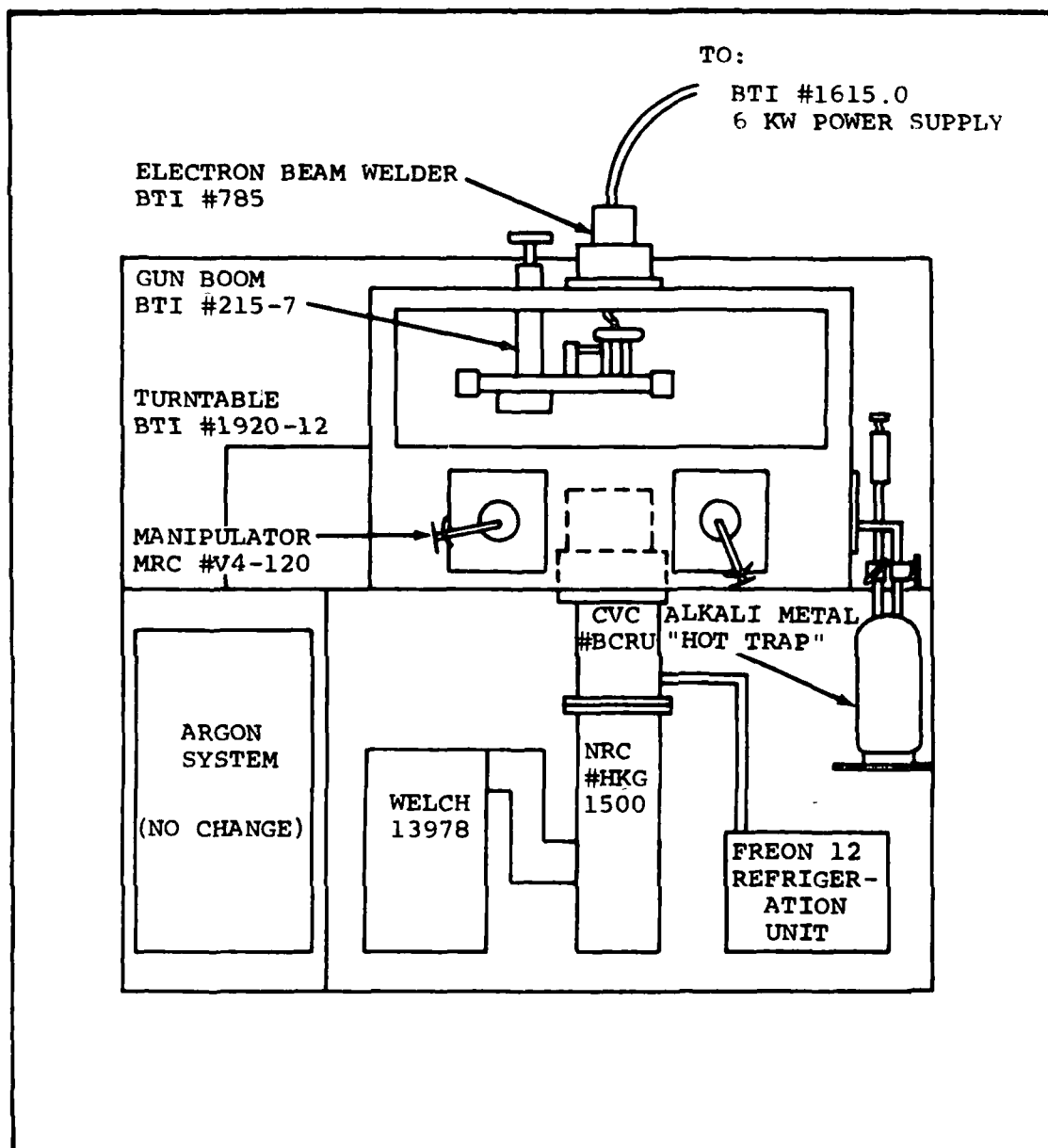


FIGURE IV-1. Schematic of the Capsule Fabrication and Loading Equipment

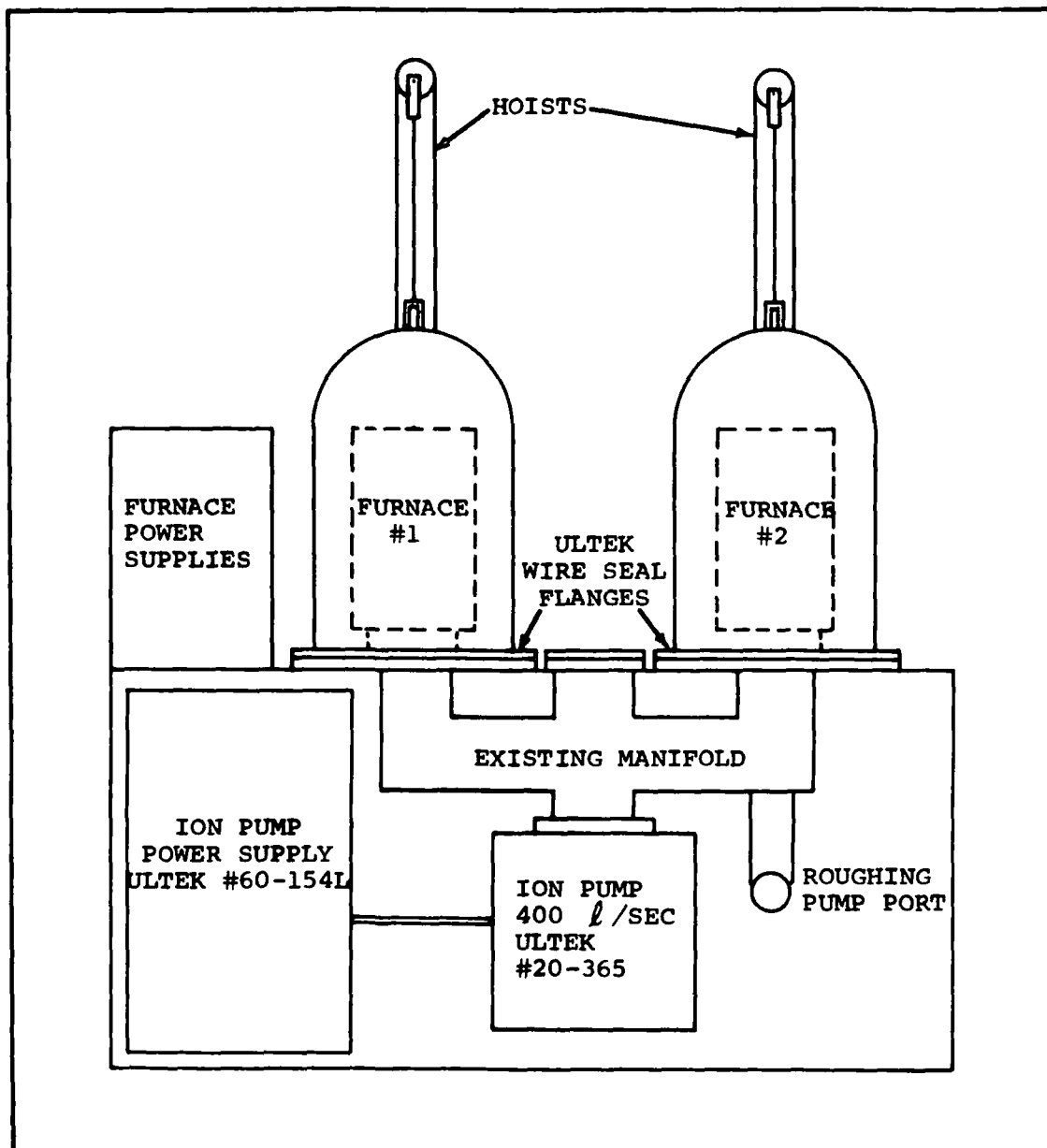


FIGURE IV-2. Schematic of Dual Vacuum Furnace Modifications Using Ion Pumping

and special braze alloys required in the bore seal test program are all long lead-time items. Delivery of most of these materials is scheduled in the second quarter. A discussion of the high-temperature seal fabrication program follows.

1) Active Metal Brazing

The evaluation of ceramic-to-metal seals which withstood 500 hour alkali metal exposure at 1600°F on Contract NAS 3-4162 was continued.

Thermalox 998 (99.8% BeO ceramic made by Brush Beryllium Co.) joined to columbium-1% zirconium alloy with active metal braze of composition 56Zr-28V-16Ti exhibited a modulus of rupture strength averaging 11,800 psi after exposure at 1600°F for 500 hours in potassium vapor. This seal system also exhibited a similar strength after exposure to lithium at 1000°F for 500 hours. This seal system will be tested for a period of 2000 hours in high purity potassium and lithium at 1600°F on this program.

Variable joint micro structure of the seals previously tested was attributed primarily to brazing parameters and aging rather than alkali metal exposure. ⁽¹⁾

The effects of 500 hour aging on the active-metal-brazed seals is being determined by microhardness surveys and electron microprobe analyses. Since the ceramic-to-braze interface region is most susceptible to aging effects, emphasis is being placed on the study of this area.

In addition to the 56Zr-28V-16Ti, seals made with 75Zr-19Cb-6Be and 48Zr-48Ti-4Be retained strength after exposure in previous tests. ⁽²⁾ These and several other high-temperature, low-oxygen active metal-braze alloys will be screened.

- (1) - Kueser, P. E., et al, Bore Seal Technology, NASA-CR-54093, Contract NAS 3-4162.
- (2) - Ibid.

The yttrium-columbium and yttrium-beryllium alloy systems will also be investigated. The Cb - Y alloys are markedly softer than pure Cb; even small additions of Y (< 1%) produce significant reduction in hardness and increases the cold workability of columbium. This effect may be caused by the impurity scavenging of yttrium. (3)

Since corrosion of the active-metal-braze joint proceeds at the ceramic-to-braze interface by reaction with the oxides which are formed, it is desirable to have braze systems which produce reacted oxides having maximum thermodynamic stability. Yttrium oxide and beryllium oxide have such stability. Potential suppliers, including Battelle Memorial Institute, Materials Research Corporation and Rodney Metals, have been contacted to obtain pure, low-oxygen yttrium metal for the braze alloy investigation.

Ceramic-to-metal seals will be made using braze alloys in foil form, when practicable, to minimize oxygen contamination.

Battelle Memorial Institute which supplied active-metal-braze alloys on NAS 3-4162 will supply active braze alloys in the form of foil for those useful compositions which are ductile enough to roll.

2) Alternate Seals

A limited number of seals will be made as a back-up for the active-metal-sealing systems. Ceramic-to-metal seals will be fabricated by utilizing several of the following alternate approaches:

- a) Pressure seals or diffusion bonding
- b) Evaporated metallizing
- c) Gradient seals

(3) - Spedding, F. H. and A. H. Daane, The Rare Earths, John Wiley & Sons, Inc., N. Y. C. 1961.

- d) Ion sputtering
- e) Combinations of ion sputtering or evaporated metallizing, chemical vapor deposition, and brazing.

In pressure bonding, a metal diffusion plating or washer is bonded at temperature to a metallized ceramic. Pressure may be applied externally or generated through thermal mismatch of ceramic and metal members. The coating on the ceramic is an alkali metal resistant metallizing.

Evaporated metallizings showed promise in the work completed on the previous program. ⁽⁴⁾ Chemical vapor deposited columbium-1% zirconium or other alkali-metal-resistant deposits will be used. This approach appears most promising as a wetting aid for active-metal brazing of large surface areas.

A gradient seal consists of an integrally bonded composite structure of metal and ceramic which varies from nearly 100 percent ceramic to 100 percent metal in a continuously graduated manner. The metal end (or ends) may then be joined to the electrical apparatus frame by conventional methods. Combinations of ceramic and metal will be chosen from rhenium, columbium, alumina or beryllia, or other appropriate materials.

Ion sputtering of molybdenum or Cb - 1Zr onto a ceramic substrate ^(5, 6) followed by a thicker coating of metal (i. e. Cb - 1Zr) applied by chemical vapor deposition ⁽⁷⁾ and brazing to a metal member appears promising for seal

- (4)- Kueser, P. E., et al, Bore Seal Technology, NASA-CR-54093, Contract NAS 3-4162.
- (5)- Mattox, D. M., Sandia Corporation, Reprint, SC-R-64-1330, September 1964.
- (6)- Maisset, L. I., and P. M. Schailbe, "Thin Films Deposited by Bias Sputtering", Journal of Applied Physics, Vol. 36, No. 1, p. 237, Jan. 1965.
- (7)- Reed, L. and R. McRae, "Evaporated Metallizing", American Ceramic Society Bulletin, January 1965.

fabrication. This method offers the possibility of a clean, highly adherent interface. The system limits oxygen concentration to one highly polarized Be-O-Cb or Be-O-Mo layer (8,9) which is immune to solution by potassium.

At the present time, methods b, d, and e are preferred and effort on supplementary seal systems will be emphasized in these areas.

c. CERAMIC MATERIALS

The ceramic bodies which survived the 500 hour, 1600°F potassium exposure on the previous program (NAS 3-4162) were:

Thermalox 998 - 99.8% beryllium oxide by Brush Beryllium Co.

Sapphire - 100% alumina by Linde Co.

Lucalox - 99.8% aluminum oxide by General Electric Co.

Experimental results established the desirability of low silica content ($< 0.05\%$) in ceramics intended for potassium exposure at 1600°F.

Sapphire and Lucalox are useful for feedthrough geometries but are not available in large sizes (greater than four-inch diameter) for actual bore seal construction. Coors Porcelain Co., Western Gold and Platinum Co., General Electric Co., Linde Co., Corning Glass Co., Alite Div., and U. S. Stoneware Co. were contacted to determine the availability of fabricable, low-silica ($< 0.05\%$) alumina. Such bodies are not available at the present time. However, several experimental bodies are being developed by Coors Porcelain Co.

- (8) - DiStefano, J. R. and A. P. Litman "Effects of Impurities in Some Refractory Metal-Alkali Metal Systems", Corrosion, Vol. 20, No. 12, p. 392, Dec. 1964.
- (9) - Weyl, W. A., Coloured Glasses, p. 343, Society of Glass Technology, Sheffield, Rp. Dawson, London.

Table IV-1 shows the spectrographic silica analyses of several Coors experimental alumina bodies, and the U. S. Stoneware (Alite Division) alumina body No. 610. The analyses of several bodies which were used on the previous program are also shown. The results indicate that a number of the Coors alumina bodies have the desirable low-silica content. When available, they will be considered for use as a back-up ceramic.

Two new ceramic bodies will be tested in potassium on the present program. These have been selected as

- a) 99.9% beryllium oxide or low silica (< 50ppm SiO₂) 99.8% beryllium oxide (Brush Beryllium Co.)
- b) 99+% yttrium oxide (Coors Porcelain Co.)

The rare earth oxides possess good thermodynamic properties. However, experience in fabricating bodies is limited (10 and 11).

The Brush 99.8% beryllia body has been ordered for use on ceramic outgassing and braze alloy screening investigations. In addition, ceramics which survived exposure on the previous program and the new ceramic bodies which withstand 500 hour exposure at 1600°F will be subjected to 2000 hour tests.

3. Program for the Next Quarter

- a.) Complete design and construction of loading and test facility.

- (10) Ploetz, G. L., C. W. Krystyniak, H. E. Dumas, "Sintering Characteristics of Rare Earth Oxides", Journal of American Ceramic Society, Vol. 41, No. 12, p. 550, 1958.
- (11) Perez, M., Y. Gorba, F. Queryroux, and R. Colloriques, Bull. Soc. Franc. Min. Crist., Vol. 84 (4), pp. 401-402, 1961.

TABLE IV-1. Spectrographic Silica Analyses of Selected Ceramics

Supplier Designation of Ceramic Body	Nominal Composition	SiO ₂ Content (a) (Weight Percent)
Alite #610 Alumina	(c)	1.5
Ei3-3 Alumina (Wesgo)	99.7 Al ₂ O ₃ -0.1 SiO ₂ -0.1 MgO	0.2
Thermalox 998 Beryllia (Brush Beryllium Co.)	99.8% BeO	0.015 ^(b)
Lucalox Alumina (G. E.)	99.7% Al ₂ O ₃ -0.25 MgO	0.02
Sapphire Alumina (Linde)	100% Al ₂ O ₃	0.015
Coors #1 Alumina	(c)	0.005
Coors #2 Alumina	(c)	0.03
Coors ATD 995 Alumina	(c)	0.04
Coors X Alumina	(c)	0.02
Coors Y Alumina	(c)	0.025
<p>(a) Eitel-McCullough, Inc. All data \pm 200%</p> <p>(b) American Spectrographic Laboratories, San Francisco</p> <p>(c) Compositions were not given by supplier; all are "high alumina".</p>		

- b.) Complete ordering of materials for screening of ceramic and brazes.
- c.) Initiate ceramic outgassing study.
- d.) Start experimental work on screening of braze alloys and alternate seals.

B. TASK 2 - STATOR AND BORE SEAL.

1. Summary of Technical Progress

- a.) A stator design for the first 5000-hour test has been completed. All long lead time materials are on order.
- b.) The two vacuum environmental test furnaces and associated control equipment were ordered. Both furnaces are now under construction and should be delivered by the 3rd week in April.
- c.) A design for vacuum furnace test specimen thermocouple feedthroughs which are compatible with special shielded thermocouples has been completed.
- d.) A test specimen thermocouple design specification has been released for quotations. (This thermocouple design will be used on Tasks 2, 3, and 4.)

2. Discussion

The assembly and detail drawings which define the components and assemblies that make up the stator are given in Appendix A. A summary of the major materials used in the constructions are also shown in the Appendix. The sequence of drawings begins with the stator assembly, followed by stack assembly drawings, details (several parts to a drawing, identified as items and assemblies) and component parts. Two stator stack drawings are included. EDSK 326602 shows a practice winding stack using plain copper conductors and steel punchings. This assembly will be used to develop coil forming procedures and coil installation techniques before working with the high-temperature conductors.

In some cases the detail drawings show parts that apply to the transformer or solenoid models. In such cases the same drawings are repeated in the drawings covering the subject models.

The stator design is based on the solid-rotor, inductor generator concept using one stack of laminations. It also typifies a motor construction. The excitation coils found in an inductor generator have been eliminated from the design since the winding is similar to the solenoid winding used in Task 4. The stack is made up of Hiperco 27 (27 Co-Fe) laminations, 0.008 inch thick, which are held in place in a Hiperco 27 frame by a bolted and pinned retaining ring. The laminations will be

annealed in a hydrogen atmosphere for one hour minimum at a temperature of 1492 to 1672°F. Forgings will be annealed in a rough machined condition in an exotherm gas atmosphere for one hour minimum at a temperature of 1492 to 1672°F.

Ceramic slot liners insulate the conductors from the laminations and from each other. Non-magnetic end bells are attached to each end of the magnetic frame.

The stator windings consist of three 12-turn coils which are installed in the frame in the same manner as in a standard three-phase machine. Thermocouples are located in two of the 12 slots filled by each winding. Additional thermocouples are positioned in the laminations; at the outer edge (OD) of the laminations, and attached to the outside of the magnetic frame and to coil end-turns.

The high-temperature stator to be used in conjunction with the bore seal will be of the same design except that additional members will be required to support the bore seal in position in the stator cavity. In addition, a special high-temperature stator winding and insulation system capable of extended operation at approximately 1400°F has been specified for the bore seal stator. The first stator will operate at 1100°F and will utilize a modified Anaconda Anadur (refractory oxide filled fiber glass) high-temperature insulation system evaluated under NAS 3-4162.

Figure IV-3 is a cutaway drawing of the vacuum furnace which shows the stator installed in the furnace hot zone. Stator thermocouple leads and winding leads are brought up through holes in the top heat shields to their respective feedthroughs in the chamber wall. The chamber is of double wall construction with baffles between the walls to channel cooling water flow. The chamber top cover is also double walled to provide a path for cooling water.

Figure IV-4 is a sketch showing the details of the vacuum furnace thermocouple feedthroughs. The thermocouples pass through the hollow tubes and are brazed to the tubes outside the chamber to form a vacuum-tight seal. An inner ceramic disc provides mechanical support and acts as a thermal radiation shield.

The thermocouple construction consists of two wires in an inconel sheath with Al_2O_3 powder (99%) insulating the wires from each other and from the sheath. The wire system is Platinel II (platinum - rhodium - gold) made by Engelhardt. Final sizing (OD) is done by drawing the complete assembly.

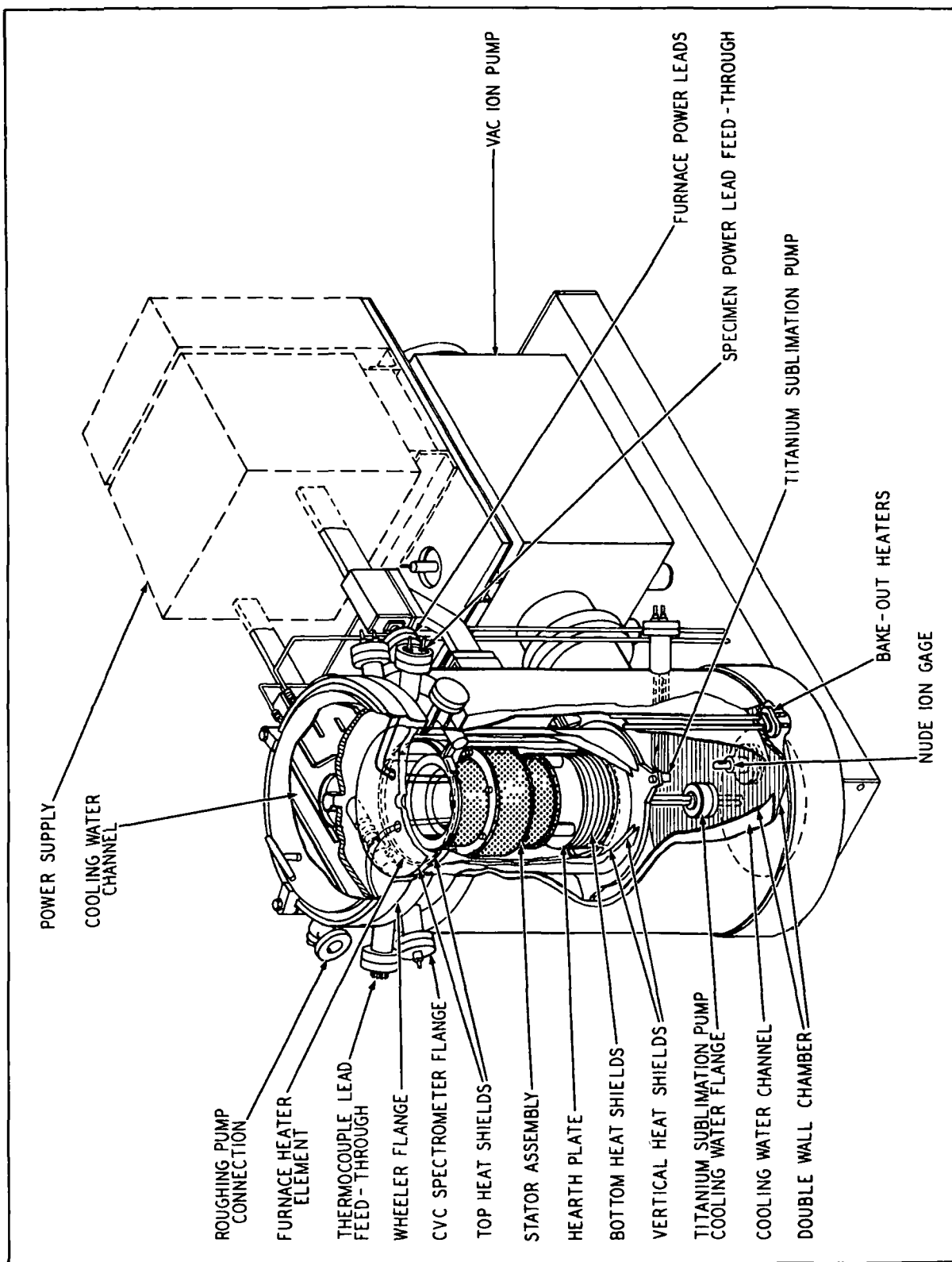


FIGURE IV -3. Cutaway View of Vacuum Furnace Showing the Stator Test Specimen Installed

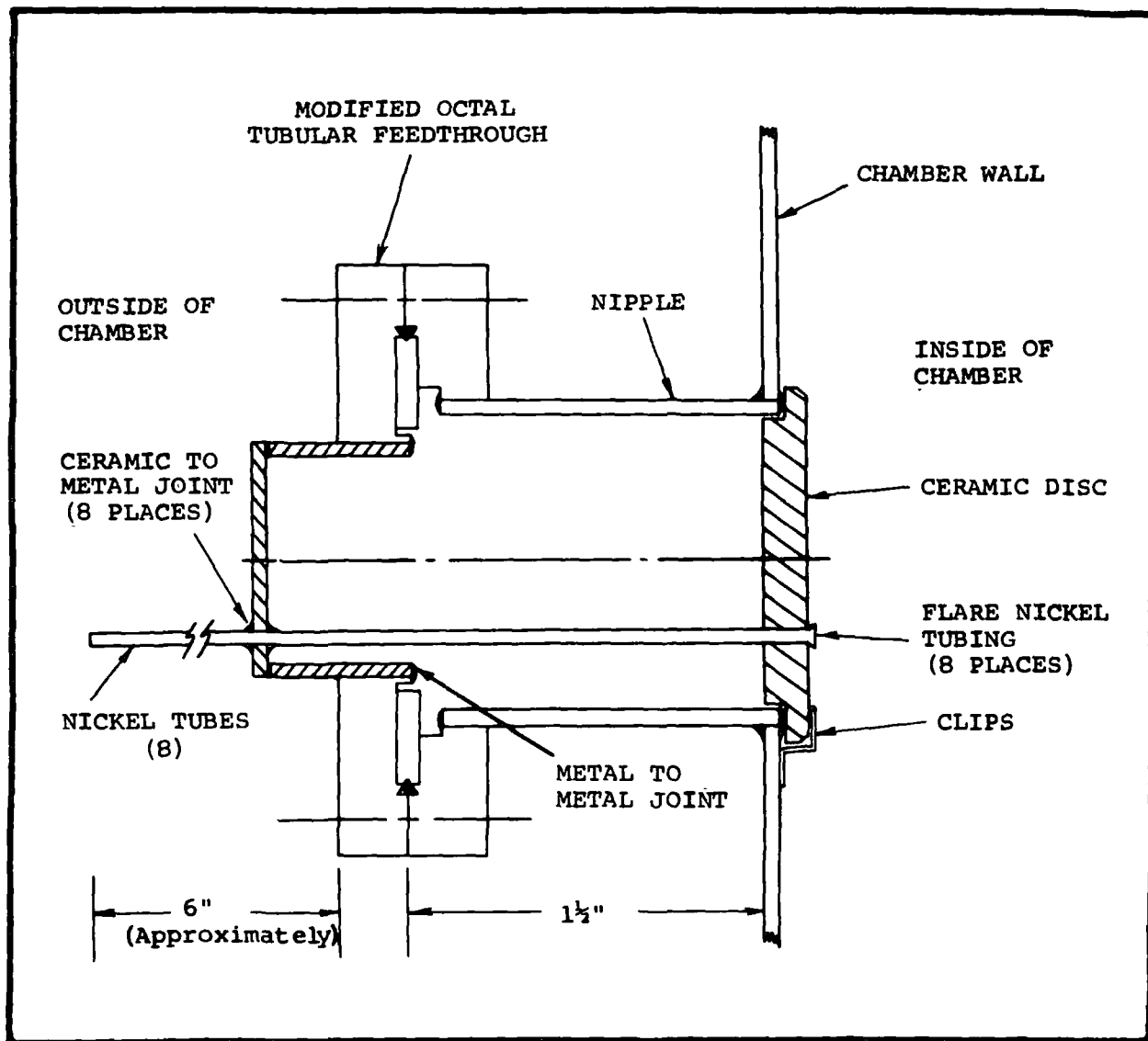


FIGURE IV-4. Details of Thermocouple Feedthrough Flange
on the Vacuum Furnace Chamber

3. Program for the Next Quarter

- a.) Expedite the procurement of long lead time materials.
- b.) Coordinate vacuum furnace delivery, installation and check-out.
- c.) Build a practice stator assembly as a check on coil forming techniques.
- d.) Continue the manufacture of stator detail parts and begin test specimen assembly.
- e.) Evaluate a new high temperature plasma arc sprayed insulation system. The insulation is high purity Linde A (Al_2O_3).
- f.) Evaluate Handy & Harmon lithobraz 'BT' and Braze 360 brazing alloys for thermocouple feedthrough brazing.
- g.) Continue the 1500°F, 5000-hour vacuum stability tests on Inconel-clad silver.

C. TASK 3 - TRANSFORMER.

1. Summary of Technical Progress

- a.) A transformer design has been completed and all long lead time materials are on order.
- b.) An engineering model release has been prepared to manufacture the parts in the shop.

2. Discussion

The assembly and detail drawings which define the components and assemblies that make up the transformer are shown in Appendix B. A summary of the major materials used in the constructions are also shown in the Appendix. The sequence of drawings begins with the transformer assembly, followed by the winding assembly, details (several parts to a drawing, identified as items) and component parts.

In some cases the detail drawings show parts that apply to the stator and solenoid models. In such cases the same drawings are repeated in the drawings covering the subject models.

Figure IV-5 is a view of the transformer assembly showing the location of the winding leads and thermocouple leads. This design is rated at 1 KVA with 600 volts on the primary winding and approximately 30 volts at the secondary winding. The core is made from E-I style Hiperc 27 laminations 0.008 inch thick. Evaluation of plasma-arc sprayed insulation is required to confirm the interlaminar insulation system that will be used. The windings are formed around a ceramic spool which provides insulation between the windings and the center leg of the core. Ceramic endplates and channels provide insulation between the winding ends and sides respectively and the laminations.

Pairs of thermocouples are installed between the primary winding and ceramic spool and between the two windings. The core has been divided into two halves by ceramic strip spacers so that four thermocouples can be buried in the core.

Non-magnetic alloy strips are used outside the laminations to provide lamination support. The laminations and support strips are held together by through studs, ceramic washers and lock nuts.

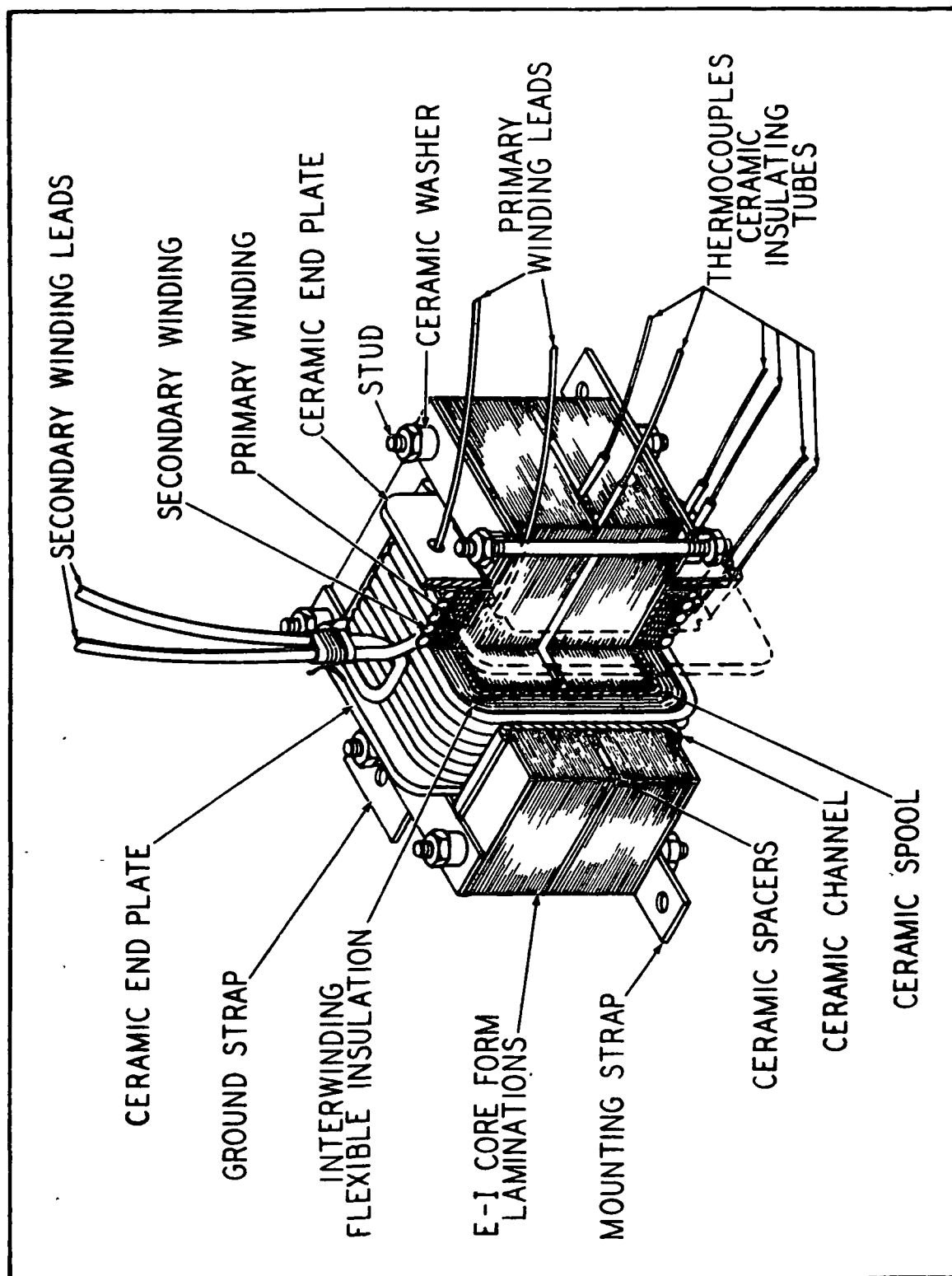


FIGURE IV-5. Cutaway View of Transformer

3. Program for the Next Quarter

- a.) Expedite the procurement of long lead time items.
- b.) Continue the manufacture of transformer detail parts and begin test specimen assembly.
- c.) Evaluate a plasma-arc sprayed Al_2O_3 lamination insulation system.

D. TASK 4 - SOLENOID.

1. Summary of Technical Progress

- a.) A solenoid design has been completed and all long lead time materials are on order.
- b.) An engineering model release has been prepared to initiate manufacture of the parts.

2. Discussion

Appendix C shows the drawings which define the assemblies and components which make up the solenoid. A summary of the major materials used in the constructions are also given in the Appendix. The sequence of drawings begin with the solenoid assembly, followed by the winding assembly, details (several parts to a drawing, identified as items and assemblies) and component parts.

Figure IV-6 is a view of the external configuration of the solenoid showing the location of the coil leads and thermocouple leads. A weight of approximately three pounds is suspended on the plunger, and when the solenoid is actuated the weight is lifted approximately 0.050 inches and held in that position.

The magnetic solenoid housing, cover and plunger are made from a Hiperco 27 forging. The coil is wound on a ceramic spool which provides insulation between the winding and the plunger and housing center core. Ceramic end plates insulate the sides of the winding from the Hiperco 27 housing and cover. Bearing surfaces for the plunger consist of a ceramic guide rod at one end of the plunger and a ceramic bushing at the opposite end.

Pairs of thermocouples are installed between the coil and the ceramic coil spool, midway between the coil ID and OD, between the coil OD and the housing, and on the outside of the housing.

3. Program for the Next Quarter

- a.) Expedite the procurement of long lead time items.
- b.) Complete the manufacture of solenoid detail parts and begin test specimen assembly.
- c.) Carry out a heat transfer calculation from the inside to the outside of the coil.

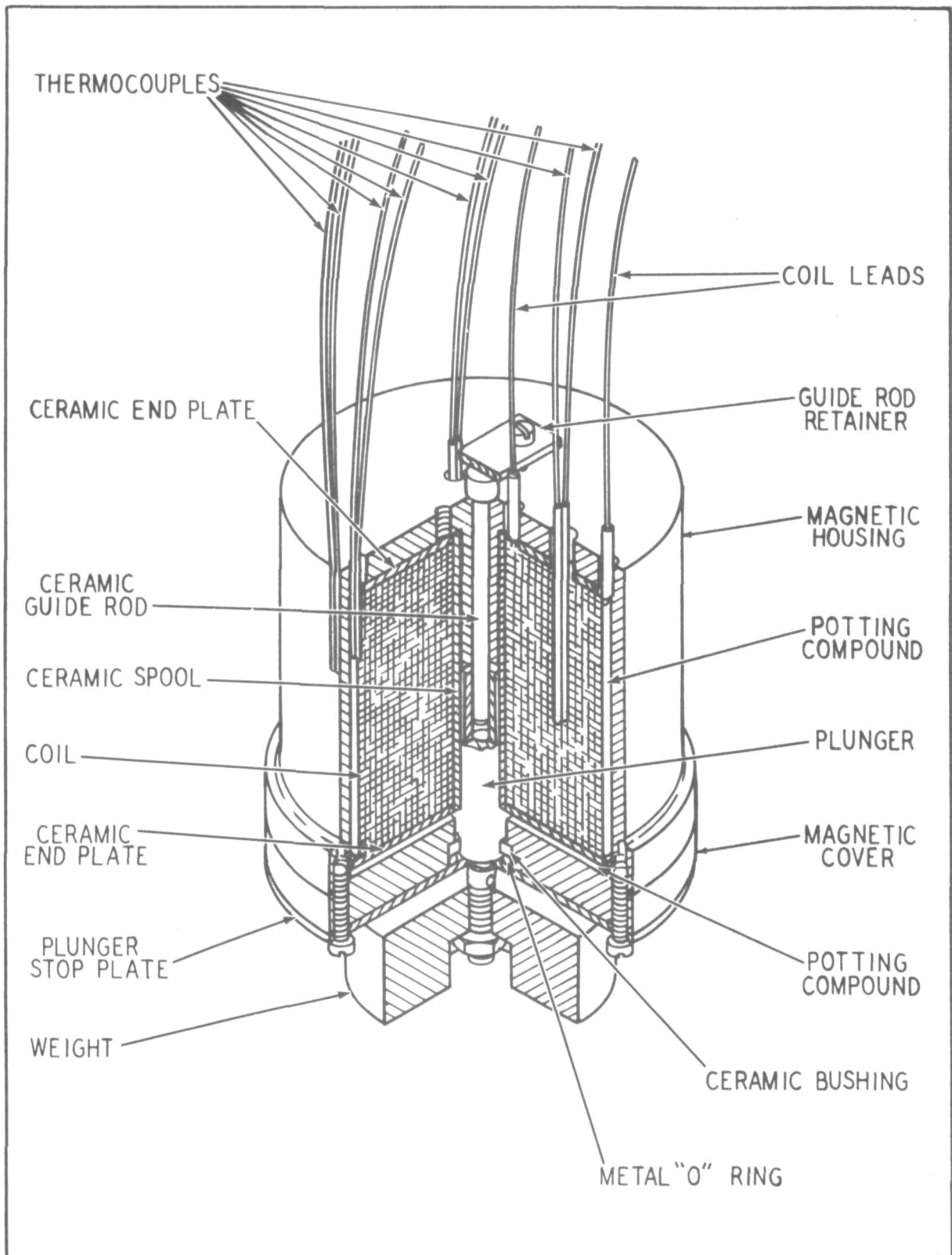


FIGURE IV-6. Cutaway View of Solenoid